

COURSE CURRICULUM

B.Sc. Microbiology

Batch: 2024-2025

Academic Year: 2024-2025 Updated on: July, 2024



VISION

• GSFCU strives to be the best compact boutique institution with a futuristic approach, encouraging student centric culture and sharpened focus on developing industry ready & employable students with all-round development.

MISSION

- Establish an institution, which promotes creativity and innovation.
- Develop unique quality standards for academic excellence and pedagogical innovations.
- Remain agile through learning ecosystem with flexible processes & systems.
- Holistic growth for industry readiness.

No.	Programme Outcomes (POs)	Blooms' Taxonomy Domain	Blooms' Taxonomy Sub Domain
PO1	Basic Knowledge: To impart knowledge regarding basic concepts of applied chemical sciences.	Remembering and Understanding	Explain, Describe, Discuss, Recall, Locate
PO2	Interdisciplinary approach: To explain the relationships between chemical sciences, biological sciences, physical sciences and mathematical sciences.	Application and Analysing	Apply, Practice, Interpret, Select, Correlate
PO3	Practical learning: To perform procedures as per laboratory standards in the areas of Chemical Sciences and to think analytically.	Analysing	Compare, Classify, Select, Investigate
PO4	Effective Communication and social Interaction: To communicate effectively in terms of reading, writing, speaking and delivering the view to others.	Understanding	Explain, Describe, outline, Predict, Summarize
PO5	Ethics: To culminate and understand the moral values for any of the subjects with respect to good practices and humanity.	Evaluating	Judge, Assess, Estimate, Predict, Argue
PO6	Environment and Sustainability: To explain the importance of ecological balance along with conservation of natural resources for human wellbeing.	Creating	Construct, Develop, Produce



No.	Programme Specific Outcomes (PSOs)	Blooms' Taxonomy Domain	Blooms' Taxonomy SubDomain
PSO1	Students will gain and apply knowledge of scientific concepts such as chemistry, physics, mathematics, organic chemistry, inorganic chemistry, physical chemistry, analytical chemistry and computer applications in chemistry to solve problems related to the field of Chemistry.	Remembering and Understanding	Explain, Describe, Discuss, Recall, Locate
PSO2	Students will be able to demonstrate learning skills to work as a team in a multidisciplinary environment.		Apply, Practice, Interpret, Select, Correlate
PSO3	Students will be able to design and develop sustainable solutions to major environmental/biological problems by applying appropriate chemistry tools.	Analysing	Compare,Classify, Select, nvestigate
PSO4	Students will be able to demonstrate effective writing and oral communication skills.	Understanding	Explain, Describe, outline, Predict, Summarize
PSO5	Students will have knowledge and understanding of norms and ethics in the field of chemistry.	Evaluating	Judge, Assess, Estimate, Predict, Argue
PSO6	Students will be able to design, perform experiments, analyze and interpret data for investigating complex problems in chemistry and related fields.	Creating	Construct, Develop, Produce

Mapping of POs & PSOs:

	PO1	PO2	PO3	PO4	PO5	PO6
PSO1	3	3	2	3	3	2
PSO2	2	3	2	3	3	2
PSO3	2	2	3	2	2	2
PSO4	3	2	3	3	2	2
PSO5	2	3	3	2	3	2
PSO6	3	2	3	2	2	3
Avg.	2.5	2.5	2.7	2.5	2.5	2.2

1: Slight (Low); 2: Moderate (Medium); 3: Substantial (High); 0 None



Definition of Credit:

1 Hour. Lecture (L) per week	1 credit
1 Hour Tutorial (T) per week	1 credit
4 Hours Practical (P) per week	2 credit
2 Hours Practical (P) per week	1 credit
1 Hour Practical (P) per week	0.5 credit
3 Hours Experiential learning	1 credit

Course code Definitions:

Lecture	L
Tutorial	Т
Practical	P
Basic Science Courses	BSC
Engineering Science Courses	ESC
Humanities and Social Sciences includingManagement courses	HSMC
Professional core courses/Major (Core)	PCC
Professional Elective courses/Minor Stream	PEC
Open Elective courses	OEC
Laboratory course	LC
Mandatory courses	MC
Non-credit courses	NC
Project (Experiential learning)	PROJ
Experiential learning ex. Internship, IndustrialVisit, Field visit, etc.	EL
Multidisciplinary courses	MDC
Ability Enhancement Course	AEC
Skill Enhancement Course	SCE
Value Added Courses	VAC
	1



Structure of Undergraduate Programme:

Sr. No.	Category	Credit Breakup
1	Professional core courses -Major (Core)	78
2	Professional Elective courses relevant to chosen specialization/branch - Minor Stream	48
4	Project work, seminar and internship in industry or elsewhere	8
5	Mandatory Courses [Environmental Sciences, Induction Programme, Indian Constitution, Essence of Indian Knowledge Tradition]	(non- credit)
	Total	134



Table: Minimum Credit Requirement

S.No.	Broad Category of Course	Minimum Credit Requirement
		3-year UG
1	Major (Core) (50% of total credit) BSMO111 Cell Biology BSMO112 Biomolecules BSMO211 Molecular Biology BSMO212 Metabolism BSMO311 Microbial Genetics BSMO312 Introduction to Microbiology and Microbial Diversity BSMO313 Bacteriology BSMO411 Industrial Microbiology BSMO412Recombinant DNA Technology BSMO413 Microbial Physiology BSMB501 Virology BSMB502 Medical Microbiology BSMB601 Immunology BSMB602 Bioinformatics & Drug Discovery, Design and Development	84
2	Ability Enhancement Courses (AEC) AECC101 Fundamentals of English AECC201 Communication Skills in English AECC301 Entrepreneurship Development AECC401 Environmental Science AECC501 Disaster Risk Management AECC601 Indian Constitution	12
3	Skill Enhancement Courses (SEC) (from major & Minor) SECC101 Foundation Course SECC104Internship SECC204 Internship SECC304 Internship SECC404 Internship SECC504 Internship	12
4	Summer Internship (mainly from Major & Minor) SECC104Internship SECC204 Internship SECC304 Internship SECC404 Internship SECC504 Internship	10
	Total	118



Category-wise Courses:

Humanities & Social Sciences Courses

i. Number of Humanities & Social Science Courses: 2

ii. Credits: 4

Sr.	Course Code	Course Name	Sem	Teachi	ours/week)	Teaching Credit					
No.			Sem	L	P	T	Total	L	P	T	Total
1.	AECC101	Fundamentals of English	I	2	0	0	2	2	0	0	2
2.	AECC201	Communication Skills in English	II	2	0	0	2	2	0	0	2
3	AECC301	Entrepreneurship Development	III	2	0	0	2	2	0	0	2
4	AECC401	Environmental Science	IV	2	0	0	2	2	0	0	2
5	AECC501	Disaster Risk Management	V	2	0	0	2	2	0	0	2
		Total									10

Note:

 $L = Lecture, \ P = Practice, \ T = \ Tutorial, \ MS \ - \ Mid \ Semester, \ CEC \ - \ Continuous \ Evaluation \ Component, \ ES \ - End \ Semester$

Basic Science Course- Discipline Specific Generic Electives

i. Number of Basic Science Course: 108

ii. Credits: 18

Sr.	G G. 1.	Course Name	C	Т	eaching (Hour	g Schei s/week)		Teaching Credit			
No.	CourseCode		Semester	L	P	Т	Total	L	P	Т	Total
1.	BSCM116	Basics of Chemistry – I	I	2	2	0	4	2	1	0	3
2.	BSM A116	Mathematics – I	I	2	0	1	3	2	0	1	3
3.	BSCM216	Basics of Chemistry – II	II	2	2	0	4	2	1	0	3
4.	BSPY216	Physics – I	II	2	2	0	4	2	1	0	3
5.	BSCM316	Chemistry – I	III	2	2	0	4	2	1	0	3
6.	BSPY316	Physics – II	III	2	2	0	4	2	1	0	3



7.	BSMA316	M athematics – II	Ш	2	0	1	3	2	0	1	
8.	BSCM416	Chemistry – II	IV	2	2	0	4	2	1	0	3
9.	BSPY416	Biophysics	IV	2	2	0	4	2	1	0	3
10.	BSM A416	Biostatistics	IV	2	0	1	3	2	0	1	3
11.	BSMB503	Microbial Biotechnology	V	4	4	0	8	4	2	0	
12.	BSMB504	Research Methodology	V	4	4	0	8	4	2	0	6
15.	BSMB505	Bioanalytical tools	V	4	4	0	8	4	2	0	
16.	BSMB506	Microbes in Sustainable Agriculture and Development	V	4	4	0	8	4	2	0	6
17.	BSMB603	Food and Dairy Microbiology	VI	4	4	0	8	4	2	0	
18.	BSM B604	Biosafety and Intellectual Property Rights	VI	4	4	0	8	4	2	0	6
19.	BSM B605	Environmental Microbiology	VI	4	4	0	8	4	2	0	
20.	BSM B606	Advances in Microbiology	VI	4	4	0	8	4	2	0	6
		Total		74	64	2	140	74	32	2	48

Professional Core Courses

i. Number of Professional Core Courses: 14

ii. Credits: 84

Sr.	Course	Course Name	Semester	Teaching Scheme (Hours/week)				Teaching Credit			
No.	Code	Course Name	Semester	L	P	T	Total	L	P	T	Total
1.	BSMO111	Cell Biology	I	3	2	0	5	3	1	0	4
2.	BSM O112	Biomolecules	I	3	2	0	5	3	1	0	4
3.	BSM O211	Molecular Biology	II	3	2	0	5	3	1	0	4
4.	BSM O212	Metabolism	II	3	2	0	5	3	1	0	4
5.	BSMO311	Microbial Genetics	III	3	2	0	5	3	1	0	4
6.	BSMO312	Introduction to Microbiology and Microbial Diversity	III	3	2	0	5	3	1	0	4
7.	BSMO313	Bacteriology	III	3	2	0	5	3	1	0	4
8.	BSM O411	Industrial Microbiology	IV	3	2	0	5	3	1	0	4



9.	BSMO412	Recombinant DNA Technology	IV	3	2	0	5	3	1	0	4
10.	BSMO413	M icrobial Phy siology	IV	3	2	0	5	3	1	0	4
11.	BSMB501	Virology	V	4	4	0	8	4	2	0	6
12.	BSMB502	Medical Microbiology	V	4	4	0	8	4	2	0	6
13.	BSMB601	Immunology	VI	4	4	0	8	4	2	0	6
14.	BSM B602	Bioinformatics & Drug Discovery, Design and Development	VI	4	4	0	8	4	2	0	6
		Total		46	36	00	82	46	18	00	64

Professional Elective Courses-

(i) Number of Professional Elective Course: 108

(ii) Credits: 16

Sr.	Garage Gal	Course Name	Semester	Т	eachin (Hour	g Schei s/week)		Teaching Credit					
No.	CourseCode	Course Name	Semester	L	P	Т	Total	L	P	Т	Total		
1.	BSCM116	Basics of Chemistry – I	I	2	2	0	4	2	1	0	3		
2.	BSMA116	Mathematics – I	I	2	0	1	3	2	0	1	3		
3.	BSCM216	Basics of Chemistry – II	II	2	2	0	4	2	1	0	3		
4.	BSPY216	Physics – I	II	2	2	0	4	2	1	0	3		
5.	BSCM316	Chemistry – I	III	2	2	0	4	2	1	0	3		
6.	BSPY316	Physics – II	III	2	2	0	4	2	1	0	3		
7.	BSM A316	Mathematics – II	III	2	0	1	3	2	0	1			
8.	BSCM416	Chemistry – II	IV	2	2	0	4	2	1	0	3		
9.	BSPY416	Biophysics	IV	2	2	0	4	2	1	0	2		
10.	BSM A416	Biostatistics	IV	2	0	1	3	2	0	1	3		
11.	BSM B503	Microbial Biotechnology	V	4	4	0	8	4	2	0			
12.	BSM B504	Research Methodology	V	4	4	0	8	4	2	0	6		



15.	BSMB505	Bioanalytical tools	V	4	4	0	8	4	2	0	
16.	BSM B506	Microbes in Sustainable Agriculture and Development	V	4	4	0	8	4	2	0	6
17.	BSM B603	Food and Dairy Microbiology	VI	4	4	0	8	4	2	0	
18.	BSM B604	Biosafety and Intellectual Property Rights	VI	4	4	0	8	4	2	0	6
19.	BSM B605	Environmental Microbiology	VI	4	4	0	8	4	2	0	
20.	BSM B606	Advances in Microbiology	VI	4	4	0	8	4	2	0	6
		Total		74	64	2	140	74	32	2	48

Project Work, Seminar and Internship In Industry Or Elsewhere

i. Number of Project Work, Seminar And Internship In Industry Or Elsewhere: 5

ii. Credits: 10

Sr.	CourseCode	Course Name	Semester	Т	eaching (Hours			Teaching Credit					
No.				L	P	T	Total	L	P	T	Total		
1.	SECC101	Industrial Internship	I	0	0	0	2	0	0	0	2		
2.	SECC201	Industrial Internship	II	0	0	0	2	0	0	0	2		
3.	SECC301	Industrial Internship	III	0	0	0	2	0	0	0	2		
4.	SECC401	Industrial Internship	IV	0	0	0	2	0	0	0	2		
5.	SECC501	Industrial Internship	V	0	0	0	2	0	0	0	2		
		Total					10				10		

 \ddot{Note} : L = Lecture, P = Practice, T= Tutorial, MS - Mid Semester, CEC - Continuous Evaluation Component, ES - End Semester

Ability Enhancement Courses

i. Number of Ability Enhancement Courses: 6

ii. Credits: 12

Sr.	Course	Common Nama	Composton		ching So lours/we			,	Геаchin	g Cr	edit
No.	Code	Course Name	Semester	L	P	T	Total	L	P	T	Total
1.	AECC101	Fundamentals of English	I	2	0	0	2	2	0	0	2



2.	Δ FCC201	Communication Skills in English	П	2	0	0	2	2	0	0	2
3.	1 A H(("3())	Entrepreneurship Development	III	2	0	0	2	2	0	0	2
4.	AECC401	Environmental Studies	IV	2	0	0	2	2	0	0	2
5.	AECC501	Disaster Risk Management	V	2	0	0	2	2	0	0	2
6.	AECC601	Indian Constitution	VI	2	0	0	2	2	0	0	2
		Total		12			12	12			12

Note: L = Lecture, P = Practice, T = Tutorial, MS - Mid Semester, CEC - Continuous Evaluation Component, ES - End Semester

Skill Enhancement Compulsory/Elective Courses

i. Number of Skill Enhancement Courses: 12

ii. Credits: 16

Sr.	Course			Teaching	gScheme(week)	Teaching Credit					
No.	Code	Course Name	Semester	L	P	Т	Total	L	P	Т	Total	
1.	SECC101	Industrial Internship	I	0	2	0	2	0	2	0	2	
2.	SECC102	Foundation Course	I	2	0	0	2	2	0	0	2	
3.	SECC201	Industrial Internship	П	0	2	0	2	0	2	0	2	
4.	SECC301	Industrial Internship	Ш	0	2	0	2	0	2	0	2	
5.	SECC401	Industrial Internship	IV	0	2	0	2	0	2	0	2	
6.	SECC501	Industrial Internship	V	0	2	0	2	0	2	0	2	

Note: L = Lecture, P = Practice, T = Tutorial, MS - Mid Semester, CEC - Continuous Evaluation Component, ES - End Semester

About the Program:

Science is the basic foundation of any technological and engineering creation. In view of the changing scenario at the national and international level in the field of Science and Technology, there is a great demand for basic sciences with considerable knowledge of its

School of Science B.Sc. Microbiology, Course Curriculum Academic Year, 2024-25



applications. GSFC University is committed to high academic standards.

The B.Sc. (Hons.) Microbiology Program is an Honours Degree which is designed for Six Semesters (Three Years) in such a way that a good basic foundation of subjects is laid and applications along with recent developments are covered. Students will also get theoretical and practical knowledge by undergoing industrial internship after every semester.

The more focused specialization course of Microbiology is designed to full fill recent demands of industrial career. The B.Sc. (Hons.) Microbiology Program provides an opportunity to make a career in R&D, Industries and Academic Institutions. Opportunity for the placement may be provided by the Institute.



		Semester – I					
Sr. No.	Course Code	Course Title	L	T	P	С	Marks
		A. Major					
1	BSMO111	Cell Biology	3	0	1	4	150
2	BSMO112	Biomolecules	3	0	1	4	150
		B. Minor					
3	BSCM116	Basics of Chemistry – I	2	0	1	3	75
4	BSMA116	Mathematics – I	2	1	0	3	75
C. M	lultidisciplinary	Additional Credits under Choice Bas	sed C	re dit S	Syste	m (Aı	ny One)
5	CBCS103	Indian History and culture	2	0	0		
6	CBCS104	Air Pollution Control	2	0	0		
7	CBCS105	Python Programming	2	0	0		
8	CBCS106	Basic of Chemical Engineering	2	0	0	2	100
9	CBCS109	Management Principles & Practices	2	0	0		
10	CBCS110	MS Office & Applications	2	0	0		
		D. Ability Enhancement Cour	se				
11	AECC101	Fundamentals of English	2	0	0	2	50
		E. Skill Enhancement Course	9				
11	SECC104	Internship	0	0	2	2	50
		F. Value Added Course					
12	VACC101	Foundation Course	0	0	2	2	50
				Т	otal	22	700



Teaching Scheme Semester – I

Sr.	Course			achin (Hour		heme ek)	Т	'eachi	ng C	redit	Evaluation Scheme					
No.	Code	Course Name	L	P	Т	Total	L	P	Т	Total	Theory: MS Marks	Theory: CEC Marks	Theory: ES Marks	Theory Marks	Practical Marks	Total Marks
								A.	Majo	r Course	S					
1.	BSMO111	Cell Biology	3	2	0	5	3	1	0	4	20	40	40	100	50	150
2.	BSMO112	Biomolecules	3	2	0	5	3	1	0	4	20	40	40	100	50	150
								B.	Mino	r Course	S					
3.	BSCM116	Basics of Chemistry - I	2	2	0	4	2	1	0	3	20	40	40	100	50	150
4.	BSMA116	Mathematics – I	2	1	0	3	2	0	1	3	20	45	75	150	0	150
				•		C.	Multi	discip	linar	y Course	es (Any one)					
5.	CBCS103	Indian History & Culture														
6.	CBCS104	Air Pollution Control														
7.	CBCS105	Python Programming	2	0	0	2	2	0	0	2	20	40	40	100	00	100
8.	CBCS109	Management Principles & Practices														
9.	CBCS110	MS Office & Applications														
							D. A	bility	Enha	ncement	Courses					
10.	AECC101	Fundamentals of English	2	0	0	2	2	0	0	2	20	40	40	100	00	100
							E. S	skill E	nhan	cement (Courses					



11.	SECC101	Industrial Internship	0	2	0	2	0	2	0	2	00	00	00	50	00	50
							F	. Valı	ie Ac	ded Cou	ırses					
12.	VACC101	Foundation Course	0	2	0	0	0	2	0	2	00	00	00	50	00	50
	Total									22						700



COURSE CODE COURSE NAME SEMESTER Cell Biology I

	Teaching S	Scheme (Hou	rs)	Teaching Credit						
Lecture	Practical	Tutorial	Total Hours	Lecture	Practical	Tutorial	Total Credit			
45	30	00	75	3	1	0	4			

Course Pre-requisites	School Level Understanding of Biology and a keen interest in learning.
Course Category	Professional Core Courses
Course focus	Employability
Rationale	The subject "Cell Biology" provides a comprehensive understanding of life's
	foundation through units on cell structure, organelles, and functions. It covers
	molecular aspects, the endomembrane system, cell division, and essential research
	tools, fostering insights into biology's intricate workings.
Course Revision/	07/11/2023
Approval Date:	
Course Objectives	To enable the student to:
(As per Blooms'	1. Gain the basic knowledge and understanding of basic concept and structure
Taxonomy)	of cells and cell organelles.
	2. Understand the molecular structure and function of major organelles.
	3. Understand and analyze the role of endomemranous cell organelles
	4. Learn and understand cell division and cell cycle.
	Understand and apply the basic tools in cell biology.
	Course Content (Theory) Weightage Contact

Course Content (Theory)	Weightage	Contact
		hours
Unit 1: Basic concept of cell and structure of organelles Introduction to	20%	9
the concept of cell and evolution of eukaryotic cells. General structure and		
constituents of cell. Similarities and distinction between plant and animal		
cells. Structure, composition and function of cell wall and the cell		
membrane. Membrane transport.		
Unit 2: Molecular structure and function of major organelles Nucleus -	20%	9
Nuclear envelope, nuclear pore complex and nuclear lamina.		
Chromatin – Molecular organization. Chloroplast, Mitochondria,		
Lysosomes, Peroxisomes, Vacuoles.		
Unit 3: The Endomembrane system	20%	9
Endoplasmic reticulum, Golgi Apparatus, Ribosomes, Ribosomes in relation		
to cell growth and division. Cytoskeleton: structure, composition and		
function. Cilia and flagella, Centrioles, Extracellular matrix and Cell		
adhesion in cell cycle regulation		
Unit 4: Cell division and cell cycle	20%	9



Mitosis	and	Meiosis. I	Eukaryotic	cell	cycle.	Cell	cycle	control	in		
prokaryo	otes a	and eukaryotes	5.								
Unit 5:	Unit 5: Basic tools in cell biology								20%	9	
Basics	of	Microscopy,	Microtom	ıy,	Density	grad	ient (centrifugati	on.		
Staining	Basics of Microscopy, Microtomy, Density gradient centrifugation. Staining techniques.										

List Of Practical	Weightage	Contact hours
1. Good Laboratory Practice and Safety in Microbiology & Biotechnology Lab.	2%	2
2. To study the principle and applications of important instruments (biological safety cabinets, autoclave, incubator, BOD incubator, hot air oven, light microscope) used in the microbiology laboratory.	2%	2
3. Preparation of various stains	2%	2
4. Simple staining of plant sections	2%	2
5. Simple staining of bacterial culture and wet mount of Hay infusion	2%	2
6. Separation of Cell organelles by Sucrose Gradient (Virtual Demo).	2%	2
7. Negative staining	2%	2
8. Observation of various stages of mitosis by permanent slides	2%	2
9. Observation of various stages of meiosis by permanent slides	2%	2
10. Sterility practices in cell and tissue culture	2%	2
11. Cell culture, preservation and revival of Animal Cell Culture (Demo) & Preservation of Bacterial culture by Glycerol stock method	2%	2
12. Measurement of cell viability of yeast cell by MTT or (Methylene blue) Trypan blue assays.	2%	2
13. Revision	2%	2
14. Revision	2%	2
15. Revision	2%	2

Instructional Method and Pedagogy:

Audio-Visual Lectures, Quizzes, Debates, Project works, Case studies, and Assignments. Practical exercises are designed to understand the theory as taught in classroom.

Course Outcomes:	Blooms' Taxonomy Domain	Blooms' Taxonomy Sub Domain
After successful completion of the above course,	Remember,	Describe
students will be able to:	Understanding	
CO1: Understand the structure, composition and function		



ofcell organelles.		
CO2: To describe physiological processes and molecular	Remember,	Explain
mechanisms regulated by cell organelles.	Understanding,	
	apply	
CO3: Understand and analyze the role of	Understanding,	Explain
endomembranous system.	Analyze	
CO4: Understand the principal and types of cell signaling.	Understanding	Describe
CO5: Remember, understand and apply the basic tools in	Remember,	Describe
cell biology.	Understanding	

Learning Resources

1. Reference books:

- 1. Essential Cell Biology by Bruce Alberts, Dennis Bray, Karen Hopkin and Alexander Johnson.
- 2. Karp, G. Cell and Molecular Biology: Concepts and Experiments. John Wiley & Sons.
- 3. De Robertis, E.D.P. and De Robertis, E.M.F. Cell and Molecular Biology. VIII Edition.
- 4. Cooper, G.M. and Hausman, R.E. The Cell: A Molecular Approach. V Edition. ASM Press

2. Journal & Periodicals:

- 1. Journal of Cell Biology
- 2. Trends in Cell Biology
- 3. Cell Biology International
- 4. Science
- 3. Other Electronic resources: NPTEL

Evaluation Scheme	Total Marks	= 150
Theory: Mid semester Marks	20 mark	CS .
Theory: End Semester Marks	40 mark	CS .
Theory: Continuous	Attendance	05 marks
Evaluation Component Marks	MCQs	10 marks
IVICIA	Open Book Assignment	15 marks
	Research Paper Review	10 marks
	Total	40 Marks
Practical Marks		·
	Attendance	05 marks
	Practical Exam	35 marks
	Viva	10 marks
	Journal	05 marks



	Total	50 Marks	

Mapping of PSOs & COs

	PSO1	PSO2	PSO3	PSO4	PSO5	PSO6
CO1	2	2	-	-	1	1
CO2	2	2	1	-	-	1
CO3	1	1	-	-	1	1
CO4	1	1	-	-	-	-
CO5	2	2	2	-	1	2

Mapping of POs & COs

	PO1	PO2	PO3	PO4	PO5	PO6
CO1	3	2	1	1	-	-
CO2	3	2	-	-	-	1
CO3	2	-	1	-	-	-
CO4	1	1	-	-	1	-
CO5	3	2	2	1	2	-

1: Slight (low); 2: Moderate (Medium); 3: Substantial (High); 0 None



COURSE CODE	COURSE NAME	SEMESTER- I
BSMO112	Biomolecules	

Teaching Scheme (Hours)			Teaching Credit				
Lecture	Practical	Tutorial	Total hours	Lecture	Practical	Tutorial	Total Credit
45	30	00	75	3	1	0	4

Course Pre-	School Level Understanding of Biological molecules and a keen interest in
requisite	learning.
Course Category	Core Professional
Course Focus	Employability
Rationale	Comprehensive understanding of biomolecules, their functions, and
	biological roles. It discusses about the structure, functions, and properties of
	carbohydrates, lipids, nucleic acids, amino acids, proteins, and enzymes,
	providing a foundation in biochemistry and molecular biology for
	understanding life processes.
Course Revision/	07/03/2024
Approval date	
Course Objectives	To enable the student:
(As per Blooms'	1. To understand the structure, function, and properties of carbohydrates and
Taxonomy)	analyze its significance in biological processes.
	2. To remember the structure, functions and classification of lipids.
	3. To understand and remember physical and chemical properties of nucleic
	acids and analyse its significance.
	4. To understand and analyse the structure and function of amino acids.
	5. To understand the nomenclature of enzymes and its significance

Course Content	Weightage	Contact
		Hours
Unit 1: Carbohydrates	20%	9
Structure, Function and properties of Monosaccharides, Disaccharides		
and Polysaccharides, Glycoprotein's and their biological functions.		
Unit 2: Lipids	20%	9
Structure and functions -Classification, nomenclature and properties of		
fatty acids, essential fatty acids. Phospholipids, sphingolipids, glycolipids,		
cerebrosides, gangliosides, Prostaglandins, Cholesterol.		
Unit 3: Nucleic acids	20%	9
Structure and functions: Physical & chemical properties of Nucleic acids,		
Nucleosides & Nucleotides, purines & pyrimidines,. Biologically important		
nucleotides, Double helical model of DNA structure and forces responsible		
for A, B & Z – DNA, denaturation and renaturation of DNA, types of		
RNA.		
Unit 4: Amino acids	20%	9
A historical prospective. Amino acids & Proteins: Structure & Function.		



Structure and properties of Amino acids, Types of proteins and their classification. Different Level of structural organization of proteins, Protein Purification. Denaturation and renaturation of proteins. Fibrous and globular proteins.		
Unit 5: Enzymes	20%	9
Nomenclature and classification of Enzymes, Holoenzyme, apoenzyme,		
Cofactors coenzyme, prosthetic groups, metalloenzymes, monomeric &		
oligomeric enzymes and Vitamins.		

List Of Practical	Weightage	Contact
		hours
1. Qualitative test for Carbohydrate: Iodine test	2%	2
2. Qualitative test for Carbohydrate: Benedict test	2%	2
3. Identification of sugars – char (Molisch, Iodine, Benedict, Barfoed,	2%	2
Seiwanoff, Osazone)		
4. Identification of sugars – char (Molisch, Iodine, Benedict, Barfoed,	2%	2
Seiwanoff, Osazone)		
5. Qualitative test for lipids: solubility, transluscent, acrolein, Hubbles test	2%	2
for saturated/unsaturated lipids, saponification, Burchard test for		
cholesterol		
6. Qualitative test for lipids: solubility, transluscent, acrolein, Hubbles test	2%	2
for saturated/unsaturated lipids, saponification, Burchard test for		
cholesterol		
7. Measure Saponification value of oil/fat	2%	2
8. Detection of DNA by Diphenyl Amine method	2%	2
9. Detection of RNA by Orcinol method	2%	2
10. Qualitative test for proteins: Biuret, Nin-hydrin spot, Sulphur	2%	2
11. Detect presence of protein from various samples (egg white,	2%	2
germinating sprouts)		
12. Revision	2%	2
13. Revision	2%	2
14. Revision	2%	2
15. Revision	2%	2

Instructional Method and Pedagogy:Audio-Visual Lectures, Quizzes, Debates, Project works, Case studies, and Assignments. Practical exercises are designed to understand the theory as taught in classroom.

Course Outcomes:	Blooms' Taxonomy Domain	Blooms' Taxonomy Sub Domain
After successful completion of the above cour	se, students will be able:	
CO1: Understand and analyze the structural diversity of carbohydrates and glycoproteins, and their roles in biological functions.	Understand, Analyse, Remember	Define, Classify & Demonstrate
CO2: Classify lipids by structure and function, analyze their roles in cellular processes, and evaluate essential fatty acids' significance in nutrition and health.	Analyse, Apply, Understand	Classify, Describe & Demonstrate
CO3: Explain the physical and chemical properties of nucleic acids, including DNA structure, RNA types, and the importance of nucleotides.	Understand, remember	Explain, Describe & Demonstrate



CO4: Analyze the historical context of amino acids and proteins, classify proteins, evaluate their structural organization, and explain denaturation processes.	Remember, Analyse	Define Describe
CO5: Describe enzyme nomenclature and classification, identify enzyme components, explain cofactors, and evaluate the role of vitamins in enzymatic reactions.	Understand, Apply	Define, Classify, Describe & Demonstrate

Learning Resources

- 1. Reference Books
 - 1. Berg, J. M., Tymoczko, J. L. and Stryer, L. (2006). Biochemistry. VI Edition. W.H Freeman andCo.
 - 2. Buchanan, B., Gruissem, W. and Jones, R. (2000) Biochemistry and Molecular Biology of Plants. American Society of Plant Biologists.
 - 3. Nelson, D.L., Cox, M.M. (2004) Lehninger Principles of Biochemistry, 4th Edition, WH Freeman and Company, New York, USA.
 - 4. Biochemistry by U Satyanarayan
- 2. Journals & Periodicals:
 - 1. JBC
 - 2. Current Science
- 3. Other Electronic resources:
 - 1. NPTL

Evaluation Scheme	Total Marks = 150				
Theory: Mid semester Marks	20 marks				
Theory: End Semester Marks	40 marks				
Theory: Continuous Evaluation Component Marks	Attendance MCQs Open Book Assignment Research Paper Review Total	05 marks 10 marks 15 marks 10 marks			
Practical Marks	Attendance Practical Exam Viva Journal Total	05 marks 35 marks 10 marks 05 marks			

Mapping of PSOs & Cos



	PSO1	PSO2	PSO3	PSO4	PSO5	PSO6
CO1	2	1	1	0	1	1
CO2	2	0	1	0	1	1
CO3	2	1	1	0	0	1
CO4	1	2	1	0	1	1
CO5	2	1	1	0	1	1

Mapping of POs & Cos

	PO1	PO2	PO3	PO4	PO5	PO6
CO1	3	2	1	0	1	0
CO2	3	2	2	0	1	0
CO3	3	2	1	0	0	0
CO4	3	2	2	0	1	0
CO5	3	2	0	0	1	0

1: Slight (low); 2: Moderate (Medium); 3: Substantial (High); 0 None



COURSE NAME	SEMESTER- I
Basics of Chemistry - I	
]	

Teaching Scheme (Hours)		Teaching Credit					
Lecture	Practical	Tutorial	Total hours	Lecture	Practical	Tutorial	Total Credit
30	30	00	60	2	1	0	3

Course Pre-	Basic understanding of high school chemistry.
requisites	
Course Category	Generic Elective
Course Focus	Skill Development
Rationale	This course aims to provide a comprehensive understanding of fundamental chemical principles, including atomic structure, bonding, solution chemistry, resonance, and chemical kinetics, essential for advanced studies in chemistry and related fields.
Course Revision/	07/03/2024
Approval date	
Course Objectives	To enable the student to:
(As per Blooms'	1: Understand the principles of atomic structure and electronic
Taxonomy)	configurations.
	2: Apply knowledge of chemical bonding theories to explain bond formation.
	3: Analyze various factors affecting solubility and solute-solvent interactions
	in solutions.
	4: Evaluate the concept of resonance and its application in inorganic and
	organic compounds.
	5: Interpret kinetic data to determine reaction rates and mechanisms.

Course Content	Weightage	Contact
		hours
Unit 1: Atomic Structure	20%	6
Rules for filling electrons in various orbitals, Electronic		
configurations of the atoms. Stability of half-filled and completely		
filled orbitals, concept of exchange energy. Relative energies of		
atomic orbitals, Anomalous electronic configurations.		
Unit 2: Chemical Bonding	20%	6
Types of bonds and factors affecting the bond formation, various		
theories, bond parameters, types of bonds in biomolecule,		
Hydrophilic and hydrophobic interactions.		
Unit 3: Solutions & Solvents	20%	6
Solutions: Solutions, types of solutions, solvation energy, lattice		
energy, Equivalent & molecular mass, mole concept, solubility &		
factors affecting solubility, Expression for concentration of		



solutions, polarity of solvents, Importance of dielectric constant of solvents, Solvents other than water, classification of solvents, Dilution factor, serial dilution, Solute— solvent interactions in		
solutions.		
Unit 4: Resonance	15%	4
Concept of resonance and resonating structures in various inorganic		
and organic compounds.		
Unit 5: Chemical kinetics	25%	8
Rate of reaction, differential rate law expressions, Order &		
molecularity, rate constant, integrated equations (1st, 2nd & 3rd		
order), nth life of a reaction, Arrhenius equations, temperature		
dependence of rate constant, energy profile diagrams. Reaction		
intermediates, Different theories on reaction rate		

List Of Practicals	Weightage	Contact hours
Polar and Non-Polar Molecules Identification	10%	2
Preparation of Solutions and Concentration Measurement	10%	2
Determination of Solubility	10%	2
Serial Dilution and Calculation of Dilution Factor	10%	2
Determination of Equivalent and Molecular Mass	15%	2
To determine the normality and strength of x N H2C2O4 by titrating it against 0.2 N NaOH solution	20%	2
To determine the normality and strength of x N FAS by titrating it against 0.5 N KMnO4 solution	20%	2

Instructional Method and Pedagogy: Chalk-Talk, Classroom Discussions, Notes, Use of GeoGebra Toolbox.

Course Outcomes: Blooms' Blooms' Taxonomy Sub Domain Domain
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After successful completion of the above course, students will be able to:		
CO1: Identify the rules for filling electrons in various orbitals and describe the electronic configurations of atoms, including the stability of half-filled and completely filled orbitals.	Identify and Describe	Knowledge and Comprehension
CO2: Explain various types of chemical bonds and illustrate factors affecting bond formation, including hydrophilic and hydrophobic interactions in biomolecules.	Explain and Illustrate	Comprehension and Application
CO3: Analyze the factors affecting solubility and solve problems related to solute-solvent interactions, including the calculation of solvation energy and concentration expressions for solutions.	Analyze and Solve	Analysis and Application
CO4: Demonstrate basic knowledge of matrices and determinants and apply it to find the inverse of a matrix.	Evaluate and Discuss	Evaluation and Comprehension
CO5: Calculate reaction rates, rate constants, and interpret kinetic data to determine reaction mechanisms, including the application of Arrhenius equations and understanding energy profile diagrams.	Calculate and Interpret	Application and Analysis

Learning	Resources						
1	Reference Books:						
	1. A Textbook of Inorganic Chemistry by J.D. Lee.						
	2. Principles of Physical Chemistry by B.R. Puri, L.R. Sharma, and M.S. Pathania.						
	3. Inorganic Chemistry by Gary L. Miessler, Paul J. Fischer, and Donald A. Tarr.						
	4. Chemical Kinetics and Dynamics by Jeffrey I. Steinfeld, Joseph S. Francisco, and						
	William L. Hase						
	5. Principles of Inorganic Chemistry by Puri Sharma Kalia						
2	Journals & Periodicals:						
	1. Chemistry Today						
3	Other Electronic Resources:						
	http://www.chemguide.co.uk/						

Evaluation Scheme	Total Marks				
Theory: Mid semester Marks	20 marks				
Theory: End Semester Marks	40 marks				
Theory: Continuous	Attendance	05 marks			
Evaluation Component Marks	MCQs	10 marks			
112002110	Open Book Assignment	15 marks			



	Open Book Assignment	10 marks
	Total	40 Marks
Practical Marks	Attendance	05 marks
	Practical Exam	20 marks
	Viva	10 marks
	Journal	10 marks
	Discipline	05 marks
	Total	50 Marks

Mapping of PSOs & COs

	PSO1	PSO2	PSO3	PSO4	PSO5	PSO6
CO1	3	0	0	0	0	0
CO2	3	1	1	0	0	0
CO3	2	0	2	0	0	0
CO4	0	0	0	0	0	0
CO5	0	0	3	0	0	3

Mapping of POs & Cos

	PO1	PO2	PO3	PO4	PO5	PO6
CO1	0	1	1	0	0	0
CO2	0	1	0	0	0	0
CO3	0	1	0	0	0	0
CO4	1	1	0	0	0	0
CO5	1	1	0	0	0	0

1: Slight (low); 2: Moderate (Medium); 3: Substantial (High); 0 None



COURSE CODE	COURSE NAME	SEMESTER
BSMA116	Mathematics-I	I

Teaching Scheme (Hours)				Teaching (Credit		
Lecture	Practical	Tutorial	Total Hours	Lecture Practical Tutorial			
30	0	15	45	2	0	1	3

Course Pre-requisites	Knowledge of basic precalculus concepts
_	
Course Category	Discipline Specific Generic Elective
Course focus	Skill development
Rationale	
Course Revision/	07/03/2024
Approval Date:	
Course Objectives	To enable the student to:
(As per Blooms'	1: Demonstrate knowledge of basic precalculus concepts and skills.
Taxonomy)	2: Evaluate limits, recognize continuity and use the properties of continuous
	functions.
	3: Find derivatives of algebraic and trigonometric functions using the definition or
	basic rules of differentiation.
	4: Find rates of change, solve related rate problems, Find extreme values in
	optimization problems.
	5: Apply the concepts and methods described in the syllabus, solve problems using
	linear algebra and will know a number of applications of linear algebra

Weightage	Contact hours
20%	6
20%	6
20%	6
200/	
20%	6
200/	
20%	6
_	20%

List Of Practical Tutorial	Weightage	Contact hours
Unit 1: Problem solving on Trigonometry.	20%	3



Unit 2: Problem solving on differentiation	20%	3
Unit 3: Problem solving on Integration.	20%	3
Unit 4: Problem solving on Matrices and Determinants.	20%	3
Unit 5: Problem solving on solving system of linear equations.	20%	3

Instructional Method and Pedagogy: Chalk-Talk, Classroom Discussions, Notes, Use of GeoGebra Toolbox.

Course Objectives:	Blooms' Taxonomy Domain	Blooms' Taxonomy Sub Domain
After successful completion of the above course, students will be able	Understand and	Describe,
to:	Apply	Demonstrate &
CO1. Understand and Apply Concerts of linear electric to find souls		Examine, Find,
CO1: Understand and Apply Concepts of linear algebra to find rank and inverse of a matrix.		Evaluate
CO2: Apply Concepts of linear algebra to solve system of linear	Evaluate	Demonstrate &
equation.		Examine, Find
CO3: Understand and Apply the knowledge of basic trigonometry and	Remember,	Define, Classify,
precalculus concepts and skills.	Understand and	Describe,
	Apply	Demonstrate &
		Examine
CO4: Evaluate derivatives of algebraic and trigonometric functions to	Evaluate	Demonstrate &
find maxima-minima of function of one variable.		Examine, Find,
		Evaluate
CO5: Evaluate integration of algebraic and trigonometric functions and		Demonstrate &
use it find Area of the region.	Evaluate	Examine, Find,
		Evaluate

Learning Reso	Learning Resources						
1.	Reference Books	:					
	1. Shanti Narayan, Integral Calculus, S.Chand & Co.Ltd,1999.						
	2. Shanti Naraya	n, Differential Calculus, S. Chand & Co. Ltd,1999.					
	3. Erwin Kreyszi	ig, Advanced Engineering Mathematics, John Wiley and Sons Inc, 1983.					
	4. G.B. Thomas J	Jr. and R.L. Finney, Calculus and Analytic Geometry, Addison- Wesley Publishers,					
	1999						
2.	Journals & Period	licals:					
3.	Other Electronic	Resources:					
	GeoGebra Toolbo	ox: https://www.geogebra.org/					
Evaluation	on Scheme	Total Marks					
Theory: Mid semester Marks		20 marks					
Theory: End So	emester Marks	40 marks					



Theory: Continuous Evaluation	Attendance	05 marks	
Component Marks	MCQs	10 marks	
	Open Book Assignment	15 marks	
	Open Book Assignment	10 marks	
	Total	40 Marks	
Practical Marks	Attendance	05 marks	
	Practical Exam	20 marks	
	Viva	10 marks	
	Journal	10 marks	
	Discipline	05 marks	
	Total	50 Marks	
Project/ Industrial Internship Marks	Quantity of the Project/Industrial in terms of Language, Presentation & format.	30 marks	
	Practical understanding of the subject on the Project/Industrial.	30 marks	
	Industry/ University mentor's feedback on the Project/ Industrial.	30 marks	
	Attendance	10 marks	
	Total	100 Marks	

Mapping of PSOs & COs

	PSO1	PSO2	PSO2	PSO3	PSO4	PSO5	PSO6	PSO7	PSO8
CO1	1	2	0	1	0	1	1	2	0
CO2	2	2	0	0	1	1	2	2	0
CO3	2	2	0	0	1	1	2	2	0
CO4	2	2	1	0	1	1	2	2	1
CO5	3	3	2	1	0	3	3	3	2

Mapping of POs & COs

TI 8	PO1	PO2	PO2	PO3	PO4	PO5	PO6	PO7	PO8
CO1	2	2	0	1	1	0	2	2	0
CO2	1	2	1	1	0	0	1	2	1
CO3	1	2	1	1	0	0	1	2	1
CO4	1	2	1	2	0	0	1	2	1
CO5	1	1	3	2	0	0	1	1	3

^{1:} Slight (low); 2: Moderate (Medium); 3: Substantial (High); 0 None



COURSE CODE	COURSE NAME	SEMESTER
CBCS103	Indian History and	I
	Culture	

Teaching Scheme (Hours)			Tea	ching Credit			
Lecture	Practical	Tutorial	Total Hours	Lecture	Practical	Tutorial	Total Credit
30	0	30	2	2	0	0	2

Course Category	An inquisitive mind, a voracious appetite for reading, a knack for analysis, good vocabulary, willingness to learn and accept different school of thoughts. Generic Elective
Course focus	
Rationale	
Course Revision/ Approval Date:	28-03-2020
Course Objectives (As per Blooms' Taxonomy)	 Study the chronology of major events that happened in India Understand the political, social economic, and cultural landscape of India and how it kept on changing and evolving Understand the various characters that shaped the destiny of India Explore the reasons why India kept on attracting people from all over the world.

	Course Content (Theory)	Weightage	Contact hours
Uni	t 1: Ancient India The Harappan Civilization Early States and Economies (c.600 BCE-600 CE) Early Societies (c. 600 BCE-600 CE)	20%	6
	t 2: Culture & Religion Cultural Developments Perceptions of Society Changes in Religious Beliefs and Devotional Texts	20%	6
Uni	An Imperial Capital: Vijayanagara Agrarian Society and the Mughal Empire The Mughal Courts	20%	6



Unit	4: Colonialism		
•	Colonialism And The Countryside: ExploringOfficial Archives	20%	6
•	1857 Revolt and its Representation		
Unit	5: Freedom struggle		
•	Civil Disobedience and Beyond	20%	6
•	Partition: Politics, Memories, Experiences		
•	Framing The Constitution: The Beginning of a NewEra		

Instructional Method and Pedagogy: History involves a lot of storytelling and analysis. Hence, it will be mix of stories, videos, maps, discussions, role plays, power point presentations, timelines

Learning	Resources					
1.	Reference Books:					
	Bashar A. L., 1954, The Wonder That was India					
	• Sarkar Sumit, Modern India, 1885-1948					
	Baron De, Bipan Chandra and Amlesh Tripathi, Freedom Struggle, NBT, Delhi					
	Grover B. L., A New Look on Modern India History, Delhi					
2.	Journals & Periodicals:					
	Indian Economic and Social History Review: SAGE Journals					
	Indian Historical Review: SAGE Journals					
3.	Other Electronic Resources:					
	https://knowindia.gov.in/culture-and-heritage/ancient-history.php					
	https://en.wikipedia.org/wiki/Culture_of_India https://www.livescience.com/28634-					
	indian-culture.html https://www.everyculture.com/Ge-It/India.html					
	https://www.youtube.com/watch?v=1K3oqU2WNY0					
	https://www.youtube.com/watch?v=5TvF-kfqlUI					
	https://www.youtube.com/watch?v=VL9Xd9fhnh4					

Evaluation Scheme	Total Marks			
Theory: Mid semester	20 marks			
Marks				
Theory: End Semester	40 marks			
Marks				
Theory: Continuous	Attendance	05 marks		
Evaluation Component	MCQs	10 marks		
Marks	Open Book Assignment	15 marks		
	Article Review	10 marks		
	Total	40 Marks		



COURSE CODE CBCS104	COURSE NAME Air Pollution Control	SEMESTER I

Teaching Scheme (Hours)			Tea	ching Credit			
Lecture	Practical	Tutorial	Total Hours	Lecture	Practical	Tutorial	Total Credit
30	0	30	2	2	0	0	2

Course Pre-requisites	Environmental Studies
Course Category	Generic Elective
Course focus	
Rationale	
Course Revision/	28-03-2020
Approval Date:	
Course Objectives	1. Impart Fundamental knowledge about awareness of
(As per Blooms'	air pollution
Taxonomy)	2. Understand the various sources which contribute in degradation of air quality
	3. Understand assessing the air quality through air quality index
	4. Understand various air pollution control methods and equipment's used by industries
	5. Understand the importance of extreme air Pollution due to Acid Rain, Global Warming in Environment.

Course Content (Theory)	Weightage	Contact hours
Unit 1: Introduction to Air Pollution: Air and its composition, Air Pollution, Sources of air pollution and its classification, Major air Pollutants and their characteristics, Specific group pollutants such as CFC, GHG etc. Air Pollutants from various industrial sectors. Impact of air pollution on human health and vegetation.	10%	4



II '4 A D.H. 4. A D'annai'r a		
Unit 2: Pollutant Dispersion: Concept of atmospheric stability. Adiabatic and Environmental Lapse rate. Plume behavior. Effect of topography, terrain and structure on Pollutant dispersion. Effect of wind on Pollutant dispersion. Concept of Maximum mixing Depth and ventilation Coefficient. Plume Rise and Effective stack height.	20%	6
Unit 3: Air Quality: Introduction to Air quality index and Comprehensive Environmental Pollution Index etc. and its application. Sampling and measurement of air pollutants. Introduction to National Ambient AirQuality Standards	20%	6
Unit 4: Impacts of Air Pollution: Extreme air Pollution scenarios: Acid Rain, Global Warming, Smog, Ozone layer depletion etc. Various treaties and protocols: Kyoto Protocol and Montreal Protocol etc	20%	6
Unit 5: Control methods and equipment: Introduction to control methods and equipment for Particulate matter and gases. Design and working of scrubbers, Electrostatic Precipitator, Gravity settlers, Cyclone separator, Filter bags etc. Other mechanisms of	30%	8

Instructional Method and Pedagogy: Presentation, Videos, Chalk-Duster and Notes

air pollution control such as Biochemical Processes, catalytic

Learning Resources				
1.	Reference Books:			
	 Air Pollution. Physical and Chemical Fundamentals, Sainfeld, J.H. McGraw Hill, N.Y. 1975. 			
	 Environmental Pollution Control and Engineering, Rao C.S., NewAge International (P) Limited, 1st Ed., 1991. Air Pollution, Perkin, H.G. McGraw Hill 1974 			
2.	Journals & Periodicals: Environmental Pollution, Environmental Science &			
	Technology			
3.	Other Electronic Resources: http://www.envis.nic.in			

Evaluation Scheme	Total Marks			
Theory: Mid semester	20 marks			
Marks				
Theory: End Semester	40 marks			
Marks				
Theory: Continuous	Attendance 05 marks			
Evaluation Component	MCQs 10 marks			
	Open Book Assignment	15 marks		

Processes etc.

School of Science B.Sc. Microbiology, Course Curriculum Academic Year, 2024-25



Marks Article Review 10 marks Total 40 Marks



COURSE CODE CBCS105	COURSE NAME Python Programming	SEMESTER I		

Teaching Scheme (Hours)			Tea	ching Credit			
Lecture	Practical	Tutorial	Total Hours	Lecture	Practical	Tutorial	Total Credit
30	0	30	2	2	0	0	2

Course Pre-requisites	Introduction to Programming Concepts
Course Category	Generic Elective
Course Focus	
Rationale	
Course Revision/ Approval Date:	28-03-2020
Course Objectives (As per Blooms'	1. Application development using Python Understand therapeutic aspects of various drugs
Taxonomy)	 Learning of Python Libraries Learning analysis of data using Python
	4. Use of Python in Research

Course Content (Theory)	Weightage	Contact
		hours
Unit 1: Introduction to Python Programming: Theory: The Context of Software Development – Software	20%	6
Learning Programming with Python		
Unit 2: Theory: Values & Variables, Expressions, Arithmetic, Conditional Statements	20%	6
Unit 3:		
Theory: Functions, Objects	20%	6
Unit 4: Theory: Lists, Dictionaries	20%	6
Unit 5: Theory: Tuples, Class Design	20%	6

Instructional Method and Pedagogy: Computer based learning Presentation, Chalk-Duster and Notes



Course Outcome:*

After successful completion of the above course, studentswill be able to:

CO1: Develop applications using Python

CO2: Students will work on multidiscipline projects

CO3: They will work on various libraries

CO4: Students can take up research projects

Learning	Resources
1	Reference Books:
	1.
2	Journals & Periodicals:
3	Other Electronic Resources:

Evaluation Scheme	Total Marks		
Theory: Mid	20	marks	
semester Marks			
Theory: End	40	marks	
Semester Marks			
Theory: Continuous	Attendance	05 marks	
Evaluation	MCQs	10 marks	
ComponentMarks	Open Book Assignment	15 marks	
	Article Review	10 marks	
	Total	40 Marks	



COURSE CODE	COURSE NAME	SEMESTER
CBCS109	Management Principles	I
	& Practices	

Teaching Scheme (Hours)			Teaching Credit				
Lecture	Practical	Tutorial	Total Hours	Lecture	Practical	Tutorial	Total Credit
30	0	30	2	2	0	0	2

Course Pre-requisites	H.S.C. from any stream
Course Category	Generic Elective
Course focus	
Rationale	
Course Revision/ Approval Date:	24-03-2020
Course Objectives (As per Blooms' Taxonomy)	 Impart Fundamental principles of Management Understand basic management functions to accomplish businessobjectives. Understand the various functional areas of management.

Course Content (Theory)	Weightage	Contact hours
Unit 1: Introduction to Management		110 011
Definition, Functions, Process, Scope and Significance of Management, Nature of Management, Managerial Roles and Managerial Skills, Difference between Management and Administration.	20%	6
Unit 2: Evolution of management thought: Early Management, Classical Approach - Scientific Management, Administrative Management, Neo - Classical Approach - Hawthorne Experiments, Modern Approach- System and Contingencyapproach	20%	6
Unit 3: Planning Meaning & Definition of Planning, Nature and Importance of Planning, Planning Process, Types of Plans	15%	4
Unit 4: Organizing & Staffing Nature & Significance of Organization, Organization Structures, Types, Advantages & Disadvantages., Recruitment & Selection	25%	8



Unit 5: Direction & Control		
Direction : Meaning, Definition, Features, Principles.	20%	6
Control: Meaning, Definitions, Process, Reasons for Resistance to control,		
Methods: TQM, Budgetary Control, Break Even Control, Kaizen, Six		
Sigma		

Instructional Method and Pedagogy: Presentation, Case study, Assignment, Chalk-Duster and Notes

Course Outcome:*	Blooms' Taxono my Domain *	Blooms' Taxonomy SubDomain*
After successful completion of the above course, studentswill be able to:		Provide
CO1: Provide a broad and integrative introduction to the theories and practices of Management.		
CO2: Understand the role, challenges, and opportunities of management in contributing to the successful operations and performance of organizations.	Cognitive	Understand
CO3: Expand the basic areas of the management process and functions from an organizational viewpoint		Expand

	Learning Resources
1.	Reference Books:
	1. Harold Koontz and Heinz Weihrich, Essentials of Management: An International and Leadership Perspective, McGraw Hill Education.
	2. Stephen P Robbins and Madhushree Nanda Agrawal, Fundamentals of Management: Essential Concepts and Applications, Pearson Education.
	 George Terry, Principles of Management, Richard D. Irwin Griffin, Management Principles and Application, Cengage Learning
	5. Peter F Drucker, Practice of Management, Mercury Books, London
2.	Journals & Periodicals: 1. Journal of Management 2. Prabandhan: Indian Journal of Management 3. Journal of International Management 4. Journal of Management and Organization
3.	Other Electronic Resources: www.omicsonline.org



Evaluation Scheme	Total Marks			
Theory: Mid semester	20 marks			
Marks				
Theory: End Semester	40 marks			
Marks				
Theory: Continuous	Attendance	05 marks		
Evaluation Component	MCQs	10 marks		
Marks	Open Book Assignment	15 marks		
	Article Review	10 marks		
	Total	40 Marks		



COURSE CODE	COURSE NAME	SEMESTER
CBCS110	MS Office &	I
	Applications	

Teaching Scheme (Hours)			Teaching Credit				
Lecture	Practical	Tutorial	Total Hours	Lecture	Practical	Tutorial	Total Credit
30	0	30	2	2	0	0	2

Course Pre-requisites	Basic knowledge of computer, typing.
Course Category	Generic Elective
Course focus	
Rationale	
Course Revision/	28-03-2020
Approval Date:	
Course Objectives	1. Create and edit any text document in MS Word.
(As per Blooms' Taxonomy)	2. Impart the knowledge and awareness of creating formula's for estimation and charts for data interpretation in MS Excel.
	3. Create at the level of detail that is appropriate for your project using MS Project.
	4. Create the presentations that include text, graphics, animation, and transitions.
	5. Understand the working around email in MS Outlook.

Course Content (Theory)	Weightage	Contact
		hours
Unit 1: Introduction About MS Word	20%	6
Creating, editing, saving and printing text documents, Font and		
paragraph formatting, Simple character formatting, Inserting tables,		
smart art, page breaks, Using lists and styles, Working with images,		
Using Spelling and Grammar check, Understanding document properties,		
Mail Merge		



Unit 2:		
Spreadsheet basics, Creating, editing, saving and printing spreadsheets,	20%	6
Working with functions & formulas, Modifying worksheets with color &		
auto - formats, Graphically representing data: Charts & Graphs,		
Speeding data entry: Using Data Forms, Analyzing data: Data Menu,		
Subtotal, Filtering Data, Formatting		
worksheets, Securing & Protecting spreadsheets		
Unit 3: MS Project	20%	6
Introduction, Initiation, Planning, Execution, Controlling, Closure.		
Unit 4: MS PowerPoint	20%	6
Opening, viewing, creating, and printing slides, Applying auto layouts,		
Adding custom animation, Using slide transitions, Graphically		
representing data: Charts & Graphs, Creating Professional Slide for		
Presentation.		
Unit 5: MS Outlook	20%	6
Introduction to Microsoft Outlook, Configuring Outlook, Creating and		
Sending Email Messages, Organizing Mail, Using the Calendar, Setting		
Reminders, Sharing and Alerts.		

Instructional Method and Pedagogy: Presentation, Videos, Chalk-Duster and Notes

Course Outcome:*	Blooms' Taxonom yDomain*	Blooms' Taxonomy SubDomain*
After successful completion of the above course, studentswill be able to:		Create
CO1: Create a format and edit documents likeletters, report, applications, templates, etc. in MS Word.		
CO2:Apply computational methods in MS Excel for estimation, optimization of various properties and work on the data	Cognitive	Apply
interpretation based on charts. CO3: Create a standardize and manage the work break down structure consisting of tasks, costs, work, and resources.		Create
CO4: Create a presentation based on students' option, ideas, projectdetails, etc. using MS PowerPoint.		Create
CO5 Create own address, constitute text body for email, provide necessary attachments, schedule events and meeting, etc. using MS Outlook.		Create

Learning Resources



1.	Reference Books:
	1. Microsoft Office 2016, by Joan Lambert and Curtis Frye, MicrosoftPress.
	2. Microsoft Office 2010 on demand, by Steve Johnson, PearsonEducation India.
	3. Microsoft Office 2016 all-in-one for dummies, by Wallace Wang, Wiley.
	4. Microsoft Office 2019 for dummies, by Wallace Wang, Wiley.
	5. Mastering MS Office, by Bittu Kumar, V&S Publishers.
	6. MS Office, by S. S. Shrivastava, Laxmi Publications.
2.	Journals & Periodicals:
3.	Other Electronic Resources: https://edu.gcfglobal.org/en/subjects/office/

Evaluation Scheme	Total Marks	
Theory: Mid semester	20 marks	
Marks		
Theory: End Semester	40 marks	
Marks		
Theory: Continuous	Attendance	05 marks
Evaluation Component	MCQs	10 marks
Marks	Open Book Assignment	15 marks
	Article Review	10 marks
	Total	40 Marks



COURSE CODE	COURSE NAME	SEMESTER- I
AECC101	Fundamentals of English	

Teaching Scheme (Hours)		Teaching Credit					
Lecture	Practical	Tutorial	Total hours	Lecture	Practical	Tutorial	Total Credit
30	00	00	30	2	0	0	2

Course Pre-	S	tudent shoul	d have cleared	l 12th Scien	ce		
requisites							
Course Categ	gory N	Iandatory C	ourse				
Course focus	S	Skills Development					
Rationale	It	enables hu	manity to exp	erience the	benefits of	chemistry who	en we
	aj	oply it in the	exploitation	of materials	and energy.		
Course Revis	ion/						
Approval Dat	te:						
Course Object	tives T	o enable the	student to:				
(As per Bloom	ns' 1	. Emphasize	the developm	ent of listen	ing and read	ing skills amor	ng
Taxonomy)	le	arners					
	2	. Equip then	n with writing	skills neede	ed for academ	nic as well as	
	W	orkplace co	ntext				
	3	. Enable lea	rners of science	e develop t	heir basic cor	mmunication s	kills in
	Е	nglish					
	4.	. Strengthen	the fundamen	tals in Engl	ish Language	•	
	5	. Build up tl	ne confidence	to communi	cate with the	world.	

Course Content	Weightage	Contact hours
Unit 1: Language Basics Parts of speech, word formation, prefix- suffix, synonyms, antonyms, homophones and standard abbreviations	20%	6
Unit 2: Elementary Reading/Writing Skills Types of the sentences, structures of the sentences, use of phrases and clauses, punctuation, creative writing and coherence, comprehension, essay, paragraph writing, creative writing	30%	9
Unit 3: Elementary Spoken Skills Greetings, farewell and introduction, making an apology, accepting an apology, making an appointment, JAM	30%	9
Unit 4: Presentation Skills Group Discussion, Debate, Public Speaking, Discussion on a specific purpose.	20%	6



Unit 5: Practicing and Identifying the Common Error Tense, subject-verb agreement, noun-pronoun agreement, articles, prepositions, modal auxiliaries, voice, reported speech	20%	6
articles, propositions, moder durinaries, voice, reported speceri		

Instructional Method and Pedagogy:

Classroom Lecture, Case Studies, Quizzes, Presentations, Role Play, Expert Lecture (Consultant)

Course Outcomes:	Blooms' Taxonomy	Blooms' Taxonomy Sub
	Domain	Domain
After successful completion	of the above course, stude	ents will be able to:
CO1: To emphasize the development of	Understand, Analyse,	Define, Classify &
listening and reading skills among	Remember	Demonstrate
learners		
CO2: To equip them with writing skills	Analyse, Apply,	Classify, Describe &
needed for academic as well as	Understand	Demonstrate
workplace context		
CO3: To enable learners of	Understand,	Define, Describe &
Engineering and Technology to	remember	Demonstrate
develop their basic communication		
skills in English		
CO4: To strengthen the fundamentals in	Remember, Analyse	Define Describe
English Language.		
CO5:To build up the confidence to	Understand, Apply	Define, Classify, Describe
communicate with the world.		& Demonstrate

Learning Resources

1. Reference Books:

- 1. Murphy, Raymond "Murphy's English Grammar with CD" Cambridge University Press, 2004.
- 2. Thorpe, Edgar and Showick Thorpe "Basic Vocabulary" Pearson Education India, 2012.
- 3. Green, David. "Contemporary English Grammar Structures and Composition" MacMillan Publishers, New Delhi, 2010.
- 4. Wren & Martin (2001), English Grammar & Composition, New York

3. Journal & Periodicals

- 1. The Journal' Basic English Grammar
- 2. Fluent U' English Language and Cultural Journal
- 3. The Journal of English Academics'
- 4. Elsevier' The research on language
- 5. Index Noedicus: A Cumulative Index to English Language Periodicals
- 6. The Illustrated English Language Periodicals

4. Other Electronic Resources

- 1. Wordsworth Language software
- 2. Jam board



Evaluation Scheme	Total Ma	rks		50 Marks					
Theory: Mid semester		20 marks							
Marks									
Theory: End Semester	40 marks								
Marks									
Theory: Continuous		Attendance	05 marks						
Evaluation Component		MCQs	10 marks]					
Marks		Skill enhancement activities /	15 marks						
		case study							
		Presentation/ miscellaneous	10 marks	1					
		activities							
		Total	40 Marks						

Mapping of PSOs & Cos

	PSO1	PSO2	PSO3	PSO4	PSO5	PSO6
CO1	1	1	1	0	3	3
CO2	1	1	1	0	3	3
CO3	1	1	1	0	3	2
CO4	1	1	1	0	3	3
CO5	1	1	1	0	3	3

Mapping of POs & Cos

	PO1	PO2	PO3	PO4	PO5	PO6
CO1	3	2	2	3	2	0
CO2	3	3	3	3	2	0
CO3	3	2	3	3	2	0
CO4	3	1	3	3	3	0
CO5	3	2	2	3	2	0

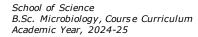


		Semester – II					
Sr. No.	Course Code	Course Title	L	Т	P	C	Mark
		A. Major					
1	BSMO211	Biomolecules	3	0	1	4	150
2	BSMO212	Metabolism	3	0	1	4	150
	1	B. Minor					
3	BSCM216	Basics of Chemistry – II	2	0	1	3	75
4.	BSPY216	Physics – I	2	0	1	3	75
C. N	Tultidis ciplinary	Additional Credits under Choice Ba	sed C	redit	Syste	m (Ar	y One)
4	CBCS202	Introduction to Social Work	2	0	0		
5	CBCS203	Web Design Fundamentals	2	0	0		
6	CBCS204	Introduction to Design Engineering	2	0	0		
7	CBCS205	Introduction to Automotive System	2	0	0	2	100
8	CBCS211	Accounting for Non – Management Students	2	0	0		
9	CBCS212	Personal Finance	2	0	0		
	1	D. Ability Enhancement Cour	rse				
10	AECC201	Communication Skills in English	2	0	0	2	50
		E. Skill Enhancement Cours	e				
11	SECC201	Internship	0	0	2	2	50
	1	F. Value Added Course	1	1	1		
12	VACC201	Vedic Mathematics /NCC/Sports	0	0	2	2	50
				7	otal	22	700



Teaching Scheme Semester – II

Sr.	Course			achin (Hour			Т	eachi'	ng C	redit			Evaluati on	Scheme		
No.	Code	Course Name	L	P	Т	Total	L	P	Т	Total	Theory: MS Marks	Theory: CEC Marks	Theory: ES Marks	Theory Marks	Practical Marks	Total Marks
	A. Major Courses															
1.	BSMO211	Molecular Biology	3	2	0	5	3	1	0	4	20	40	40	100	50	150
2.	BSMO212	Metabolism	3	2	0	5	3	1	0	4	20	40	40	100	50	150
	B. Minor Courses															
3.	BSCM216	Basics of Chemistry - II	2	2	0	4	2	1	0	3	20	45	75	150	0	150
4.	BSPY216	Physics - I	2	2	0	4	2	1	0	3	20	45	75	150	0	150
						C. 1	Multi	discip	linary	Course	s (Any one)					
5.	CBCS202	Introduction to Social Work														
6.	CBCS203	Web Design Fundamentals														
7.	CBCS204	Introduction to Design Engineering	2	0	0	2	2	0	0	2	20	40	40	100	00	100
8.	CBCS205	Introduction to Automotive System														
9.	CBCS211	Accounting for Non - Management Students														
10.	CBCS212	Personal Finance														
							D. Al	bility	Enha	ncement	Courses					
11.	AECC201	Communication Skills in English	2	0	0	2	2	0	0	2	20	40	40	100	00	100





	E. Skill Enhancement Courses															
12.	SECC201	Industrial Internship	0	2	0	2	0	2	0	2	00	00	00	50	00	50
	F. Value Added Courses															
13.	VACC101	NCC/Sports	0	2	0	0	0	2	0	2	00	00	00	50	00	50
	Total									22						700

Note: L = Lecture, P = Practice, T= Tutorial, MS - Mid Semester, CEC - Continuous Evaluation Component, ES - End Semester



COURSE CODE BSMO211	COURSE NAME Molecular Biology	SEMESTER- I
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	Teaching	Scheme (Hou	rs)	Tea			
Lecture	Practical	Tutorial	Total hours	Lecture	Practical	Tutorial	Total Credit
45	30	00	75	3	1	0	4

Course Pre	School Level Understanding of Biology and a keen interest in learning.
-requisite	
Course Category	Core Professional
Course Focus	Employability
	The molecular biology course explores fundamental aspects of genetic material, replication, transcription, and translation. It delves into DNA's role as hereditary material, its structures in diverse organisms, replication mechanisms, transcription processes, and translation machinery, fostering a deep understanding of gene expression.
Course Revision/ Approval date	07/11/2023
(As per Blooms' Taxonomy)	To enable the student to: 1. Understand the historic perspective, types and structure of genetic material. 2. Gain knowledge about DNA replication mechanism in both prokaryotes and eukaryotes 3. Comprehend the process of transcription in prokaryotes and eukaryotes 4. Familiarize with post-transcriptional processing. 5. Develop an understanding of translation in both prokaryotes and eukaryotes.

Course Content	Weightage	Contact hours
Unit 1: Structures and types of Genetic Material DNA as heritable material, Structure and types of DNA in prokaryotes	20%	9
and eukaryotes, Salient features of double helix, Genome organization		
in Prokaryotes and Eukaryotes.		
Unit 2: Replication of DNA	20%	9
Types of replication in Prokaryotes and Eukaryotes, Enzymes and		
proteins involved in DNA replication. Mechanism of DNA replication		
in Prokaryotes and Eukaryotes, Telomere replication, Various		
models of DNA replication.		



Unit 3: Transcription in Prokaryotes and Eukaryotes	20%	9
Structure, Function and Biological Properties of RNA, The structure		
and function of gene, promoters and terminators. Transcription		
Initiation, elongation and Termination, RNA polymerases.		
Unit 4: Post-Transcriptional Processing	20%	9
Concept of introns and exons, RNA splicing, concept of alternative		
splicing, Polyadenylation and capping, Processing of rRNA.		
Unit 5: Translation (Prokaryotes and Eukaryotes)	20%	9
The Genetic Code, Codons and anti-codons, the ribosomes.		
Translation mechanism in prokaryotes and eukaryotes. Gene		
Expression in Prokaryotes: the LAC Operon. Gene expression in		
Eukaryotes.		

List Of Practical	Weightage	Contact hours
1: Good Laboratory Practice and Safety in Molecular Biology Lab.	2%	2
2: To study the principle and applications of important instruments (pH meter, Centrifuge, Spectrophotometer, Electrophoretic unit) used in the Molecular Biology laboratory.	2%	4
3: To study the principle and applications of important instruments (pH meter, Centrifuge, Spectrophotometer, Electrophoretic unit) used in the Molecular Biology laboratory.	2%	4
4: Pipetting skills	2%	2
5: Basic calculations (normality, molarity, weight conversion) used in molecular biology	2%	2
6: Preparation of solutions for Molecular Biology experiments	2%	2
7: Virtual Lab demonstrations: DNA Isolation, Purification and quantitation	2%	2
8: Virtual Lab demonstrations: RNA Isolation, Purification and quantitation	2%	2
9: Virtual Lab demonstrations: Protein Isolation, Purification and quantitation	2%	2
10: Model Building -DNA structure, replication	2%	2
11: Revision	2%	2
12: Revision	2%	2
13: Revision	2%	2

Instructional Method and Pedagogy:
Audio-Visual Lectures, Quizzes, Debates, Project works, Case studies, and Assignments. Practical exercises are designed to understand the theory as taught in classroom.



Course Outcomes:	Blooms' Taxonomy Domain	Blooms' Taxonomy Sub Domain
After successful completion of the above course, students will be able to: CO1: Understand basic concepts of molecular biology including genome, DNA, RNA	Remember, Understanding	Describe
CO2: Understand and compare the mechanisms of DNA replication in both prokaryotes and eukaryotes and differentiate between them.	Remember, Understanding, Analyze	Explain, Compare
CO3: Understand and compare the mechanisms of DNA transcription in both prokaryotes and eukaryotes and differentiate between them.	Remember, Understanding, Analyze	Explain, Compare
CO4: Understand and compare the post transcriptional modifications of RNA and concepts of splicing and capping.	Remember, Understanding, Analyze	Explain, Describe, Compare
CO5: Understand and compare the mechanisms of translation in both prokaryotes and eukaryotes and differentiate between them	Remember, Understanding, Analyze	Describe, Compare

Learning Resources

1. Reference books

- 1. Molecular Biology of the Cell by Bruce Alberts, Alexander Johnson, Julian Lewis, et al.
- 2. Principles of Molecular Biology by Burton E. Tropp
- 3. Essential Cell Biology by Bruce Alberts, Dennis Bray, Karen Hopkin and Alexander Johnson.
- 4. Cooper, G.M. and Hausman, R.E. The Cell: A Molecular Approach. V Edition. ASMPress
- Molecular Biology by David P. Clark and Nanette J. Pazdernik

2. Journal & Periodicals

- 1. Journal of Molecular Biology
- 2. Nucleic Acid Research
- 3. Molecular Biology Reports
- 4. Current Science
- 3. Other Electronic resources: NPTEL

Evaluation Scheme	Total Marks
The M: 1	20
Theory: Mid semeste	er 20 marks
Marks	
Theory: End Semeste	er 40 marks
Marks	



Theory: Continuous Evaluation Component	A.:. 1	05 1
Marks	Attendance	05 marks
	MCQs	10 marks
	Open Book Assignment	15 marks
	Research Paper Review	10 marks
	Total	40 Marks
Practical Marks		
Tractical Walks	Attendance	05 marks
	Practical Exam	35 marks
	Viva	10 marks
	Journal	05 marks
	Total	50 Marks

Mapping of PSOs & COs

	PSO1	PSO2	PSO3	PSO4	PSO5	PSO6
CO1	3	2	1	1	1	1
CO2	2	1	-	-	1	1
CO3	2	1	-	-	1	1
CO4	2	1	-	-	1	1
CO5	2	1	-	-	1	1

Mapping of POs & COs

	PO1	PO2	PO3	PO4	PO5	PO6
CO1	3	2	2	1	-	2
CO2	1	1	-	1	1	1
CO3	1	1	-	-	1	1
CO4	1	1	-	-	-	1
CO5	1	1	-	-	-	1



COURSE CODE	COURSE NAME	SEMESTER- II
BSMO212	Metabolism	

	Teaching	Scheme (Hour	rs)	Tea	ching Credit		
Lecture	Practical	Tutorial	Total hours	Lecture	Practical	Tutorial	Total Credit
45	30	00	75	3	1	0	4

Course Pre-	School Level Understanding of Biological molecules pathways and a keen
requisite	interest in learning.
Course Category	Core Professional
Course Focus	Employability
Rationale	This course provides a comprehensive study of cellular metabolism, covering
	the breakdown and synthesis of carbohydrates, lipids, amino acids, and
	nucleotides. Students will gain insights into energy production and
	biomolecule regulation, enhancing their understanding of vital biological
	processes.
Course Revision/	09/11/2023
Approval date	
Course Objectives	To enable the student :
(As per	1. To remember, understand and analyze the knowledge of carbohydrate
Blooms'	metabolism.
Taxonomy)	2. To apply the knowledge of carbohydrate metabolism to explain cellular
	respiration process.
	3. To remember, understand and apply the lipid metabolism.
	4. To remember, understand and apply the amino acid metabolism.
	5. To remember, understand and apply the nucleic acids metabolism.

Course Content	Weightage	Contact
		Hours
Unit 1: Carbohydrate metabolism-I	20%	9
Glycolysis: Fate of pyruvate under aerobic and anaerobic conditions		
and regulation. HMP, PPP, Gluconeogenesis, Glycogenolysis and		
glycogen synthesis.		
Unit 2: Carbohydrate metabolism-II	20%	9
TCA cycle, it's central role, interactions, organization, reactions,		
regulation. Electron Transport Chain, Oxidative phosphorylation		
Unit 3: Lipid metabolism	20%	9
Triglyceride, cholesterol, phospholipid, lipoproteins and		
eicosanoids.ß-oxidation of fatty acids, ketogenesis, Fatty Acid		
synthesis. Synthesis of membrane phospholipids in prokaryotes and		
eukaryotes.		
Unit 4: Amino acid metabolism	20%	9
Nitrogen cycle, incorporation of ammonia into biomolecules.		
Transamination, Deamination and Urea Cycle. Overview of amino		



acid synthesis		
Unit 5: Nucleotide metabolism	20%	9
Metabolic specializations in Microorganisms. De novo synthesis of		
purine and pyrimidine nucleotides. Biosynthesis of		
deoxyribonucleotides		

List Of Practical	Weightage	Contact
		hours
1. Sugar fermentation of microorganisms.	2%	2
2. Estimation of reducing and non-reducing sugar	2%	2
3. Blood glucose estimation	2%	2
4. Estimation of HDL & LDL	2%	2
5. Estimation of serum urea	2%	2
6. Estimation of serum uric acid.	2%	2
7. Estimation of serum creatinine.	2%	2
8. Demonstration of enzyme assay	2%	2
9. Detection of amino acid by paper chromatography	2%	2
10. Determination of pyruvate by 2,4- dinitrophenyl hydrazine	2%	2
method		
11. Measurement of free radicals by spectrophotometric method	2%	2
12. Revision	2%	2
13. Revision	2%	2
14. Revision	2%	2
15. Revision	2%	2

Instructional Method and Pedagogy:

Audio-Visual Lectures, Quizzes, Debates, Project works, Case studies, and Assignments. Practical exercises are designed to understand the theory as taught in classroom.

Learning Resources

- 1. Reference Books
 - 1. Berg, J. M., Tymoczko, J. L. and Stryer, L. (2006). Biochemistry. VI Edition. W.H Freeman andCo.
 - 2. Buchanan, B., Gruissem, W. and Jones, R. (2000) Biochemistry and Molecular Biology of Plants. American Society of Plant Biologists.
 - 3. Nelson, D.L., Cox, M.M. (2004) Lehninger Principles of Biochemistry, 4th Edition, WH Freeman and Company, New York, USA.
 - 4. Biochemistry by U Satyanarayan
- 2. Journals & Periodicals:
 - 1. JBC
 - 2. Current Science
- 3. Other Electronic resources:
 - 1. NPTEL

Course Outcomes:	Blooms' Taxonomy Domain	Blooms' Taxonomy Sub Domain
After successful completion of the above	e course, students will be able t	to:



CO1: Analyze glycolysis pathways, evaluate pyruvate fate in aerobic and anaerobic conditions, and explain regulation mechanisms in glycolysis-related processes.	Understand, Analyse, Remember	Evaluate, Classify & Demonstrate
CO2: Analyze the TCA cycle's central role, its interactions, reactions, and regulation, and evaluate the electron transport chain and oxidative phosphorylation.		Classify, Describe & Demonstrate
CO3: Analyze lipid structures including triglycerides, cholesterol, and phospholipids, and evaluate metabolic pathways such as β-oxidation, ketogenesis, and fatty acid synthesis.	•	Define, Describe & Demonstrate
CO4: Analyze the nitrogen cycle, evaluate the incorporation of ammonia into biomolecules, and explain transamination, deamination, and the urea cycle.	Remember, Analyze	Explain, Describe
CO5: Understand metabolic adaptations in microorganisms, evaluate de novo synthesis pathways of purine and pyrimidine nucleotides, and explain deoxyribonucleotide biosynthesis.	Understand, Apply	Define, Classify, Describe & evaluate

Evaluation Scheme	Total Marks = 150					
Theory: Mid semester Marks	20 marks					
Theory: End Semester Marks	40 marks					
Theory: Continuous Evaluation	Attendance	05 marks				
Component Marks	MCQs	10 marks				
	Open Book Assignment	15 marks				
	Research Paper Review	10 marks				
	Total 40 Marks					
Practical Marks						
	Attendance	05 marks				
	Practical Exam	35 marks				
	Viva	10 marks				
	Journal	05 marks				



Total	50 Marks

Mapping of PSOs & Cos

	PSO1	PSO2	PSO3	PSO4	PSO5	PSO6
CO1	2	2	2	0	0	1
CO2	2	2	2	0	0	1
CO3	2	2	2	0	0	1
CO4	2	2	2	0	1	1
CO5	2	2	2	0	1	1

Mapping of POs & Cos

	PO1	PO2	PO3	PO4	PO5	PO6
CO1	3	2	1	0	1	1
CO2	3	2	1	0	1	1
CO3	3	2	1	0	1	0
CO4	2	2	1	0	0	1
CO5	2	2	0	0	0	1



COURSE CODE	COURSE NAME	SEMESTER- II
BSCM216	Basics of Chemistry - II	

	Teaching	Teaching Scheme (Hours)		Teaching Credit			
Lecture	Practical	Tutorial	Total hours	Lecture	Practical	Tutorial	Total Credit
30	30	00	60	2	1	0	3

Course Pre-	Basic knowledge of general chemistry.			
requisites				
Course Category	Generic Elective			
Course Focus	Skill Development			
Rationale	This course aims to provide a comprehensive understanding of fundamental			
	and advanced concepts in electrochemistry, organic chemistry, coordination			
	chemistry, stereochemistry, and organic reaction mechanisms, preparing students for further studies or careers in chemistry-related fields.			
Course Revision/	07/03/2024			
Approval date				
Course Objectives	To enable the student to:			
(As per Blooms'	1. Knowledge: Describe the principles and applications of			
Taxonomy)	electrochemistry, organic reactions, coordination compounds,			
	stereochemistry, and organic reaction mechanisms.			
	2. Comprehension: Interpret electrode potentials, Nernst equation, reaction			
	mechanisms, and stereochemical representations.			
	3. Application: Apply knowledge to solve problems related to			
	electrochemical cells, organic reactions, coordination complexes, and			
	stereochemical configurations.			
	4. Analysis: Analyze and evaluate redox reactions, reaction mechanisms,			
	and properties of coordination compounds.			
	5. Synthesis: Design experiments and propose mechanisms for organic			
	reactions, and predict properties and applications of coordination			
	compounds.			

Course Content	Weightage	Contact
		hours
Unit 1: Electrochemistry	20%	6
Electrochemistry: Electrode potential, related problems, Nernst		
equation & its applications, emf of the cell, related problems, Redox		
reactions in cells, free energy change & standard EMF of the cell.		
Unit 2: Organic Chemistry	20%	6
Nomenclature, Introduction to functional groups, chemical &		
physical properties, Oxidation, reduction, elimination, addition		
and substitution reactions, reaction intermediates, Heterocyclic		



compounds, Configuration & projection formula, Optical &		
geometrical isomerism, Tautomerism, Enantiomerism & its		
applications. Aromatic compounds.		
Unit 3: Coordination Chemistry	20%	6
Introduction to co-ordinations compounds, Crystal field theory,		
Color & magnetic properties of complexes, Chelation &		
applications, biologically relevant coordination compounds		
Unit 4: Stereochemistry	20%	6
Conformations with respect to ethane, butane and cyclohexane.		
Interconversion of Wedge Formula, Newmann, Sawhorse and		
Fischer representations. Concept of chirality (upto two carbon		
atoms).		
Unit 5: Mechanism of Organic Reactions	20%	6
Bond fission, Electrophilic and nucleophilic reagents, Types of		
organic reactions, Stability and reactivity of carbon ions, stability		
and reactivity of free radicals		

List Of Practicals	Weightage	Contact hours
Measurement of electrode potential using standard hydrogen		
electrode (SHE).		
Identification of functional groups using chemical tests (e.g.,		
bromine water test for alkenes).		
Tollens' test for aldehydes, etc., on known organic compounds.		
Perform simple distillation, liquid-liquid extraction, and thin-layer		
chromatography (TLC) experiments.		
Redox titrations applications with two examples.		

Instructional Method and Pedagogy: Chalk-Talk, Classroom Discussions, Notes, Use of GeoGebra Toolbox.

Course Outcomes:	Blooms' Taxonomy Domain	Blooms' Taxonomy Sub Domain
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After successful completion of the above course, students will be able to:		
CO1: Recall and define fundamental concepts in electrochemistry, organic chemistry, coordination chemistry, stereochemistry, and organic reaction mechanisms.	Remember	Knowledge
CO2: Demonstrate understanding by explaining the principles underlying electrochemical processes, organic reactions, stereochemical configurations, and coordination chemistry phenomena.	Demonstrate	Define, Classify, Describe,
CO3: Solve problems involving electrode potentials, Nernst equation, redox reactions, organic reaction mechanisms, and coordination compound properties.	Analyze and Solve	Analysis and Application
CO4: Analyze and interpret data from electrochemical experiments, organic reaction mechanisms, stereochemical relationships, and coordination compound properties to draw conclusions and solve complex problems.	Analyze	Analysis and Application
CO5: Synthesize information to propose mechanisms for organic reactions, design experiments related to electrochemistry and coordination chemistry, and develop strategies to solve practical challenges in these fields.	Synthesis	Application and Analysis

Learn	ing Resources						
1	Reference Books:						
	5. A Textbook of Inorganic Chemistry by J.D. Lee.						
	6. Principles of Physical Chemistry by B.R. Puri, L.R. Sharma, and M.S. Pathania.						
	7. Inorganic Chemistry by Gary L. Miessler, Paul J. Fischer, and Donald A. Tarr.						
	8. Chemical Kinetics and Dynamics by Jeffrey I. Steinfeld, Joseph S. Francisco, and						
	William L. Hase						
	5. Principles of Inorganic Chemistry by Puri Sharma Kalia						
2	Journals & Periodicals:						
	2. Chemistry Today						
3	Other Electronic Resources:						
	http://www.chemguide.co.uk/						

Evaluation Scheme	Total Marks			
Theory: Mid semester Marks	20 marks			
Theory: End Semester Marks	40 marks			
Theory: Continuous	Attendance	05 marks		
Evaluation Component Marks	MCQs	10 marks		
T.Turib	Open Book Assignment	15 marks		



	Open Book Assignment	10 marks
	Total	40 Marks
Practical Marks	Attendance	05 marks
	Practical Exam	20 marks
	Viva	10 marks
	Journal	10 marks
	Discipline	05 marks
	Total	50 Marks

Mapping of PSOs & COs

	PSO1	PSO2	PSO3	PSO4	PSO5	PSO6
CO1	3	0	0	0	0	0
CO2	3	1	1	0	0	0
CO3	2	0	2	0	0	0
CO4	0	0	0	0	0	0
CO5	0	0	3	0	0	3

Mapping of POs & Cos

	PO1	PO2	PO3	PO4	PO5	PO6
CO1	0	1	1	0	0	0
CO2	0	1	0	0	0	0
CO3	0	1	0	0	0	0
CO4	1	1	0	0	0	0
CO5	1	1	0	0	0	0



COURSE CODE	COURSE NAME	SEMESTER- II
BSPY216	Physics – I	

	Teaching Scheme (Hours)		Teaching Credit				
Lecture	Practical	Tutorial	Total hours	Lecture	Practical	Tutorial	Total Credit
30	30	00	60	2	1	0	3

Course Pre-requisites	Understanding of basic physics up to school level (10+2 level).
Course Category	Minor (Compulsory)
Course focus	Skill development
Rationale	The fundamental concepts of physics help in understanding the laws of nature and the behaviour of diverse materials under specific conditions. This course is designed to help in understanding the fundamentals of new and emerging technologies that cut across traditional science disciplines, to pursue graduate studies in science.
Course Revision/	07/03/2024
Approval Date:	
Course Objectives	To enable the student to:
(As per Blooms'	1: To understand & remember the basic laws of optical concepts and apply
Taxonomy)	these concepts to understand the working of different optical instruments.
Tuxonomy)	2: To understand & remember the basic quantities governing in the regime
	of electricity and electronics and analyse respective phenomena in allied areas.
	3: To get the knowledge of electromagnetism and their applications .
	4: To understand the fundamental building blocks of matter and remember their behaviour to under different conditions.
	5: Understand the basics of advanced instrumentation.

Weightage	Contact
	hours
20%	6
20%	6
	20%



Unit 3: Magnetostatics	20%	6
Magnetic force, magnetic field and magnetization, Hysteresis. EM Waves, Emission and absorption spectrum, Application of EM Waves.		
Unit 4: Properties of matter	20%	6
Introduction and application of nano-materials, composite materials, Ecofriendly materials.		
Unit 5: Instrumentation Physics Fundamental of optical microscopy, scanning electron microscopy. transmission electron microscopy, X-ray diffractometer, Raman effect and its	20%	6
applications		

List Of Practical	Weightage	Contact hours
Based on theory contents		

Instructional Method and Pedagogy:

Utilizing models, Power point Presentations, films on various topics of physics, group discussions and seminars are some of the methods adopted to improve the student ability to grasp the principles of physics. The hands-on sessions during laboratory sessions will help students to apply the concepts learnt and analyse the results and draw conclusions.

Course Outcome:	Blooms' Taxonomy Domain	Blooms' Taxonomy Sub Domain
After successful completion of the above course, students	will be able to:	
CO1: Describe & demonstrate the fundamental concept related to optics and working principles of various optic instruments.		Define, Classify, Describe, Demonstrate
CO2: Define & classify different electric and electron materials and demonstrate & analyse the working different electronic systems.	Linderstand	Define, Classify, Describe, Demonstrate & Examine
CO3: Define the fundamental notions related electrostatics & magnetostatics and, explain the basics EM waves, their classification and respective applications		Define, Classify, Describe & Demonstrate
CO4: Classify the materials based on their properties ar analyse their behaviour under different conditions.	d Understand, Analyse	Classify, Describe, Demonstrate & Examine
CO5: Describe the fundamental notions and working principles related to the advanced instrumentation techniques and analyse their applications in allied areas.	<u> </u>	Describe & Demonstrate



Learn	ing Resources
1.	Reference Books: 1. Jearl Walker, David Halliday, Robert Resnick, Fundamentals of Physics , Wiley, 2011. 2. D. C. Tayal, Electricity and Magnetism , Himalaya Publishing House, 1988. 3. F. A. Jenkins and H. E White, Fundamentals of Optics, McGraw-Hill Publishing, 4 th edition, 2001. 4. Ch Sateesh Kumar, M. Muralidhar Singh, Ram Krishna, Advanced Materials Characterization, 1 st Edition, CRC press, 2023.
2.	Journals & Periodicals: 1. Journal of Undergraduate Reports in Physics (JURP) 2. Journal of Young Investigators (JYI) 3. Columbia Undergraduate Science Journal (CUSI) 4. Student Journal of Physics (SJP) 5. Indian Journal of Physics (IJP)
3.	Other Electronic Resources: Feynman Lectures in Physics: https://www.feynmanLectures.caltech.edu/

Evaluation Scheme	Total Marks		
Theory: Mid	20 marks		
semester Marks			
Theory: End Semester Marks	40 marks		
Theory: Continuous Evaluation	Attendance	05 marks	
Component Marks			
•	MCQs	10 marks	
	Open Book Assignment	15 marks	
	Article Review	10 marks	
	Total	40 Marks	
Practical Marks			
	Attendance	05 marks	
	Practical Exam	20 marks	
	Viva	10 marks	
	Journal	10 marks	
	Discipline	05 marks	
	Total	50 Marks	

Mapping of PSOs & Cos



	PSO1	PSO2	PSO3	PSO4	PSO5	PSO6
CO1	1	2	1	1	0	1
CO2	1	2	0	1	0	0
CO3	1	2	0	1	0	1
CO4	1	2	0	1	0	0
CO5	1	2	1	1	1	1

Mapping of POs & Cos

	PO1	PO2	PO3	PO4	PO5	PO6
CO1	1	2	0	1	1	1
CO2	1	2	0	1	1	1
CO3	1	2	1	1	1	1
CO4	1	2	1	1	1	1
CO5	2	2	2	1	2	1

1: Slight (low); 2: Moderate (Medium); 3: Substantial (High); 0 None



COURSE CODE	COURSE NAME	SEMESTER
CBCS202	Introduction to Social	II
	Work	

Teaching Scheme (Hours)		Tea	ching Credit				
Lecture	Practical	Tutorial	Total Hours	Lecture	Practical	Tutorial	Total Credit
30	0	30	2	2	0	0	2

Course Pre-requisites	
Course Category	Generic Elective
Course focus	
Rationale	
Course Revision/ Approval Date:	
Course Objectives (As per Blooms' Taxonomy)	 To know the basic values and principles of Social Work To equip candidates with the knowledge of working with people and the ability in problem solving through Field experience To promote among students a sense of commitment and dedication to strive for equity, social justice, social
	harmony and peace 4. To sensitize the students to involve themselves for the cause of poor, subaltern under privileged and disadvantaged section of the society. 5. To develop confidence among the students to feel themselves as change agents for Social change and transformation.

Course Content	\mathcal{E}	Contact hours
Unit 1: An Introduction to Social Work	20%	6
Theory:		
 Social Work: Concept, Meaning, Definition and Objectives Social Work: Nature, Scope and Functions Emergence of Social Work: UK, USA, India 		
4. Development of Social Work Education in India		
Practical: (Give the list of Experiments)		



Unit 2: Values and Ethics in SocialWork Practice	20%	6
 Theory: 1. Assumptions and Values of the Social Work 2. Codes of Ethics 3. Principles of Social Work 4. Social Work and its Relation to Human Rights and Social Justice Practical: (Give the list of Experiments) 		
Unit 3: Social Work Profession in India	200/	
Theory: 1. Profession: Meaning Definition and Attributes 2. Professionalization of Social Work in India 3. Issues and Challenges before Social Work Profession 4. Status of Social Work Profession in India Practical: (Give the list of Experiments)	20%	6
Unit 4: Social Work Philosophy & Methods	20%	6
Theory: 2. 1. Basic values and Ethics, Philosophy and Principles of social work, 3. Types, Functions and Agencies of social control Practical: (Give the list of Experiments)		
Unit 5: Approaches and Ideologies	20%	6
Theory:		
 Professional v/s Voluntary Approaches to Social Work Ideology of Action Groups and Social Movements Generalist Approach to Social Work Practice Influence of Political Ideology on Social Welfare Policies and Programmes Practical: (Give the list of Experiments) 		

Instructional Method and Pedagogy: Presentation, Videos, Chalk-Duster and Notes

	1. Social Work curriculum is designed to provide avariety of Opportunities to develop and enhance skills.
Course Outcomes	
	2. The course work will Aid learning, observation and analysis of
	the realities and experience participation in social intervention.
	2 The leaves is a second to second to being an
	3. The learner is encouraged to move to being an
	independent worker.
	4. Will introduce students to their social responsibilities and
	contribute to the society
	5. will update students on various governmental and non-
	governmental organizations and their workings
Additional Information to	Block Field Work with reputed GOs, NGOs and INGOs
enhance learning	



Learning	Resources
1.	Textbook:
	1. Skidmore, Rex A.(1982), Introduction to Social Work, New Jersey,
	Thackeray, Milton G. Prentice-Hall, Englewood Cliffs.
	2. Singh, D. K. and Bhartiya, A.K. (2010). Social Work: Concept and Methods.
	Lucknow: New Royal Book Company.
	3. Reamer & Fredric (2005) Social Work Values and Ethics, NewDelhi:
	Rawat Publication
2.	Reference books:
	1. Annie Pullen-Sansfaçon (2013), The Ethical Foundations of Social Work,
	Stephen Cowden Routledge, □ Banks, S. (1995).
	2. Ethics and Values in Social Work: Practical Social Work Series, London:
	Macmillan Press Ltd. □ Compton, B. R. (1980).
	3. Introduction to Social Welfare and Social Work. Illinois: The Dorsey
	Press. Desai, Murli, (2006).
	4. Ideologies and social Work: Historical and Contemporary Analyses, Rawat Publication, New Delhi □ Friedlander, Walter A. (1977)
	5. Concepts and Methods of Social Work, New Delhi: Prentice Hall ofIndia
	Pvt. Ltd Heun, Linda R., Heun, Richard E. (2001)
	6. Developing Skills for Human Interaction, London: Charles E. Merrill Co. □
	Jacob, K. K. (Ed.) (1994) Social Work Education in India – Retrospect and Prospect
	Udaipur, Himansu Publications.
	Third Millennium (Some Concerns and Challenges), Sriniketan, Department of
	Social Work, Visva-Bharati. National Association of Social Workers. (2008). Code
	of Ethics of the National Association of Social Workers. Washington, D.C.: NASW
	Press. O'Hagan, Kieran,
	4. Kingsley, Jessica (2003) Competence in Social Work Practice- APractical
2	Guide for Professionals, London
3.	Journal Deviations
<u>4.</u> 5.	Periodicals Other Electronic recoveres
Э.	Other Electronic resources

Evaluation Scheme	Total Marks				
Theory: Mid semester	20 marks				
Marks					
Theory: End Semester	40 marks				
Marks					
Theory: Continuous	Attendance	05 marks			
Evaluation Component	MCQs 10 marks				
Marks	Open Book Assignment 15 marks				
	Article Review	10 marks			
	Total	40 Marks			



COURSE CODE	COURSE NAME	SEMESTER
CBCS203	Web Design	II
	Fundamentals	

Teaching Scheme (Hours)		Teaching Credit					
Lecture	Practical	Tutorial	Total Hours	Lecture	Practical	Tutorial	Total Credit
30	0	30	2	2	0	0	2

Course Pre-requisites	Fundamentals of Internet and Web		
Course Category	Generic Elective		
Course focus			
Rationale			
Course Revision/			
Approval Date:			
Course Objectives (As per Blooms' Taxonomy)	1. To develop the fundamental understanding of Web Design and learning the development of Professional Website 2. To understand the basics of web design using HTML 5 3. To learn the fundamentals of scripting using JavaScript 4. To understand the presentation and design aspects of web page using CSS 5. To learn the website development using Content		
	5. To learn the website development using Content Management System (CMS)		

Course Content	Weightage	Contact			
		Hours			
Unit I: Introduction to Web Development	20%	6			
World Wide Web and Hypermedia, Hypertext Design, Introduction to					
HTML, HTML Elements & Tags, Formatting, Links, Images, Tables,					
Web Forms, Form Elements, Formatting Tags, Images, Multimedia					
Unit II: HTML 5	20%	6			
Semantic Elements, Form Elements, Form Attributes,					
Form Input Types, Media Elements, Canvas, SVG					
Unit III: Java Script	20%	6			
Understanding Java Script, Java Script Variables, String, Array, Objects,					
Functions, Conditional Statements, Looping Statement, Understanding					
&Managing the Events, Working with Document Object Model (DOM)					
Unit IV: Cascading Style Sheets(CSS) Introduction, CSS Box Model,	20%	6			
Managing Borders, Backgrounds, Text Effects,					
Transitions, Animation, Multiple Columns, Inline, Internal and	Fransitions, Animation, Multiple Columns, Inline, Internal and				
External CSS					
Unit V: Developing Website using CMS	20%	6			



Introduction of Content Management System(CMS), CMS with	
WordPress, Google Sites, Magento Fundaments of Webhosting	
Introduction to Bootstrap	

Instructional Method and Pedagogy: Demonstration with Hands-on Practice, Powerpoint Presentation

	After the successful completion of this course, students will be able
	to:
	1. Understand the concepts of Website Development
Course Outcomes	
	2. Design Web Site using different presentation styles
	3. Demonstrate the Dynamic Website Development
	4. Understand the concept of Web Hosting and Management

	Learning Resources
1.	Textbook:
	1. Sams Teach Yourself HTML, CSS and JavaScript, All in One byJulie C.
	Meloni, Pearson Education, Inc.
2.	Reference Book:
	1. HTML 5 Black Book, Dreamtech Press

Evaluation Scheme	Total Marks				
Theory: Mid semester	20 marks				
Marks					
Theory: End Semester	40 marks				
Marks					
Theory: Continuous	Attendance	05 marks			
Evaluation Component	MCQs 10 marks				
Marks	Open Book Assignment 15 marks				
	Article Review	10 marks			
	Total	40 Marks			



COURSE CODE	COURSE NAME	SEMESTER
CBCS204	Introduction to Design	II
	Engineering	

Teaching Scheme (Hours)		Teaching Credit					
Lecture	Practical	Tutorial	Total Hours	Lecture	Practical	Tutorial	Total Credit
30	0	30	2	2	0	0	2

Course Pre-requisites	Fundamentals of Internet and Web
Course Category	Generic Elective
Course focus	
Rationale	
Course Revision/ Approval Date:	
Course Objectives	1. To familiarize the students with Engineering Design process and induce design thinking.
(As per Blooms' Taxonomy)	

Course Content	Weightage	Contact hours
Unit 1: Introduction: Engineering design meaning and significance Definition, considerations and assumptions; role of adesigner, design vocabulary: design levels and design process	10%	6
Unit 2: Problem Definition: Need & Goal Objective tree – illustrative examples; market analysis and information gathering; identification of needs and user requirements; Functions and specifications. Quality Function Deployment (QFD) method – House of Quality – Kano model for customer satisfaction.	30%	8
Unit 3: Conceptual design: generation of alternative concepts Design space – Morphological chart and design thinking; Brainstorming – creativity – developing concepts from functions; Illustrative cases.	30%	8
Unit 4: Concepts evaluation and selection Decision matrix; applying metrics to objectives towards selection of preferred concepts. Numerical evaluation matrices - Priority checkmark method – The best-of-class chart.	30%	8

Instructional Method and Pedagogy: Project Based Learning Peer Based Learning Interactive Sessions



Course Outcomes	After the successful completion of this course, students will be able to:		
Course Outcomes	1. identify problems, gather data, prepare need statements, generate alternative conceptual solutions and choose the best concept		

Learning Resources				
1.	Textbooks:			
	Engineering Design – a project based introduction, Clive L. Dym, Patrick Little and			
	Elizabeth J Orwin, Wiley India edition, 2016.			
2.	Reference Books:			
	Engineering design process, Yousef Haik and Tamer Shahin, Cengage Learning,			
	2011.			

Evaluation Scheme	Total Marks		
Theory: Mid semester	20 marks		
Marks			
Theory: End Semester	40 marks		
Marks			
Theory: Continuous	Attendance	05 marks	
Evaluation Component	MCQs	10 marks	
Marks	Open Book Assignment	15 marks	
	Article Review	10 marks	
	Total	40 Marks	



COURSE CODE	COURSE NAME	SEMESTER
CBCS205	Introduction to	II
	Automotive System	

Teaching Scheme (Hours)		Teaching Credit					
Lecture	Practical	Tutorial	Total Hours	Lecture	Practical	Tutorial	Total Credit
30	0	30	2	2	0	0	2

Course Pre-requisites	Fundamentals of Internet and Web	
Course Category	Generic Elective	
Course focus		
Rationale		
Course Revision/		
Approval Date:		
Course Objectives	1. To give knowledge about fundamentals of Automotive	
(As per Blooms' Taxonomy)	engineering. 2. To enable student to understand main systems and assemblies essential for Automobile. 3. To provide students knowledge about modern vehicle technology. 4. To enable students to understand basic automotive concepts of off road vehicles. 5. To enable students to understand alternative fuels and energy systems.	

	Course Content	Weightage	Contact hours
Unit 1:		20%	06
•	Automotive Engines		
•	Material Science and Technology		
Unit 2:		20%	06
•	Automotive Chassis		
•	Vehicle Body Engineering		
•	Industrial Engineering		
Unit 3:		20%	06
•	Automotive Pollution and Control		
•	Quality Control and Reliability Engineering		
•	Automotive Electrical Systems and Electronics		
Unit 4:		20%	06
•	Operations Research and Industrial Management		
•	Ethics in Engineering and Transport Management		



Instructional Method and Pedagogy: Hands on activities/ Power Point Presentati on/ Chalk/ Duster, Power Point Presentation

Course Outcomes	1. Students will get knowledge about fundamentals of Automotive engineering.
Course outcomes	2. Students will understand main systems and assemblies essential for Automobile.
	3.Students will have basic knowledge about modern vehicle technology.
	4. Students will understand basic automotive concepts of off road vehicles.
	5. Students will understand alternative fuels and energy systems.

Learning	Resources
1.	Textbooks:
	1. Karoly Jarmai and Betti Bollo, Vehicle and Automotive Engineering, Springer
	<u>Publications</u>
2.	Reference Books:
	1. Nikolaos Gkikas, Automotive Ergonomics Driver Vehicle Interaction, CNC Publications
3.	Journals & Periodicals:
	1. Automotive Engineering, SAE International
5.	Other Electronic Resources:
	1. Automotive Engineering, Study.com

Evaluation Scheme	Total Marks	
Theory: Mid semester	20 marks	
Marks		
Theory: End Semester	40 marks	
Marks		
Theory: Continuous	Attendance	05 marks
Evaluation Component	MCQs	10 marks
Marks	Open Book Assignment	15 marks
	Article Review	10 marks
	Total	40 Marks
		,



COURSE CODE	COURSE NAME	SEMESTER
CBCS211	Accounting for Non –	II
	Management Students	

Teaching Scheme (Hours)		Teaching Credit					
Lecture	Practical	Tutorial	Total Hours	Lecture	Practical	Tutorial	Total Credit
30	0	30	2	2	0	0	2

Course Pre-requisites	Fundamentals of Internet and Web
Course Category	Generic Elective
Course focus	
Rationale	
Course Revision/	
Approval Date:	
Course Objectives	To demystify debits and credits
(A DI	2. To read a financial statement
(As per Blooms'	3. To understand corporate transactions
Taxonomy)	4. To put together an income statement

	Course Content	Weightage	Contact hours			
Unit	1: Introducing Accounting and financial statements	20%	6			
•	Generally Accepted Accounting Principles					
Unit	2: Financial Statements	20%	6			
•	Balance Sheet and its Components					
•	The Income Statement					
Unit	3 Double-Entry Accounting	20%	6			
•	The General Journal					
•	The General Ledger					
•	Adjusting Journal Entries					
•	Closing Journal Entries					
Unit	nit 4: Forms of Business Organisations 20% 6					
•	Sole Proprietorship					
•	Partnership					
•	Corporation					
Unit	5: Financial Statement Analysis	20%	6			
•	Ratio Analysis					
•	Trend Analysis					
•	Comparative Statement Analysis					

Instructional Method and Pedagogy: Hands on activities/ Power Point Presentati on/ Chalk/ Duster

1. The candidate will gain the ability to read, interpret
and analyse financial statements.



	2. Analyse financial statements to assess financial performance and position of a company.		
	performance and position of a company.		
Additional Information to	Any site visit required or expert talk required on specific		
enhance learning	topics.		

Learni	ng Resources
1.	Textbook: Accounting for Non-Accountants, Dr. Wayne A Label
2.	Reference books: The Essentials of Finance and Accounting for Nonfinancial Managers
	by Edward Fields
	Finance for Non-Financial Managers by Gene Sicillinao
3.	Journal & Periodicals:
	Journal of Accounting, Auditing & Finance: SAGE Journals
	Accounting & Finance: Wiley Online Library
	Journal of Finance and Accounting: Science Publishing Group

Evaluation Scheme	Total Marks				
Theory: Mid semester	20 marks				
Marks					
Theory: End Semester	40 marks				
Marks					
Theory: Continuous	Attendance	05 marks			
Evaluation Component	MCQs	10 marks			
Marks	Open Book Assignment	15 marks			
	Article Review	10 marks			
	Total	40 Marks			
	L				



COURSE CODE	COURSE NAME	SEMESTER
CBCS212	Personal Finance	II

Teaching Scheme (Hours)			Tea	ching Credit			
Lecture Practical Tutorial Total Hours				Lecture	Practical	Tutorial	Total Credit
30	0	30	2	2	0	0	2

Course Pre-requisites	Domain Knowledge of Income and Saving
Course Category	Generic Elective
Course focus	
Rationale	
Course Revision/	
Approval Date:	
Course Objectives	1. To equip the students basic knowledge of finance theory
(A DI	2. To understand the important of financial status
(As per Blooms'	3. To know the allocation of funds in different financial instruments
Taxonomy)	4. To implement and monitor the financial plan.

Course Content	Weightage	Contact
		hours
Unit 1: Introduction	20%	5 Hours
1. Planning Personal Finances:		
2. Personal Finance Basics		
3. Time Value Of Money		
Unit 2 Managing Funds	15%	5 Hours
1. Managing Personal Finances:		
2. Financial Services: Saving Plans And		
Payment Accounts		
3. Introduction To Consumer Credit, Choosing		
A Source Of Credit		
Unit 3: Insuring resources:	15%	5 Hours
1. Property And Motor Vehicle Insurance,		
2. Health, Disability And Long Term Care		
Insurance And		
3. Life Insurance		
Unit 4: Investing Financial Resources	30%	8 Hours
1. Investing Fundamentals,		
2. Investing in stocks, Bonds, Mutual funds, Real estate		
Unit 5 Practical: Prepare a Financial Plan for a Person / Manager /	20%	7 Hours
Entrepreneur / Director		



Instructional Method and Pedagogy: Case analysis, Power Point Presentati on/ Chalk/ Duster

	1. Familiarize with income and importance of financial goals				
	2. Know the importance of retirement planning				
Course Outcomes	3. Understand methodology of revision of financial goals				

Learning 1	Resources					
1.	Textbooks: Personal Finance, JackR. Kapoor, Les R. Dlabay, Robert J.					
	Hughes, McGraw- Hill HigherEducation					
2.	Reference Books:					
	1. V.K.Singhania, Taxman's direct taxes planning and management,					
	Taxmann					
	2. Corporate Tax Planning, McGraw- Hill Higher Education					
3.	News Paper					
	1. Economic Times					
	2. Business Standard					
4	News Channel CNBC					
	NDTV Profit					

Evaluation Scheme	Total Marks				
Theory: Mid semester	20 marks				
Marks					
Theory: End Semester	40 marks				
Marks					
Theory: Continuous	Attendance	05 marks			
Evaluation Component	MCQs	10 marks			
Marks	Open Book Assignment	15 marks			
	Article Review	10 marks			
	Total	40 Marks			



COURSE CODE	COURSE NAME	SEMESTER II
AECC201	Communication Skills in English	

Teaching Scheme (Hours)			Teaching Credit				
Lecture	Practical	Tutorial	Total Hours	Lecture	Practical	Tutorial	Total Credit
30	00	00	30	30	00	00	2

Course Pre-	Student should have cleared First Semester of Bachelor of Science					
requisites						
Course Category	Mandatory Course					
Course focus	Communicational Skills					
Rationale	It enables humanity to experience the benefits of chemistry when we					
	apply it in the exploitation of materials and energy.					
Course Revision/	14/03/2023					
Approval Date:						
Course Objectives	1. To enable learners, develop their basic communication skills in					
(A DI 1	English.					
(As per Blooms'	2. To equip them with writing skills needed for academic as well as					
Taxonomy)	workplace context.					
	3. To prepare students for professional communication at world level.					
	4. To develop corporate communicational attitude.					
	5. To strengthen digital communication using technological modules					
	and expertise.					

Course Content (Theory)	Weightage	Contact
		hours
Unit 1: Communicative Skills		
Basics of Communication, Verbal & Non-verbal, Communication,	20%	6
Barriers to Effective Communication, Strategies of Effective		
Communication		
Unit 2: Grammar & Vocabulary:		
Types of sentences, Synonyms, Antonyms, Tenses - Past, Present &		
Future, Homophones, Modals, Verb forms, Phrasal Verbs, Error	15%	5
correction, commonly misused words, technical		
terms		
Unit 4: Writing Skills & Speaking Skills: Letter writing -	25%	7
Complaint & Leave, Article, Precise writing, Report writing, Note-		
taking and Note-making, Creative Writing Introducing self, Interview		
Skills, Public Speaking, Debates, Role plays, Group Discussion.		
Unit 3: Listening & Reading Skills:		
Definitions (Listening & Reading), Types of Listening, Barriers to		
Effective Listening, Traits of a Good Listener, Types of Reading,	30%	9
Techniques of Effective Reading, Reading Tasks (Critical &	2070	



Inferential)		
Unit 4: Writing Skills & Speaking Skills:		
Letter writing - Complaint & Leave, Article, Precise writing, Report	25%	7
writing, Note-taking and Note-making, Creative Writing Introducing		
self, Interview Skills, Public Speaking, Debates, Role plays, Group		
Discussion.		
Unit 5: ICT/ Digital/ E-Skills: Computer Assisted Language		
Learning (CALL), Mobile Assisted Language Learning (MALL),		
Emails, Blogs, Digital/ E-Portfolio, Filling Online Application Forms	20%	6

Course Objectives	Blooms' Taxonomy Domain	Blooms' Taxonomy Sub Domain
After successful completion of the above course, stude	nts will be able to:	
CO1: To emphasize the development of listening	Understand,	Define, Classify &
and reading skills among learners	Analyse,	Demonstrate
	Remember	
CO2: To equip them with writing skills needed for	Analyse, Apply,	Classify, Describe &
academic as well as workplace context	Understand	Demonstrate
CO3: To enable learners of Engineering and	Understand,	Define, Describe &
Technology develop their basic	remember	Demonstrate
communication skills in English		
CO4: To strengthen the fundamentals in English	Remember,	Define Describe
Language.	Analyse	
CO5: To build up the confidence to communicate	Understand,	Define, Classify,
with the world.	Apply	Describe &
		Demonstrate

Lear	ming Resources
1.	Textbook: An Introduction to Professional English and Soft Skills by B K Das
2.	Reference Books:
	1. Murphy, Raymond.(1998), Intermediate English Grammar, New York
	2. Wren & Martin (2001), English Grammar & Composition, New York
	3. Mudambadithaya G.S.,(2002) English Grammar and composition
	4. Digne, Flinders and Sweeney(2010) Cambridge University press
	5. Lupton, Mary Jane (1998). Maya Angelou: A Critical Companion. Westport,
	: Greenwood Press. ISBN 978-0-313-303225.
	6. Booher, Diana. (2004), Booher's Rules of Business Grammar, OUPUr, Penny
	(2002), Grammar Practice Activities, OUP



Evaluation Scheme	Total Marks				
Theory: Midsemester	20 marks				
Marks					
Theory: End Semester	40 marks				
Marks					
Theory: Continuous	Attendance 05 marks				
Evaluation Component Marks	MCQs	10 marks			
	Skill enhancement activities / 15 marks case study				
	Presentation/ miscellaneous activities	10 marks			
	Total	40 Marks			

	PSO1	PSO2	PSO3	PSO4	PSO5	PSO6
CO1	1	1	1	0	3	3
CO2	1	1	1	0	3	3
CO3	1	1	1	0	3	2
CO4	1	1	1	0	3	3
CO5	1	1	1	0	3	3

Mapping of POs & Cos

	PO1	PO2	PO3	PO4	PO5	PO6
CO1	3	2	2	3	2	3
CO2	3	3	3	3	2	3
CO3	3	2	3	3	2	2
CO4	3	1	3	3	3	3
CO5	3	2	2	3	2	2



COURSE CODE	COURSE NAME	SEMESTER
VACC201	Vedic Mathematics	II

Teaching Scheme (Hours)				Teaching	g Credit		
Lecture	Practical	Tutorial	Total Hours	Lecture Practical Tutorial Cr			
30	0	0	30	2	0	0	2

Course Pre-requisites	None			
Course Category	Value Added Elective			
Course focus	Skill development			
Rationale	The present course will give the idea about different concepts			
	of Vedic Mathematics like Arithmetic, Algebra, Geometry and			
	Trigonometry.			
Course Revision/				
Approval Date:				
Course Objectives	To enable the student to:			
(As per Blooms'	1: Understand Concepts of Vedic Mathematics to promote			
Taxonomy)	joyful learning of mathematics,.			
	2: Learn Vedic Mathematics to enhance computation skills.			
	3:. Develop logical and analytical thinking			
	4: Apply Vedic mathematics to solve problems of Algebra,			
	geometry and Trigonometry.			
	5: Understand the rich heritage of mathematical temper of			
	Ancient India			



Course Content (Theory)	Weightage	Contact hours
Unit 1: History of Vedic Maths and its Features, Vedic Maths formulae: Sutras and Upsutras Addition in Vedic Maths: Without carrying, Dot Method 77, Subtraction in Vedic Maths: Nikhilam Navatashcaramam Dashatah (All from 9 last from 10), Fraction -Addition and Subtraction	20%	6
Unit 2: "Miracle Multiplication and Excellent Division, Multiplication in Vedic Maths: Base Method (any two numbers up to three digits, Multiplication by Urdhva Tiryak Sutra, Miracle multiplication: Any three-digit number by series of I's and 9's, Division by Urdhva Tiryak Sutra (Vinculum method)"	20%	6
Unit 3: Lightening Squares and Rapid Cubes, Squares of any two-digit numbers: Base method, Square of numbers ending in 5: Ekadhikena Purvena Sutra, Easy square roots: Dwandwa Yoga (duplex) Sutra, Square root of 2: Baudhayana Shulbasutra, Cubing: Yavadunam Sutra	20%	6
Unit 4: Enlighten Algebra and Geometry, Factoring Quadratic equation: Anurupyena, Adyamadyenantyamanty Sutra, Concept of Baudhayana (Pythagoras) Theorem,	20%	6
Unit 5: Circling a square: Baudhayana Shulbasutra, Concept of pi: Baudhayana Shulbasutra, Concept angle (8) 0o, 300, 450, 600 and 900: Baudhayana number	20%	6

 $\label{lem:conditional} \textbf{Instructional Method and Pedagogy:} \ \textbf{Chalk \& board, group discussions, assignments, }, \\ \textbf{Online materials, Practice examples}$

Course Outcomes:	Blooms' Taxonomy Domain	Blooms' Taxonomy Sub Domain
After successful completion of the above course, studen	ts will be able to:	
CO1: Apply techniques of Vedic Mathematics to	Remember,	Explain, Use,
solve the problems of Addition and subtraction.	Understand,	Solve,
	Apply	
CO2: Understand and Apply techniques of Vedic	Remember,	, Identify, Use,
Mathematics to solve the problems of Multiplication	Understand,	Solve
and Division.	Apply	
CO3: Apply techniques of Vedic Mathematics to find	Understand,	Describe,
squares, cubes, square roots and Cube roots.	Apply	Identify, Solve,
		Use, Find
CO4: Understand and Apply techniques of Vedic	Remember,	Describe, Use,
Mathematics to solve quadratic equation.	Apply	Solve
CO5: Understand the rich heritage of mathematical	Understand	classify,
temper of Ancient Indi.		Explain,
		Identify, Use,
		solve.



Learning R	Resources
1.	Reference Books: 1. A Modern Introduction to Ancient Indian Mathematics, T S Bhanumurthy, Wiley Eastern Limited, New Delhi 2. Enjoy Vedic Mathematics, S M Chauthaiwale, R Kollaru, The Art of Living, Bangalore 3. Magical World of Mathematics, VG Unkalkar, Vandana Publishers, Banglore.
	Text Books:
	 The Essential of Vedic Mathematics, Rajesh Kumar Thakur, Rupa Publications, New Delhi, 2019 Vedic Mathematic by Bharati Krishna Tirthaji, Motilal Banarasidas,
	New Delhi, 2015.
	3. Vedica Mathematics made easy, Dhaval Bhatiya, Jaico Publishing, New Delhi, 2011
2.	Journals & Periodicals:
3.	Other Electronic Resources:

Evaluation Scheme	Total Marks	
Theory: Mid semester Marks	20 marks	
Theory: End Semester Marks	40 marks	
Theory: Continuous		
Evaluation Component	Attendance	5
Marks	MCQs	10
	Open Book Assignment	15
	Article Review / Presentations / Practice Assignments	10
	Total	40
	L	



COURSE CODE	COURSE NAME	SEMESTER
VACC202	Health and Physical	II
	Education	

Teaching Scheme (Hours)				Teaching Credit			
Lecture	Practical	Tutorial	Total Hours	Lecture	Practical	Tutorial	Total Credit
15	30	0	45	1	2	0	2

Course Pre-requisites	None
Course Category	Value Added Elective
Course focus	Promoting physical, mental, and emotional well-being through various activities and educational content
Rationale	Health and Physical Education (HPE) as a subject provides students with a comprehensive understanding of various aspects related to physical well-being, mental health, and the importance of leading a healthy lifestyle.
Course Revision/ Approval Date:	
Course Objectives (As per Blooms' Taxonomy)	To enable the student to: 1. Remembering: Recall facts, terminology, and basic principles related to physical fitness, nutrition, safety guidelines, and health promotion. 2. Understanding: Explain the significance of physical activity for overall health and wellbeing. Interpret health-related information, such as nutrition labels or exercise prescriptions. 3. Applying: Demonstrate proper techniques and skills in various physical activities, sports, and exercises. Apply safety guidelines and procedures in physical activity settings. Utilize knowledge of nutrition to plan balanced meals and snacks. 4. Analyzing: Evaluate personal fitness levels and set goals for improvement. 5. Evaluating: Assess the validity of fitness assessments and measurements.
	☐ Evaluate personal health behaviors and make adjustments to improve overall wellness.
	6. Creating: □ Design personalized fitness plans or workout routines based





on individual goals and preferences. ☐ Develop educational materials or presentations to promote health and
fitness in the community.
Create innovative strategies to overcome barriers to physical activity
participation or healthy eating.



Course Content (Theory)	Weightage	Contact hours
Unit 1: Physical Fitness, Wellness and Lifestyle	20%	3
- Meaning and Importance of physical fitness, wellness and lifestyle		
- Components of physical fitness, wellness and lifestyle		
Unit 2: Sports Nutrition	20%	3
- Concept of Balanced Diet and Nutrition		
- Macro and Micro Nutrients: Food sources and functions		
- Nutritive and Non- Nutritive Components of Diet		
Unit 3: Sports Training	20%	3
- Meaning and Concept of Sports Training		
- Principles of Sports Training		
- Warming up and limbering down		
- Skill, Technique and Style		
Unit 4: Management of Sports Injuries	20%	3
- Sports injuries: Classification (Soft Tissue Injuries – Abrasion,		
Contusion, Laceration, Incision, Sprain & Strain;		
- Bone & Joint Injuries - Dislocation, Fractures - Green Stick,		
Comminuted, Transverse, Oblique & Impacted)		
Unit 5: Test and Measurement in Sports	20%	3
- Define Test, Measurement and Evaluation		
- Importance of Test, Measurement and Evaluation in Sports		
- Calculation of BMI		

List Of Practical Tutorial	Weightage	Contact
		hours
Unit 1: Development of Strength :Lever- I	20%	5
Unit 2: Development of Speed: Level - I	20%	5
Unit 3: Development of Endurance: Level - I	20%	5
Unit 4 Development of Flexibility: Level - I	20%	5
Unit 5: Development of Coordination: Level - I	20%	5

Instructional Method and Pedagogy: Chalk & board, group discussions, assignments, , Online materials, Practice examples

Course Outcomes:	Blooms' Taxonomy Domain	Blooms' Taxonomy Sub Domain
After successful completion of the above course, studen	ts will be able to:	
CO1: Understand the significance of physical activity	Remember,	Define, State,
for overall health and well-being.	Understand,	classify,
Apply safety guidelines and procedures in physical	Apply	Explain, Use,
activity settings.		Examine
Create innovative strategies to overcome barriers to		
physical activity participation or healthy eating.		
CO2: Create nutrition to plan balanced meals and	Remember,	Define, State,
snacks.	Understand,	classify,
Understand the significance of balanced diet for	Apply	Explain,
overall health and well-being.		Identify, Use
CO3: Apply the proper techniques and skills in	Understand,	Describe,
various physical activities, sports, and exercises.	Apply	Identify, Solve,



Apply safety guidelines and procedures in physical		Use, Find
activity settings.		
CO4: Recall facts, terminology, and basic principles	Remember,	Describe,
related to physical fitness, nutrition, safety guidelines,	Apply	Identify, Use,
and health promotion.		Solve, Find
CO5: Assess the validity of fitness assessments and	Understand,	classify,
measurements.	Apply	Explain,
Evaluate personal health behaviors and make		Identify,
adjustments to improve overall wellness.		Use, Solve,
		Examine

Learning I	Resources
1.	Reference Books:
	1. Physical education and sports science -Umesh kr. Ahlawat
	Text Books:
	1. Health and Physical Education – National Council of Educational
	Research and Training
	2. Saraswati's Health & Physical Education -V.K. Sharma
	3. Physical Education: Central Board of Secondary Education
2.	Journals & Periodicals:
3.	Other Electronic Resources:

Evaluation Scheme	Total Marks							
Theory: Mid semester Marks	20 marks							
Theory: End Semester Marks	40 marks							
Theory: Continuous Evaluation Component Marks	Attendance	5						
IVI CII IX	MCQs	10						
	Open Book Assignment	15						
	Article Review / Presentations / Practice Assignments	10						
	Total	40						
Practical Marks								
	Attendance	5						
	Practical Exam	35						
	Viva	05						
	Journal	05						
	Total	50						



		Semester – III						
Sr. No.	Course Code	Course Title	L	T	P	C	Marks	
	A. Major							
1	BSMO311	Microbial Genetics	3	0	1	4	150	
2	BSMO312	Introduction To Microbiology And Microbial Diversity	3	0	1	4	150	
3	BSMO313	Bacteriology	3	0	1	4	150	
		B. i. Minor (Compulsory)						
4	BSCM316	Chemistry - I	2	0	1	3	75	
		B. ii. Minor (Elective)						
5	BSMA316	Mathematics - II	2	1	0	3		
6	BSPY316	Physics - II	2	0	1	3	75	
C. M	Iultidisciplinary	Additional Credits under Choice Ba	sed C	redit	Syste	m (Aı	ny One)	
7	CBCS302	World Geography						
8	CBCS303	App Development				2		
9	CBCS305	Computer Aided Design	2	0	0		100	
10	CBCS312	Project Management	-					
11	CBCS210	Research Methodology	-					
		D. Ability Enhancement Cour	se		<u> </u>			
8	AECC301	Entrepreneurship Development	2	0	0	2	50	
		E. Skill Enhancement Cours	e					
9	SECC301	Internship	0	0	2	2	50	
		F. Value Added Course	•					
10	VACC201	NCC/Sports	0	0	2	2	50	
				1	otal	22	700	



Teaching Scheme Semester – III

G.	C			achin (Hour	_		Т	each	ing C	Credit			Evaluat	ion Schen	me	
Sr. No.	Course Code	Course Name	L	P	Т	Total	L	P	Т	Total	Theory: MS Marks	Theory: CEC Marks	Theory: ES Marks	Theory Marks	Practical Marks	Total Marks
	A. Major Courses															
1.	BSMO311	Microbial Genetics	3	2	0	5	3	1	0	4	20	40	40	100	50	150
2.	BSMO312	Introduction To Microbiology And Microbial Diversity	3	2	0	5	3	1	0	4	20	40	40	100	50	150
3	BSMO313	Bacteriology	3	2	0	5	3	1	0	4	20	40	40	100	50	150
	B. Minor Courses															
3.	BSCM316	Chemistry – I	2	2	0	4	2	1	0	3	20	40	40	100	50	75
4.	BSMA316	Mathematcs - II	2	0	1	3	2	0	1	3	20	45	75	150	0	75
5.	BSPY316	Physics - II	2	2	0	4	2	1	0	3	20	40	40	100	50	75
							C. 1	Multio	liscip	linary Co	ourses (Any	one)				
4.	CBCS302	World Geography														
5.	CBCS303	App Development														
6.	CBCS305	Computer Aided Design	2	0	0	2	2	0	0	2	20	40	40	100	00	100
7.	CBCS312	Project Management														
8.	CBCS210	Research Methodology														



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	D. Ability Enhancement Courses															
10.	AECC301	Entrepreneurship Development	2	0	0	2	2	0	0	2	20	40	40	100	00	100
	E. Skill Enhancement Courses															
10.	SECC301	Industrial Internship	0	2	0	2	0	2	0	2	00	00	00	50	00	50
								F	. Valı	ie Addeo	l Courses					
11.	VACC301	NCC/Sports	0	2	0	0	0	2	0	2	00	00	00	50	00	50
	Total									22						700



COURSE CODE BSMO311 COURSE NAME MICROBIAL GENETICS SEMESTER III

	Teacl	ning Scheme	(Hours)			ching edit	
Lecture	Practical	Tutorial	Total Hours	Lecture	Practical	Tutorial	Total Credit
45	30	0	75	3	1	0	4

Course Pre-requisites	Fundamental knowledge of concepts related to genes, genomes and						
	chromosomes.						
Course Category	Professional Core Course						
Course focus	Employability						
Rationale	To have an overview of genome organisation of bacteria. The subject also						
	explains mechanisms of genetic exchange in bacteria and transposable						
	elements present in prokaryotes.						
Course Revision/Approval	07/03/2024						
Date:							
Course Objectives (As per Blooms' Taxonomy)	1. Remember Understanding microbial genes, genomes, and gene expression.						
	2. Analyze and Differentiate Types of plasmids						
	3. Understand Microbial replication, transcription and translation.						
	4. Illustrate and Discuss Phage Genetics						
	5. Evaluate and Apply Knowledge of Transposable Elements						

Course Content (Theory)	Weightage	Contact
		hours
Unit 1: Genome Organization	20%	6
E. coli, Saccharomyces, Tetrahymena Mutations and mutagenesis: Definition and		
types of Mutations; Physical and chemical mutagens; Molecular basis of		
mutations; Functional mutants (loss and gain of function mutants); Uses of		
mutations. Reversion and suppression: True revertant; Intra- and inter-genic		
suppression; Ames test; Mutator Genes		
Unit 2: Plasmids	20%	6
Types of plasmids – F plasmid, R Plasmids, colicin genic plasmids, Ti plasmids,		
linear plasmids, yeast- 2 µ plasmid, Plasmid replication and partitioning, Host		
range, plasmidincompatibility, plasmid amplification, Regulation of copy		
number, curing of plasmids.		
Unit 3: Mechanisms of Genetic Exchange	20%	6
No. of Transformation - Discovery, mechanism of natural competence.		
Conjugation - Discovery, mechanism, Hfr and F' strains, Interrupted mating		
technique and time of entry mapping. Transduction - Generalized transduction,		
specialized transduction, LFT & HFT lysates, Mapping by recombination and co-		
transduction of markers.		





Unit 4: Phage Genetics Features of T4 genetics, Genetic basis of lytic versus lysogenic switch of phage lambda	20%	6
Unit 5: Transposable elements Prokaryotic transposable elements – Insertion Sequences, composite and non- composite transposons, Replicative and Non replicative transposition, Mu transposon. Eukaryotic transposable elements - Yeast (Ty retrotransposon), Drosophila (P elements), Maize (Ac/Ds) Uses of transposons and transposition.	20%	6

List Of Practical	Weightage	Contact hours
1: Study the effect of chemical (HNO2) and physical (UV) mutagens on bacterial cells	10%	4
2: Study survival curve of bacteria after exposure to ultraviolet (UV) light	10%	4
3. Isolation of chromosomal DNA from E. Coli	10%	4
4. Agarose gel electrophoresis of genomic DNA and purity by spectrophotometry	10%	4
5. Isolation of Plasmid DNA from E. coli.	10%	4
6. Study different conformations of plasmid DNA through Agarose gel electrophoresis.	10%	4
7. Demonstration of Bacterial Conjugation	10%	2
8. Demonstration of Bacterial Transformation	10%	2
9. Demonstration of Bacterial Transduction	10%	4
10. Demonstration of AMES test	10%	4

Instructional Methodand Pedagogy: (Max. 100 words)

PPT, Demonstration, Video, Case study



Afters	Course Outcomes: uccessful completion of the above course, students will be able:	Blooms' Taxonomy Domain	Blooms' Taxonomy Sub Domain
CO1	To understanding Genome Structure and Mutation Mechanisms	Remember, understand	Explain, Describe, Discuss, Recall,
CO2	To gain knowledge of Plasmid Types and Functionslearn the processes behind mutations and other genetic changes	knowledge	Apply, Practice, Interpret, Select, Correlate
CO3	To understand and apply the mechanisms of Genetic Exchange and Mapping Techniques	Understand, apply, Analyses and Evaluation	Compare, Classify, Select, Investigate
CO4	To learn genetic Basis of Phage Lifecycle Decisions	Remember, understand	Construct, Develop, Produce
CO5	To gain insights into prokaryotic and eukaryotic transposable elements	Understand	Explain, Describe

T	
Learning Res	
1.	Reference books:
	1. Watson JD, Baker TA, Bell SP et al. (2008) Molecular Biology of the Gene, 6th Ed.,
	Benjamin Cummings.
	2. Gardner EJ, Simmons MJ, Snustad DP (2008). Principles of Genetics. 8th Ed. Wiley-India.
	3. Russell PJ. (2009). i Genetics- A Molecular Approach. 3rd Ed, Benjamin Cummings.
	4. Sambrook J and Russell DW. (2001). Molecular Cloning: A Laboratory Manual. 4th Edition, Cold Spring Harbour Laboratory press.
	Maloy SR, Cronan JE and Friefelder D (2004) Microbial Genetics 2nd EDITION., Jones and Barlett Publishers.
2.	Journals & Periodicals
	1. Microbial Genomics
	2. Genes
	3. Genomics
	4. The Scientist
3.	Other Electronic resources: https://www.the-cientist.com/tag/microbiology,genetics-
	genomics

Evaluation Scheme	Total Marks		
Theory: Midsemester Marks	20 marks		
Theory: End Semester Marks	40 marks		
Theory: Continuous Evaluation Component	Attendance	05 marks	



Marks	MCQs	10 marks
	Open Book Assignment	15 marks
	Research Paper Review	10 marks
	Total	40 Marks
Practical Marks		
	Attendance	05 marks
	Practical Exam	30 marks
	Viva	10 marks
	Journal	05 marks
	Total	50 Marks

	PSO1	PSO2	PSO3	PSO4	PSO5	PSO6
CO1	2	2	1	1	1	0
CO2	2	2	1	1	1	0
CO3	2	1	2	0	1	2
CO4	2	1	2	0	1	2
CO5	2	0	0	1	0	0

Mapping of POs & COs

	PO1	PO2	PO3	PO4	PO5	PO6
CO1	3	2	1	1	1	0
CO2	3	2	1	1	1	0
CO3	2	2	3	0	1	2
CO4	2	2	3	0	1	2
CO5	2	ı	1	1	0	1



COURSE CODE COURSE NAME SEMESTER
BSMO312 INTRODUCTION TO MICROBIOLOGY AND MICROBIAL DIVERSITY

Teaching Scheme (Hours) Teaching Credit							
Lecture	Practical	Tutori al	Total Hours	Lecture	Practic al	Tutori al	Total Credi t
45	30	0	75	3	1	0	4

Course Pre-requisites	Fundamental knowledge of concepts related to microbiology
Course Category	Major
Course focus	Employability
Rationale	To have an overview and understanding of microbiology and microbial diversity. The subject also explains the life cycle and industrial importance of microorganisms.
Course Revision/Approval Date:	07/03/2024
Course Objectives (As per Blooms' Taxonomy)	 Remember the History of Microbiology Analyze different types of microorganisms Understand about algae and their life cycle Apply the importance of fungi. Explain the characteristics of protozoa

Course Content (Theory)	Weightage	Contact
		hours
Unit 1: History of Development of Microbiology	20%	6
Development of microbiology as a discipline, Spontaneous generation vs.		
biogenesis. Noteworthy Contributions of Scientists. Role of microorganisms in		
fermentation, Germ theory of disease, Development of various microbiological		
techniques and golden era of microbiology, Development of the field of soil		
microbiology and important contributions. Establishment of fields of medical		
microbiology and immunology through notable contributions.		
Unit 2 Diversity of Microbial World	20%	6
A. Systems of classification:		
Binomial Nomenclature, Whittaker's five kingdom and Carl Woese's three		
kingdom classification systems and their utility. Difference between prokaryotic		
and eukaryotic microorganisms		
B. General characteristics of different groups:		
Acellular microorganisms (Viruses, Viroids, Prions) and Cellular		
microorganisms (Bacteria, Algae, Fungi and Protozoa) with emphasis on		
distribution and occurrence, morphology, mode of reproduction and economic		
importance.		



Unit 3 Algae	20%	6
History of phycology with emphasis on contributions of Indian scientists;		
General characteristics of algae including occurrence, thallus organization, algae		
cell ultra structure, pigments, flagella, eyespot food reserves and vegetative,		
asexual and sexual reproduction. Different types of life cycles in algae with		
suitable examples: Haplobiontic, Haplontic, Diplontic, Diplobiontic and		
Diplohaplontic life cycles. Applications of algae in agriculture, industry,		
environment and food.		
Unit 4: Fungi	20%	6
Historical developments in the field of Mycology including significant		
contributions of eminent mycologists. General characteristics of fungi including		
habitat, distribution, nutritional requirements, fungal cell ultra- structure, thallus		
organization and aggregation, fungal wall structure and synthesis, asexual		
reproduction, sexual reproduction, heterokaryosis, heterothallism and parasexual		
mechanism. Economic importance of fungi with examples in agriculture,		
environment, Industry, medicine, food, biodeterioration and mycotoxins.		
Unit 5: Protozoa	20%	6
General characteristics with special reference to Amoeba, Paramecium,		
Plasmodium, Leishmania and Giardia An overview of Scope of Microbiology		

List Of Practical	Weightage	Conta ct hours
1: Preparation of various media, sterilization and testing for sterility	10%	2
2: Study of Air microflora	10%	4
3: Preparation of Winogredsky's column and study of different groups	12%	4
4. Study of Spirogyra and Chlamydomonas, Volvox using temporary Mounts	12%	4
5. Growth on Sabouraud's agar / PDA and wet mount of fungus	10%	4
6. Study of Rhizopus, Penicillium, Aspergillus using temporary mounts	12%	4
7. Study of the following protozoans using permanent mounts/photographs: Amoeba, Entamoeba, Paramecium and Plasmodium	36%	12

Instructional Method and Pedagogy:

Audio-Visual Lectures, Quizzes, PPT, Demonstration, Video, Case studies

Course Objectives: After successful completion of the above course, students will be able to:		Blooms' Taxonomy Domain	Blooms' Taxonomy Sub Domain
CO1	Describe Key Historical Developments in Microbiology	Remember	Explain, Describe, Discuss, Recall, Locate





Academ	iic Tear, 2024-23		
CO2	Classify and Differentiate Microorganisms	Apply	Apply, Practice, Interpret, Select, Correlate
CO3	Analyze and Explain the Characteristics and Applications of Algae	Analyses and Evaluation	Compare, Classify, Select, Investigate
CO4	Evaluate the Importance and Characteristics of Fungi	Evaluate	Construct, Develop, Produce
CO5	Describe and Interpret the General Characteristics of Protozoa	Understand	Explain, Describe, outline, Predict, Summarize

Learning Res	sources					
1.	Reference Books:					
	1. Atlas RM. (1997). Principles of Microbiology. 2nd edition. WM.T.Brown Publishers.					
	2. Pelczar MJ, Chan ECS and Krieg NR. (1993). Microbiology. 5th edition. McGraw Hill Book Company.					
	3. Stanier RY, Ingraham JL, Wheelis ML, and Painter PR. (2005). General Microbiology. 5th edition. McMillan.					
2.	Journals & Periodicals					
	1. Microbiological Research					
	2. FEMS Microbiology Ecology					
	3. Journal of Microbiology					
	4. Microbiology Today					
3.	Other Electronic resources: https://microbiologysociety.org/					

Evaluation Scheme	Total Marks		
Theory: Midsemester Marks	20 marks		
Theory: End Semester Marks	40 marks		
Theory: Continuous Evaluation Component Marks	Attendance	05 marks	
	MCQs	10 marks	



	Open Book Assignment	15 marks
	Research Paper Review	10 marks
	Total	40 Marks
Practical Marks		
	Attendance	05 marks
	Practical Exam	30 marks
	Viva	10 marks
	Journal	05 marks
	Total	50 Marks

	PSO1	PSO2	PSO3	PSO4	PSO5	PSO6
CO1	2	2	1	1	2	2
CO2	2	2	1	1	2	2
CO3	2	2	2	1	2	2
CO4	2	2	2	1	2	2
CO5	2	0	0	0	0	1

Mapping of POs & COs

<u> </u>						
	PO1	PO2	PO3	PO4	PO5	PO6
CO1	2	1	2	1	2	1
CO2	2	1	2	1	2	1
CO3	2	1	2	1	2	1
CO4	2	1	2	1	2	1
CO5	2	2	0	0	0	0



COURSE CODE BSMO313 COURSE NAME BACTERIOLOGY SEMESTER III

Teaching Scheme (Hours)						ching redit	
Lecture	Practical	Tutorial	Total Hours	Lecture Practical Tutorial			Total Credit
45	30	0	75	3	1	0	4

Course Pre-requisites	Fundamental concepts of microbiology.
Course Category	Major
Course focus	Employability
Rationale	To have an overview of bacterial systematics and classification. The subject also explains the culture methods and techniques used in microbiology for better understanding of bacterial morphology.
Course Revision/Approval Date:	07/03/2024
Course Objectives (As per Blooms' Taxonomy)	 Remember To recognize, identify and differentiate the internal and external structures of bacterial cells. Apply To gain understanding of cultivation, preservation and control of bacteria. Analyses To develop basic skills necessary to work with bacterial strains. Create To know general techniques for isolation of pure cultures of bacteria. Understand To identify categories of bacteria and analyze their classification and diversity.

Course Content (Theory)	Weightage	Contact
		hours
Unit 1: Cell organization	20%	6
Cell size, shape and arrangement, glycocalyx, capsule, flagella, endo flagella,		
fimbriae and pili. Cell-wall: Composition and detailed structure of		
Archaebacterial cell wall and eubacterial cell wall, Gram and acid-fast staining		
mechanisms, lipopolysaccharide (LPS), spheroplasts, protoplasts, and L-forms.		
Effect of antibiotics and enzymes on the cell wall. Cell Membrane: Structure,		
function and chemical composition of bacterial and archaeal cell membranes.		
Cytoplasm: Ribosomes, mesosomes, inclusion bodies, nucleoid, chromosome		
and plasmids Endospore: Structure, formation, stages of sporulation.		
Unit 2: Growth, nutrition and Reproduction in Bacteria	20%	6
Nutritional requirements in bacteria and nutritional categories; Culture media and		
its types. Physical methods of microbial control. Chemical methods of microbial		
control. disinfectants, types and mode of action. Asexual methods of		
reproduction, logarithmic representation of bacterial populations, phases of		
growth, calculation of generation time and specific growth rate		



Unit 3: Bacteriological techniques & Microscopy	20%	6
Cultivation, maintenance and preservation/stocking of pure cultures; cultivation		
of anaerobic bacteria, and accessing nonculturable bacteria. Bright Field		
Microscope, Dark Field Microscope, Phase Contrast Microscope, Fluorescence		
Microscope, Confocal microscopy, Scanning and Transmission Electron		
Microscope		
Unit 4: Bacterial Systematics	20%	6
Aim and principles of classification, systematics and taxonomy, concept of		
species, taxa, strain; conventional, molecular and recent approaches to polyphasic		
bacterial taxonomy, evolutionary chronometers, rRNA oligonucleotide		
sequencing, signature sequences, and protein sequences. Differences between		
eubacteria and archaebacteria		
Unit 5: Important archaeal and eubacterial groups	20%	6
Archaebacteria: General characteristics, Classification and phylogenetic		
overview.		
Eubacteria: Morphology, metabolism, ecological significance and economic		
importance of following groups:		
Gram Negative: Non proteobacteria: General characteristics and classification		
with suitable examples.		
Gram Positive: Low G+ C (Firmicutes): General characteristics with suitable		
examples High G+C (Actinobacteria): General characteristics with suitable		
examples. Cyanobacteria: An Introduction."		

List Of Practical	Weightage	Conta ct hours
1: Study of Bacterial structure and morphology: Simple staining, Gram staining, Acid-fast staining (Demo), motility by hanging drop method	20%	8
2: Special staining: Capsule, endospore, metachromatic granule, flagella, Cell wal	20%	8
3. Measurement of cell size by micrometry method	10%	4
Pure culture study and biochemical characterization of bacteria (E. coli, S. aureus, Bacillus, Streptococci)	10%	4
5. Isolation and enumeration technique: Serial dilution, Standard Plate Count, Colony Forming Unit	10%	4
6. Preservation by Slant and Glycerol stock	10%	4
7. Isolation of anaerobic bacteria	10%	4
8. Preparation of various media (solid, semi solid, broth, synthetic, complex, defined, enriched media, differential and selective, and enrichment media)	20%	8
9. Isolation of Actinomycetes from soil (slide culture technique)	10%	4

Instructional Method and Pedagogy:

Audiovisuals, Seminars, Quiz, PPT, Demonstration, Case studies



	Course Objectives:	Blooms' Taxonomy Domain	Blooms' Taxonomy Sub Domain
CO1	Describe characteristics of bacterial cells, cell organelles, cell wall composition and various appendages like capsules, flagella or pili.	Remember	Explain, Describe, Discuss, Recall, Locate
CO2	Describe the nutritional requirements of bacteria for growth.	Apply	Apply, Practice, Interpret, Select, Correlate
CO3	Understand & apply microscopy techniques for identification of bacteria.	Analyses and Evaluation	Compare, Classify, Select, Investigate
CO4	Remember & Understand bacterial systematics	Remember & understand	Construct, Develop, Produce
CO5	Develop knowledge and understanding of the bacteria which grow under extreme environments.	Understand	Explain, Describe, outline, Predict, Summarize

1.	Reference books:				
	1. Atlas RM. (1997). Principles of Microbiology. 2nd edition. WM.T.Brown Publishers.				
	2. Black JG. (2008). Microbiology: Principles and Explorations. 7th edition. Prentice Hall				
	3. Madigan MT, and Martinko JM. (2014). Brock Biology of Microorganisms. 14th edition. Parker J. Prentice Hall International, Inc.				
	4. Srivastava S and Srivastava PS. (2003). Understanding Bacteria. Kluwer Academic Publishers, Dordrecht				
	5. Stanier RY, Ingraham JL, Wheelis ML and Painter PR. (2005). General Microbiology. 5th edition McMillan.				
	6. Tortora GJ, Funke BR, and Case CL. (2008). Microbiology: An Introduction. 9th edition Pearson Education.				
	7. Willey JM, Sherwood LM, and Woolverton CJ. (2013). Prescott's Microbiology. 9th edition. McGraw Hill Higher Education.				
	8. Cappucino J and Sherman N. (2010). Microbiology: A Laboratory Manual. 9th edition. Pearson Education Limited				
2.	Journals & Periodicals				
	1. Journal of Bacteriology				
	2. Microbiological Research				
	3. World Journal of Microbiology and Biotechnology				
	4. Microbiology Today				
3.	Other Electronic resources: https://microbiologysociety.org/why-microbiology-matters/what-is-microbiology/bacteria.html				



Evaluation Scheme	Total Marks				
Theory: Mid semester Marks	20 marks				
Theory: End Semester Marks	40 marks				
Theory: Continuous Evaluation Component	Attendance	05 marks			
Marks	MCQs	10 marks			
	Open Book Assignment	15 marks			
	Research Paper Review	10 marks			
	Total	40 Marks			
Practical Marks					
	Attendance	05 marks			
	Practical Exam	30 marks			
	Viva	10 marks			
	Journal	05 marks			
	Total	50 Marks			

	PSO1	PSO2	PSO3	PSO4	PSO5	PSO6		
CO1	2	0	1	0	1	1		
CO2	2	1	2	1	2	2		
CO3	2	0	1	1	0	1		
CO4	2	2	3	1	1	3		
CO5	2	2	1	0	2	3		

Mapping of POs & COs

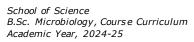
	PO1	PO2	PO3	PO4	PO5	PO6
CO1	3	0	1	2	2	3
CO2	2	1	2	2	1	2
CO3	2	2	2	0	0	1
CO4	2	2	3	0	1	2
CO5	2	1	2	1	1	2



COURSE CODE COURSE NAME SEMESTER
BSCM316 CHEMISTRY-I III

Teaching Scheme (Hours)						ching edit	
Lecture	Practical	Tutorial	Total Hours	Lecture	Practical	Tutorial	Total Credit
30	30	0	60	2	1	0	3

			,	
Course Pre-requisites	Basic knowledge of physical and organic chemistry.			
Course Category	Generic Elective			
Course focus	Employability			
Rationale	The concepts of physical chemistry help in understanding the Chemical Energetics Review of thermodynamics and the Laws of Thermodynamics and importance of Chemical Equilibrium. The fundamental concepts of organic chemistry help in understanding the chemistry of halogentaed hydrocarbons, phenols, ethers, epoxides, reactions of Carbonyl Compounds.			
Course Revision/	07/03/2024			
Approval Date:				
Course Objectives	To enable the student to: 1. To impart the knowledge of thermodynamics .			
(As per Blooms' Taxonomy)	 Concept of chemical equilibrium and ionic equilibria. To understand basic organic chemistry reactions. Detailed explanation of preparation and reactions of alkyl and aryl halides. 			
	5. Preparation of alcohols and phenols and the reactions involving them.6. Knowledge of preparation of ethers and different types of reactions.			
Course Content (Theory) Weightage Contant hour				
Unit 1: Chemical Energetics Review of thermodynamics and the Laws of Thermodynamics (1st, 2nd & 3rd). Important principles and definitions of thermochemistry. Concept of standard state and standard enthalpies of formations, integral and differential enthalpies of solution and dilution. Calculation of bond energy, bond dissociation energy and 20% 6				
	resonance energy from thermochemical data. Kirchhoff's equation.			





Unit 2: Chemical Equilibrium Free energy change in a chemical reaction. Law of chemical equilibrium. Distinction between G and Go, Le Chatelier's principle. Relationships between Kp, Kc and Kx for reactions involving ideal gases		6
Unit 3: Ionic Equilibria Strong, moderate and weak electrolytes, degree of ionization, factors affecting degree of ionization, ionization constant and ionic product of water. Ionization of weak acids and bases, pH scale, common ion effect. Salt hydrolysis: calculation of hydrolysis constant, degree of hydrolysis and pH for different salts. Buffer solutions.	20%	6
Unit 4: Alcohols and Phenols (Up to 5 Carbons) Alcohols: Preparation: Preparation of 1o, 2o and 3o alcohols: using Grignard reagent, Ester hydrolysis, Reduction of aldehydes, ketones, carboxylic acid and esters. Reactions: With sodium, HX (Lucas test), esterification, oxidation (with PCC, alk. KMnO4, acidic dichromate, conc. HNO3). Phenols: (Phenol case) Preparation: Cumene hydroperoxide method, from diazonium salts. Reactions: Electrophilic substitution: Nitration, halogenation and sulphonation. Reimer- Tiemann Reaction, Gattermann-Koch Reaction, Houben-Hoesch Condensation, Schotten Baumann Reaction	20%	6
Unit 5: Ethers, Aldehydes and ketones (aliphatic and aromatic) Cleavage of ethers with HI. Aldehydes and ketones (Formaldehyde, acetaldehyde, acetone and benzaldehyde) Preparation: from acid chlorides and from nitriles. Reactions — Reaction with HCN, ROH, NaHSO3, NH2-G derivatives. Iodoform test. Aldol Condensation, Cannizzaro's reaction, Wittig reaction, Benzoin condensation. Clemensen reduction and Wolff Kishner reduction. MeerweinPondorff Verley reduction.		6

Instructional Method and Pedagogy:

PPT, Demonstration, Video, Case study

List Of Practical	Weightage	Conta ct hours
Based on Unit 1 - 5		



Course Outcome:	Blooms' Taxonomy Domain	Blooms' Taxonomy Sub Domain
After successful completion of the above course, students will be able to: CO1: Interpret the Laws of thermodynamics.	Understanding	Describe
CO2: Explain the free energy changes during chemical equilibria. CO3: Recognise the fundamentals of Organic chemistry, electrophilic and nucleophilic reactions. CO4: Explain the properties, preparation and reactions of alcohols and phenols CO5: Explain the different types of reactions of aldehydes and ketones.	Understandand Apply Understand and Apply Remember and Understand Understanding	Explain and examine Explain and examine Define and explain Classify and Explain

	Learning Resources						
1.	Textbook/ Reference books:						
	1. T. W. Graham Solomons: Organic Chemistry, John Wiley and Sons.						
	2. Peter Sykes: A Guide Book to Mechanism in Organic Chemistry, Orient Longman						
	H. Petrucci: General Chemistry 5th Ed. Macmillan Publishing Co.: New York (1985).						
	3. I.L. Finar: Organic Chemistry (Vol. I & II), E. L. B. S.						
	4. R. T. Morrison & R. N. Boyd: Organic Chemistry, Prentice Hall.						
	5. Arun Bahl and B. S. Bahl: Advanced Organic Chemistry, S. Chand.						
	6. G. M. Barrow: Physical Chemistry Tata McGraw-Hill (2007).						
	7. G. W. Castellan: Physical Chemistry 4th Edn. Narosa (2004).						
	8. J. C. Kotz, P. M. Treichel & J. R. Townsend: General Chemistry Cengage Lening						
	India Pvt. Ltd., New Delhi (2009).						
	9. B. H. Mahan: University Chemistry 3rd Ed. Narosa (1998).						
2.	Journals & Periodicals:						
	Journal: Journal of Chemical Sciences						
	Periodicals: Chemistry Today						
3.	Other Electronic Resources:						
	Other Electronic resources: NPTEL, SWAYAM, MERLOT (Links available in GSFC						
	University Link).						

Evaluation Scheme	Total Marks				
Theory: Midsemester Marks	20 marks				
Theory: End Semester Marks	40 marks				
Theory: Continuous					
Evaluation Component Marks	Attendance 05 marks				
	MCQs	10 marks			
	Open Book Assignment	15 marks			



	Research Paper Review	10 marks
	Total	40 Marks
Practical Marks		
	Attendance	05 marks
	Practical Exam	20 marks
	Viva	10 marks
	Journal	10 marks
	Discipline	05 marks
	Total	50 Marks

	PSO1	PSO2	PSO3	PSO4	PSO5	PSO6
CO1	1	2	1	0	0	2
CO2	1	2	1	1	0	2
CO3	1	2	1	1	0	0
CO4	2	2	1	0	0	0
CO5	1	2	2	0	0	0

Mapping of POs & COs

g 011 05 00 005						
	PO1	PO2	PO3	PO4	PO5	PO6
CO1	1	1	2	1	0	0
CO2	1	1	1	1	0	0
CO3	1	1	1	1	0	0
CO4	1	1	1	1	0	0
CO5	3	1	1	1	0	0



COURSE CODE	COURSE NAME	SEMESTER
BSMA316	MATHEMATICS-III	III

Teaching Scheme (Hours)				Teaching Credit			
Lecture	Practical	Tutorial	Total Hours	Lecture	Practical	Tutorial	Total Credit
30	30	0	60	2	1	0	3

Basic knowledge of algebra and calculus. Basic knowledge of derivatives					
and integration.					
Discipline Specific elective course					
Skill development					
This course involves solving the system of linear equations,					
approximation of root of non-linear equation. Students also learn					
interpolation methods for equi-distant and non equi-distant data. Also they					
develop the skill to find numerical integration and differentiation and					
techniques to solve IVP.					
07/03/2024					
07/05/2021					
1 C.1 . M.1 .1 1 C					
1: Solve: Make the students familiarise with the ways of solving					
complicated mathematical problems numerically					
2:Understand: To help them become familiar with MATLAB and other					
convenient numerical software such as Microsoft Excel and with simple					
programming					
3: Recognize , Find : Obtain numerical solutions to non-algebraic					
equations and system of linear equations.					
4: Understand: Describe and understand of the several errors and					
approximation in numerical methods					
5: Understand, Find: Understanding of several available Solution					
techniques for differential Equations in One Variable. Study of Curve					
Fitting and Interpolation.					
Blooms' Taxonomy word should be highlighted					

Course Content (Theory)	Weightage	Contact
		hours
Unit 1:	20%	6
Solution of Algebraic and Transcendental Equations: Bisection, False		
position, Newton Raphson Method, Secant Method. Solution using Matlab		
Unit 2:	20%	6
Solution of system of Linear Equations: LU decomposition method, Gauss		
Jacobi Method, Gauss Seidel method. Solution using Matlab		
Unit 3:	20%	6
Interpolation: Newton's forward and backward interpolation, Newton's divided		
difference interpolating polynomials, Lagrange Interpolating polynomials.		
Solution using Matlab.		
Unit 4:	20%	6
Numerical Integration: Trapezoidal rule, Simpson's one third and 3/8th rule.		
Solution using Matlab		
Curve Fittings: General Linear Least Squares, Fitting of quadratic and		
exponential curves. Solution using Matlab.		
Unit 5:	20%	6



Numerical methods for Solution of ordinary differential equation: Euler's method, Modified Euler's Method, Runge Kutta forth ordered method, Solution using Matlab.

List Of Practical	Weightage	Contact hours
1: Introduction to MATLAB, Matrix algebra, functions	7%	2
2: Loops: For, if else, while Programme for Bisection Method	7%	2
3: Programme for Regula-falsi and Secant Method	7%	2
4: Programme for Newton-Raphson's Method	7%	2
5: Programme for Difference Table	7%	2
6. Programme for Newtons's Forward and Beckward Interpolation	8%	2
7. Programme for Newtonds Divided Difference Interpolation Method	7%	2
8. Programme for Lagrange's Method	7%	2
9. Use of Curve fitting Toolbox	7%	2
10. Programme for Numerical integrations (Trapezoidel and simpson's rules)	8%	2
11. Solving system of linear equations	7%	2
12. Plotting 2D and 3D graphs	7%	2
13. Programme for Euler's Method	7%	2
14. Practive test and Revision	7%	4

Instructional Method and Pedagogy: (Max. 100 words) Chalk-board, Presentation, Use of Mathlab, Excel and Geogebra. Group Discussion, Case Study, Quizziz application.

Course Objectives:	Blooms' Taxonomy Domain	Blooms' Taxonomy Sub Domain
After successful completion of the above course, students will be able to:	Apply	Describe, Find
CO1: Apply, solve: Apply numerical methods to find out solution of non-algebraic equations using different methods		
CO2: Apply, solve: Apply numerical methods to find numerical	Understand,	Demonstrate &
solution of system of linear equations.	Remember	Examine, Find
CO3: Demonstrate , find : Apply various interpolation methods	Understand,	Demonstrate &
and finite difference concepts	Remember	Examine, Find
CO4: Demonstrate: Work out numerical integration and Cure	Evaluate	Examine, Find
Fitting whenever and wherever routine methods are not		
applicable		
CO5: Solve: Work numerically on the ordinary differential	TT - 14 1	Define, Classify,
equations using different methods through the theory of finite	Understand,	Describe,
differences.	Remember,	Demonstrate &
	Apply & Analyse	Examine

Learning Resources



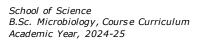
1.	Reference Books: 1 S. S. Sastry, Introductory methods of Numerical Analysis, 5th Edition, Prentice-Hall India, 2012 2. G. Shankar Rao, Numerical Analysis, New Age International Pvt. Ltd., 2006 3. P.C. Biswal, Numerical Analysis, Prentice-Hall India, 2008
2.	Journals & Periodicals: Mathematics Open
	<u>Mathematics Open</u>
3.	Other Electronic Resources:
	MATLAB, Microsoft Excel, Geogebra Toolbox

Evaluation Scheme	Total Marks	
Theory: Mid semester Marks	20 marks	
Theory: End Semester Marks	40 marks	
Theory: Continuous Evaluation Component Marks	Attendance MCQs Open Book Assignment Open Book Assignment Total	05 marks 10 marks 15 marks 10 marks 40 Marks
Practical Marks	Attendance Practical Exam Viva Journal Discipline Total	05 marks 20 marks 10 marks 10 marks 50 Marks

	PSO1	PSO2	PSO3	PSO4	PSO5	PSO6
CO1	1	2	1	0	0	2
CO2	1	2	1	1	0	2
CO3	1	2	1	1	0	0
CO4	2	2	1	0	0	0
CO5	1	2	2	0	0	0

Mapping of POs & COs

	022 05 00	005				
	PO1	PO2	PO3	PO4	PO5	PO6
CO1	1	1	2	1	0	0





CO2	1	1	1	1	0	0
CO3	1	1	1	1	0	0
CO4	1	1	1	1	0	0
CO5	3	1	1	1	0	0



COURSE CODE	COURSE NAME	SEMESTER
BSPY317	PHYSICS-II	III

Teaching Scheme (Hours)				Teaching Credit			
Lecture	Practical	Tutorial	Total Hours	Lecture Practical Tutorial To			
30	30	0	60	2	1	0	3

Course Pre-requisites	Understanding of basic physics up to school level (10+2 level).
Course Category	Generic Elective
Course focus	Employability
Rationale	To gain basic knowledge of physics.
Course Revision/ Approval Date:	14/03/2020
Course Objectives (As per Blooms' Taxonomy)	To enable the student to: 1:Understand the physical significance of mathematical operations. 2: Employ the knowledge of electrostatics in daily life applications. 3: Understand the basics of magnetostatics and its applications. 4: Remember the laws of thermodynamics and their applications. 5: Explain the thermodynamic potentials and transport properties.

Course Content (Theory)	Weightage	Contact hours
Unit 1: Vector Analysis Scalar and Vector product, gradient, divergence, Curl and their significance, Vector Integration, Line, surface and volume integrals of Vector fields, Gauss-	12%	6
divergence theorem and Stoke's theorem of vectors		
Unit 2: Electrostatics Electrostatic Field, Electric flux, Gauss's theorem of electrostatics. Applications of Gauss theorem - Electric field due to a point charge, infinite line of charge, uniformly charged spherical shell and solid sphere, plane charged sheet, charged conductor. Electric potential as a line integral of electric field, electric potential due to a point charge, electric dipole, uniformly charged spherical shell and solid sphere. Calculation of electric field from potential. Capacitance of an isolated spherical conductor. Parallel-plate, spherical and cylindrical condenser. Energy per unit volume in the electrostatic field. Dielectric medium, Polarization, Displacement vector. Gauss's theorem in dielectrics. Parallel plate capacitor completely filled with dielectric	23%	14
Unit 3: Magnetostatics Biot-Savart's law and its applications - straight conductor, circular coil,	23%	14



solenoid carrying current. Divergence and curl of magnetic field. Magnetic vector potential. Ampere's circuital law. Magnetic properties of materials: Magnetic intensity, magnetic induction, permeability, magnetic susceptibility. Brief introduction of dia-, para- and ferromagnetic materials. Electromagnetic Induction: Faraday's laws of electromagnetic induction, Lenz's law, self and mutual inductance, L of single coil, M of two coils. Energy stored in the magnetic field.		
Unit 4:: Laws of Thermodynamics Thermodynamic Description of system: Zeroth Law of thermodynamics and temperature. First law and internal energy, conversion of heat into work, Various Thermodynamic Processes, Applications of First Law: General Relation between CP and CV, Work Done during Isothermal and Adiabatic Processes, Compressibility and Expansion Coefficient, Reversible and irreversible processes, Second law and Entropy, Carnot's cycle & theorem, Entropy changes in reversible & irreversible processes, Entropy-temperature diagrams, Third law of thermodynamics, Unattainability of absolute zero	22%	14
Unit 5: Thermodynamic Potentials Enthalpy, Gibbs, Helmholtz and Internal Energy functions, Maxwell's relations and applications - Joule-Thomson Effect, Clausius-Clapeyron Equation, Expressions for (CP – CV), CP/CV, TdS equations. Kinetic Theory of Gases: Maxwell's law of distribution of velocities and its experimental verification, Mean free path (Zeroth Order), Transport Phenomena: Viscosity, Conduction and Diffusion (for vertical case), Law of equipartition of energy (without derivation of expressions) and its applications to specific heat of gases; mono-atomic and diatomic gases.	20%	12

List Of Practical	Weightage	Contact hours
1: Study of working and characteristics of a Van de Graff generator	8	3
2:Determination of the magnetic moment of a given magnet using magnetometer at Gauss A and Gauss B position using Deflection magnetometer.	14	4
3: Determination of the ratio of magnetic moment of the given bar magnets using Vibration magnetometer.	14	4
4: To study the variation of magnetic field with distance along the axis of a circular coil carrying current by plotting a graph and calculate the radius of the coil using given laboratory setup.	14	4
5: To verify Stefan-Boltzmann law of thermal radiation by electrical method.	8	3
6: To verify the relation between the thermal emfs of a thermocouple and temperature difference between two hot junctions and verify Seebeck Effect.	14	4
7: Determine the molar heat capacities of air at constant volume Cv and at constant pressure Cp.	14	4
8: To determine the Coefficient of Thermal Conductivity of Copper by Searle's Method.	14	4



Instructional Methodand Pedagogy: (Max. 100 words)
PPT, Demonstration, Video, Case study

Course Outcome:	Blooms' Taxonomy Domain	Blooms' Taxonomy Sub Domain
After successfulcompletion of the above course, students will be able to:		
CO1 Describe the physical significance of mathematical operations.	Understanding	Describe
CO2 Employ the knowledge of electrostatics in daily life applications.	Understandand Apply	Explain and examine
CO3 Explain the use of magnetostatics in various applications	Understandand Apply	Explain and examine
CO4 Interpret the laws of thermodynamics and understand its applications	Remember and Understand	Define and explain
CO5 Explain the thermodynamic potentials and transport properties	Understanding	Classify and Explain

	Learning Resources
1.	Reference Books:
	 C. Chattopadhyay, R. Rakshit, Electricity and Magnetism (with Electro-Magnetic Theory and Special Theory of Relativity), Current Distributors, 1989. Brij Nandan Lal, N. Subrahmanyam, Heat Thermodynamics and Statistical Physics, S. Chand Limited, 2008.
2.	Journals & Periodicals: Journal of Undergraduate Reports in Physics (JURP), Society of Physics Students.
3.	Other Electronic Resources: Feynman Lectures in Physics: https://www.feynmanLectures.caltech.edu/

Evaluation Scheme	Total Marks			
Theory: Midsemester Marks	20 marks			
Theory: End Semester Marks	40 marks			
Theory: Continuous Evaluation Component	Attendance	05 marks		



Marks	MCQs	10 marks
	Open Book Assignment	15 marks
	Research Paper Review	10 marks
	Total	40 Marks
Practical Marks		
	Attendance	05 marks
	Practical Exam	20 marks
	Viva	10 marks
	Journal	10 marks
	Discipline	05 marks
	Total	50 Marks

	PSO1	PSO2	PSO3	PSO4	PSO5	PSO6
CO1	1	1	0	0	0	1
CO2	1	1	0	0	0	1
CO3	1	1	0	0	0	1
CO4	1	2	0	0	0	1
CO5	1	2	0	0	0	1

Mapping of POs & COs

	PO1	PO2	PO3	PO4	PO5	PO6
CO1	1	1	1	0	0	0
CO2	1	1	1	0	0	0
CO3	1	1	2	0	0	0
CO4	1	2	2	0	0	1
CO5	1	2	2	0	0	1



COURSE CODE	COURSE NAME	SEMESTER
AECC301	ENTREPRENEURSHIP	III
	DEVELOPMENT	

Teaching Scheme (Hours)				Teaching Credit			
Lecture	Practical	Tutorial	Total Hour	Lecture Practical Tutorial Tot			
			S				
30	0	0	30	2	0	0	2

G B 11	
Course Pre-requisites	Knowledge and skills of entrepreneurship.
Course Category	Ability Enhancement Compulsory Course
8 0	
Course focus	Entrepreneurship
Rationale	Entrepreneurs have been instrumental in spurring social change and
	improving the way people live and work. They help raise the standard of
	living for everyone by creating jobs and making products safer, less
	expensive, and more functional.
	A
Course Revision/ Approval	14/03/2020
Date:	
Course Objectives	To enable the student to:
(As per Blooms'	1: Students will develop skills for evaluating, articulating, refining, and
Taxonomy)	pitching a new product or service offering
	2: Identify the elements of success of entrepreneurial ventures
	3: Analyze Feasibility of the project (Financial and Non-Financial) and
	interpret business plan.
	4: Demonstrate and present successful work, collaboration and
	division of tasks in a multidisciplinary and multicultural team.
	5: Demonstrate understanding and application of the tools necessary
	to create sustainable and viable Businesses.

Course Content (Theory)	Weightage	Contact hours
Unit 1: Entrepreneurship Concept, knowledge and skills requirement; characteristic of successful entrepreneurs; role of entrepreneurship in economic development; entrepreneurship process; factors impacting mergence of entrepreneurship; managerial vs. entrepreneurial approach and emergence of entrepreneurship. Entrepreneurial Motivation.	20%	6



Unit 2: Creativity and Entrepreneurship Steps in Creativity; Product Design & Influencing Factors (Legal, Ethical & Environmental); Generating business idea –sources of new ideas, methods of generating ideas, creative problem solving, opportunity recognition; environmental scanning, competitor and industry analysis.	20%	6
Unit 3: Feasibility Study (Non-financial Aspects) Market feasibility, Technical feasibility, operational feasibility, Legal feasibility, Human Resource Feasibility, Supply Feasibility.	20%	6
Unit 4:Feasibility Study (financial Aspects) Cost classification- Fixed vs. Variable; Cost Determination- Material, Labour, Overheads; Product Profitability- Concepts of Break-even, Margin of Safety, Angle of Incidence, Key-factor, Profit-Volume ratio; Balance Sheet & Profit & Loss Account- Concepts & Structure; Budgeting; Financing Schemes from Government, specially schemes for women; Venture Capital & Angel Investing	20%	6
Unit 5: Detailed Project Report & Business Plan Project Report- components; Preparation of Business Plan; Pitching the Business Plan, Attracting Angel Investors. (A group of THREE students will prepare a DPR, and Business Plan on selected product or service in the course as a Project/Assignment.	20%	6

Course Outcomes:	Blooms' Taxonomy Domain	Blooms' Taxonomy Sub Domain
After successful completion of the above course, students will be able to:		
CO1 Students will develop skills for evaluating, articulating, refining, and pitching a new product or service offering.	Evaluate	Define, Classify and describe
CO2 Identify the elements of success of entrepreneurial ventures	Apply and Understand	Define,Classify, Describe,Demostrate and Examine
CO3 Analyze Feasibility of the project (Financial and Non-Financial) and interpret business plan CO4 Demonstrate and present successful work,	Analyse	Define, Classify, Describe, Demostrate
CO4 Demonstrate and present successful work, collaboration and division of tasks in a multidisciplinary and multicultural team.	Create and Understand Create and	and Examine Define, Classify, Describe, Demostrate
CO5: Demonstrate understanding and application of the tools necessary to create sustainable and viable Businesses	Understand	and Examine Define,Describe and Demostrate



	Learning Resources						
1.	Textbook:	Textbook: 1. Fundamentals of Entrepreneurship.					
		*					
2.	Reference books	2. Managing Entrepreneurship.					
2.		oreneurship: New Venture Creation.					
	-	erns of Entrepreneurship.ship.					
	_	ika SS. Entrepreneurship and Small Business M	Ianagement, Sultan Chand &				
	Sons.	r i i i i i i i i i i i i i i i i i i i					
3.		onal Journal of Entrepreneurship.					
4.	Periodicals -https://www.jemi.edu.pl/						
5.	Other Electronic r	esources:https://innovation-entrepreneurship.sp	ringeropen.com/				
Evalu	uation Scheme	Total Marks					
Theory: Mid	lsemester Marks	20 marks					
Theory: End	Semester Marks	40 marks					
	Continuous n Component	Attendance	05 marks				
	Iarks	MCQs	10 marks				
	Open Book Assignment 15 marks						
		Article Review	10 marks				
		Total	40 Marks				

	PSO1	PSO2	PSO3	PSO4	PSO5	PSO6
CO1	3	2	1	3	2	0
CO2	2	0	1	1	3	0
CO3	2	0	1	0	0	0
CO4	2	3	2	2	0	1
CO5	3	1	3	2	1	2

Mapping of POs & COs

	PO1	PO2	PO3	PO4	PO5	PO6
CO1	2	2	1	3	1	1
CO2	1	2	1	1	3	1
CO3	2	2	3	2	2	2
CO4	2	2	1	3	2	2
CO5	2	2	3	3	2	3



	COURSE CODE COURSE NAME WORLD GEOGRAPHY		2 2	T 0	P 0	C 2
Total C	Credits: 2	Total Hours in semester: 30	Γ	otal M	arks: 1	100
1		Course Pre-requisites: Zeal to lear	n the s	ubject		
2		Course Category: Skill Enhanceme	ent Co	urses.		
3		Course Revision/ Approval	Date:			
4		Course Objectives :				
4.1 To	4.1 To give students basic knowledge about geography and general geographic features.					itures.
4.2 To	4.2 To enable students to understand physical geography and its impact on human life.					life.
	4.3 To enable students to social geography and its impact on country.					
4.4 To enable students to economical geography and its impact on country.					·	
4.5 To	4.5 To provide students competitive advantage in competitive examinations for civil services.					

Course Content	Weighta	Contact	Pedagogy
	ge	hours	
Unit 1:	20%	06	Power
			Point
General Geography: Earth in Solar system, Motion of			Presentati
the Earth, Concept of time, Season, Internal Structure			on/ Chalk/
of the Earth, Major land forms and their features.			Duster
Atmosphere-structure and composition, elements and			
factors of Climate, Air masses and Fronts, atmospheric			
disturbances, climate change. Oceans: Physical,			
chemical and biological characteristics, Hydro-logical			
Disasters, Marine and Continental resources.			
Unit 2: Physical: World, India and Gujarat : Major	20%	06	Power
physical divisions, Earthquakes, landslides, Natural	-		Point
drainage, climatic changes and regions, Monsoon,			Presentati
Natural Vegetation, Parks and Sanctuaries, Major Soil			on/ Chalk/
types, Rocks and Minerals.			Duster
Unit 3: Social: World, India and Gujarat : distribution,	20%	06	Power
density, growth, Sex-ratio, Literacy, Occupational			Point
Structure, SC and ST Population, Rural-Urban			Presentati
components, Racial, tribal, religious and linguistic			on/ Chalk/
groups, urbanization, migration and metropolitan			Duster
regions.			



Unit 4: Economic: World, India and Gujarat: Major		06	Power
sectors of economy, Agriculture, Industry and Services,			Point
their salient features. Basic Industries-Agro, mineral,			Presentati
forest, fuel and manpower based Industries, Transport			on/ Chalk/
and Trade, Pattern and Issues.			Duster
Unit 5: World Maps: Types of Maps and its	20%	06	Hands on
significance, Use of Maps and its applications			activities/
			Power
			Point
			Presentati
			on/ Chalk/
			Duster

	Learning Resources					
1.	Textbooks:					
	1. D R Khullar, Physical, Human and Economic Geography for					
	Civil Services Examination, GPK Publications					
2.	Reference Books:					
	1. Majid Hussain, World Geography, KP Publications					
3.	Journals & Periodicals:					
	1. The Geographical Journal, Royal Geographic Society					
5.	Other Electronic Resources:					
	1. NAVIC					
	2. BHUVAN					
	3. Google Map					

Evaluation Scheme	Total Marks				
Mid semester Marks	30				
End Semester Marks	50				
	Attendance	5 marks			
	Quiz	5 marks			
Continuous Evaluation	Skill enhancement activities / job preparations	5 marks			
Marks	Presentation/ miscellaneous activities	5 marks			
	1. Able to understand basic knowledge about geography and general geographic features.				



Course Outcomes	Able to physical geography and its impact on human life.
	3. Understand social geography and its impact on country.
	4. Will get knowledge about economical geography and its impact on country.
	5. Understanding of maps and its use in specific applications.



			L	T	P	С
COURSE CODE		COURSE NAME	2	0	0	2
СВ	CS303	App Development Programming		Ü	Ū	_
Total (Credits:2	Total Hours in semester: 30	1	otal M	arks:1	00
1	Course P	re-requisites: Prior knowledge required	of Obj	ect Ori	ented	
	Programm	ning Concepts (C++ or Java)				
2	Course Category:					
	Engineeri	ng Specific Elective				
3	Course R	evision/ Approval Date:				
4	4 Course Objectives :					
4.1 To fa	acilitate stud	ents to understand android SDK				
4.2 To help students to gain a basic understanding of Android application development						
4.3 To inculcate working knowledge of Android Studio development tool						
4.4 To inculcate testing of application before deployment						
4.5 To us	se of commo	on Android APIs				

Course Content	Weightage	Contact hours	Pedagogy
Unit 1: Introduction to Android Theory: The Android Platform, Android	20%	6	Computer Based Learning,
SDK, Eclipse Installation, Android Installation, Building you First Android application, Understanding Anatomy of Android Application, Android Manifest			Chalk – Talk, Presentation
file. Unit 2: Android Application Design Essentials Theory: Anatomy of an Android applications, Android terminologies, Application Context, Activities, Services, Intents, Receiving and Broadcasting Intents, Android Manifest File and its common settings, Using Intent Filter, Permissions.	20%	6	Computer based learning, Chalk – Talk, Presentation
Unit 3: Android User Interface Design Essentials Theory: User Interface Screen elements, Designing User Interfaces with Layouts, Drawing and Working with Animation.	20%	6	Chalk – Talk, Presentation
Unit 4: Testing Android applications	20%	6	Computer



Theory: Publishing Android application, Using Android preferences, Managing Application resources in a hierarchy, working with different types of resources.			based learning, Chalk – Talk, Presentation
Unit 5: Using Common Android APIs Theory: Using Android Data and Storage APIs, Managing data using Sqlite, Sharing Data between Applications with Content Providers, Using Android Networking APIs, Using Android Web APIs, Using Android Telephony APIs, Deploying Android Application to the World.	20%	6	Computer based learning, Chalk – Talk, Presentation

Learning Resources		
1.	Textbooks:	
2.	Reference Books:	
3.	Journals & Periodicals:	
5.	Other Electronic Resources:	

Evaluation Scheme	Total Marks	
Mid semester Marks		
End Semester Marks		
	Attendance	5 marks
	Quiz	5 marks
Continuous Evaluation	Skill enhancement activities / case study	5 marks
Marks	Presentation/ miscellaneous activities	5 marks
	1. Identify various concepts of mobile programming the make it unique from programming for other platforms	
Course Outcomes	omes 2. Critique mobile applications on their design pros and	
	cons	
	3. Utilize rapid prototyping techniques to design and develop sophisticated mobile interfaces	
	4. Program mobile applications for the Android operating	







system that use basic and advanced phone features
5. Deploy applications to the Android marketplace for
distribution.



emic Year, 20	024-25		170,60	April 1	Wannau are eur	0.00
COURSE CODE CBCS305 Total Credits: 2		COURSE NAME	L 2	T 0	P 0	C 2
		COMPUTER AIDED DESIGN Total Hours in semester: 30	Т	otal M	arks:	100
1	Course Pre-requisites:				100	
2	Course Category: Generic Elective					
3	Course Revision/Approval Date: 11th March 2020					
4	Course Objectives: to have fundamental understanding of computer aided design concept and learn how to use Solidworks CAD tool					

Course Content	Weightage	Contact hours	Pedagogy
Unit 1: Introduction	10%	6	Software based
Basic concept of CAD, 2D sketching,		O	learning
Basic tools, Editing & Modifying			1001111111g
sketches, adding dimension & relation to			
sketches, advance dimension technique			
Unit 2: Solid Modelling	30%	8	Project Based
			Learning
Understanding of parent- child relations,			Software Based
features, wireframe and surface model			Learning
understanding, extrude, mirror.			Interactive
			Sessions
Unit 3: Assembly	30%	8	Project Based
,			Learning
Inserting components, top-down			Software Based
assembly, sub-assemblies, using mate in			Learning
assembly, bottom-up assembly			Interactive
			Sessions
Unit 4: Drafting	30%	8	Project Based
			Learning
Drawing Overview, Smart			Software Based
Dimensioning, Annotation & Symbols,			Learning
Sheet format & Templates, Ballooning			Interactive
for Nomenclature Adding Bill of material			Sessions
& Tables			

Learning R	Resources
1.	Textbooks:
	Solidworks 2019 and Engineering Graphics: An integrated Approach by
	Randy H. Shih
2.	Reference Books:
	Ibrahim Zeid "Mastering CAD CAM" Tata McGraw-Hill Publishing Co.2007





Evaluation Scheme	Total Marks	
Mid semester Marks	30	
End Semester Marks	50	
	Attendance	5 marks
	Quiz	5 marks
Continuous Evaluation	Skill enhancement activities / case study	5 marks
Marks	Presentation/ miscellaneous activities	5 marks
	Student will be able to construct 3-D solid models, 2-D	
drawings, and assembly and sub-assembly structures Course Outcomes		semoly structures



		COURSE NAME Project Management	L 30	T	P 0	C 2
Total C	Credits:02	Total Hours in semester: 30	Γ	otal M	[arks:1	.00
1	1 Course Pre-requisites: Domain Knowledge of project making					
2	Course Category: CBCS					
3	Course Revision/ Approval Date:					
4	Course Objectives:					
1. To und	1. To understand the importance of project management.					
2. To und	2. To understand how the project will organize, plan.					
3. The project has to monitor and control the success of the project.						

Course Content	Weight age	Contact hours	Pedagogy
Unit 1: The world of Project Management 1. Definition of project, 2. Project Management Vs. General Management, 3. Three goals of project, The life cycle of projects	20%	5 Hours	Lecture
Unit 2 Project Planning 2. The contents of a project plan The planning process-overview	15%	5 Hours	Lecture/ Cases
Unit 3: Organizing the project 1. The PM's Roles 2. The PM's responsibility to the project, Selection of a Project Manager, 3. Project Management as a profession	15%	5 Hours	Lecture/ Cases / Role Playing
Unit 4: Budgeting the Project 1. Methods of budgeting, 2. Cost estimating, 3. Improving Cost Estimates, 4. Budget Uncertainty and risk management	30%	8 Hours	Quizzes/ Cases /
 Unit 5 Monitoring and Controlling the Project The plan-monitor-control cycle, Data collection and reporting, Earned value, Project control, Designing the control system, 	20%	7 Hours	Cases Analysis and discussion



Learning F	Resources:
1.	.Text Book: Project management: A systems approach to planning scheduling and controlling by Harlod Kerzner and Van Nostr and, John Wiley& Sons, 11thed., 2013, ISBN:978-1-118-
	02227-6.
2.	Reference Books: 1. Project Management by Samuel J. Mantel, Jr., Jack R. Meredith, Scott M. Shafer, Margaret M.Sutton with M. R. Gopalan (WILEY-INDIA) 2. Project Management and Appraisal by Sitangshu Khatua (Oxford)
3.	News Paper 1. Economic Times 2. Business Standard

	Evaluation Scheme				
Mid semester Marks		20 Marks			
End Semester Marks		40 Marks			
	Attendance	e	5 marks		
	Quiz		5 marks		
	Skill enhancement ac	tivities / case	5 marks		
Continuous Evaluation	study				
Marks 40	Presentation/ miscellan activities	eous	5 marks		
	1. Design a solution to a				
Course Outcomes	Develop strategies to control, and close pro				
	3. Analyze and evaluate a practices, and theorie		s strategies,		



COURSE CBCS		COURSE NAME Research Methodology	02	T 0	P 0	<u>C</u> 2			
Total C	Credits:02	Total Hours in semester: 30	T	otal N	Iarks:1	00			
1	Course Pr	e-requisites: Numerical Knowledge							
2	Course Category: CBCS Generic Elective								
3	Course Re	evision/ Approval Date: Academic Council	meet						
4	Course Ol	bjectives:							
4.1 To e	quip the stu	idents with the basic understanding of the	research	metho	odology				
4.2 To ur	nderstand th	ne different types of research methods.							
4.3 To ur	4.3 To understand importance of systematic approach to problem solving								
4.4 To d	evelop the	ability to interpretation of research results	and find	lings					

	000	Contact hours	Pedagogy
Unit 1: Introduction	age 15%	5 Hours	Lecture/
1. Objective, Types, and Methods	13%	3 Hours	Class
2. Process. Research Problem,			participation
ŕ			participation
3. Concept of Research and Its Application in Various sectors.			
	150/	5Hours	I a atuma /
Unit 2 Research Designs:	15%	SHours	Lecture/
1.Understanding Research design			Cases
2.Qualitative			
3.Quantitative Research			
Unit 3: Scaling Techniques	20%	5 Hours	Lecture/ /
1. Importance Measurement			Cases / Role
2. Scaling Techniques			Playing
3. Measurement in Research,			
4. Types of Measurement Scales			
Unit 4: Data Collection	30%	8 Hours	Quizzes/
1. Primary and Secondary Methods of Data			Cases /
Collection – Surveys, Observation,			
Experimentation			
2. Hypothesis testing, Descriptive Statistics			
Unit 5 Report Preparation	20%	7 Hours	Cases
Types and Layout of Research Report			Analysis and discussion



Learning I	Resources
1.	Textbooks: Research Methodology, D. K. Bhattacharyya, Excel
2.	Reference Books:
	1. Research Methodology for Management and social Science, Adithan
	Bhujange, Excel
	2. 2.CR Kothari, Research Methodology
3.	Journals & Periodicals:
	1. Journal of Indian Business Research
	2. International Journal of Statistics and Analysis

Evalua Sche		Total Marks: 100
Mid semester Marks	30 Mar	rks
End Semester Marks	50 Mar	rks
	Attendance	5 marks
	Quiz	5 marks
Continuous Evaluation	Skill enhancement activities / c study	ase 5 marks
Marks 20	Presentation/ miscellaneous activities	5 marks
_	1 Understand problem an	71 1
	2. Gain exposure to globa	ally accepted theories
Course Outcomes	3. Understand methodologies	s of conducting research



		Semester – IV					
Sr. No.	Course Code	Course Title	L	T	P	С	Marks
	A. Major						
1	BSMO411	Industrial Microbiology	3	0	1	4	150
2	BSMO412	Recombinant DNA Technology	3	0	1	4	150
3	BSMO413	Microbial Physiology & Metabolism	3	0	1	4	150
		B. i. Minor (Compulsory)		l			
4	BSCM417	Chemistry - II	2	0	1	3	75
		B. ii. Minor (Elective)					
5	BSMA415	Biostatistics	2	1	0	3	
6	BSPY415	Biophyscics	2	0	1	3	75
C. M	Iultidisciplinary	Additional Credits under Choice Ba	sed C	redit	Syste	m (Aı	ny One)
7	CBCS		2	0	0	2	100
		D. Ability Enhancement Cour	se				
8	AECC401	Environmental Science	2	0	0	2	50
		E. Skill Enhancement Course	e				
9	SECC401	Internship	0	0	2	2	50
		F. Value Added Course					
10	VACC401	NCC/Sports	0	0	2	2	50
				1	otal	22	700



Teaching Scheme Semester – III

G.	C			aching Hours			Т	eachi	ing C	redit			Evaluat	tion Schei	me	
Sr. No.	Course Code Course Na	Course Name	L	P	Т	Total	L	P	Т	Total	Theory: MS Marks	Theory: CEC Marks	Theory: ES Marks	Theory Marks	Practical Marks	Total Marks
	A. Major Courses															
1.	BSMO411	Industrial Microbiology	3	2	0	5	3	1	0	4	20	40	40	100	50	150
2.	BSMO412	Recombinant DNA Technology	3	2	0	5	3	1	0	4	20	40	40	100	50	150
3	BSMO413	Microbial Physiology & Metabolism	3	2	0	5	3	1	0	4	20	40	40	100	50	150
									В.	Minor Co	urses					
3.	BSCM415	Chemistry – II	2	2	0	4	2	1	0	3	20	40	40	100	50	75
4.	BSMA415	Biostatistics	2	0	1	3	2	0	1	3	20	45	75	150	0	75
5.	BSPY415	Biophyscics	2	2	0	4	2	1	0	3	20	40	40	100	50	75
							C. N	Aultic	liscip	linary Co	ourses (Any	one)				
6.	CBCS		2	0	0	2	2	0	0	2	20	40	40	100	00	100
								D. At	oility	Enhance	ment Course	S				
7.	AECC401	Environmental Science	2	0	0	2	2	0	0	2	20	40	40	100	00	100



		E. Skill Enhancement Courses														
8.	SECC301	Industrial Internship	0	2	0	2	0	2	0	2	00	00	00	50	00	50
	F. Value Added Courses															
9.	VACC401	VACC401 NCC/Sports 0 2 0 0 2 0 2 0 2 00 00 50 00 50														
	Total									22						700

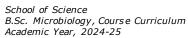


COURSE CODE	COURSE NAME	SEMESTER
BSMO411	INDUSTRIAL MICROBIOLOGY	IV

	Teaching Sci	heme (Hours)			Teachin	g Credit	
Lecture	Practical	Tutorial	Total Hours	Lecture	Practical	Tutorial	Total Credit
45	30	0	75	3	1	0	4

Course Pre-requisites	Fundamental knowledge of industrial use of microorganisms.
Course Category	Major
Course focus	Employability
Rationale	To have an overview and understanding of industrially important microorganisms. The subject also explains the different types of fermentation processes, bioreactors and production of microbial products.
Course Revision/Approval Date:	07/03/2024
Course Objectives (As per Blooms' Taxonomy)	 Remember the Basics of Industrial Microbiology and Fermentation Processes Understand Types of Bioreactors and Fermentation Parameter Measurements Analyses To study microbial classification. Analyze Techniques in Upstream and Downstream Processing Evaluate Microbial Production of Industrial Products Apply Methods for Enzyme Immobilization

Course Content (Theory)	Weightage	Contact hours
Unit 1: Introduction to industrial microbiology and Fermentation Process Brief history and developments in industrial microbiology. Types of fermentation processes - Solid-state and liquid-state (stationary and submerged) fermentations; batch, fed-batch (eg. baker's yeast) and continuous fermentations.	20%	6
Unit 2: Types of bioreactors and measurement of fermentation parameters Components of a typical bioreactor. Types of bioreactors: Laboratory, pilot-scale and production fermenters, continuously stirred tank reactor, air-lift fermenter. Measurement and control of fermentation parameters: pH, temperature, dissolved oxygen, foaming and aeration.	20%	6
Unit 3: Upstream and Downstream processing Upstream processing: Sources of industrially important microbes and their isolation, preservation and maintenance methods. Crude and synthetic media: Molasses, corn-steep liquor, sulphite waste liquor, whey, yeast extract, soybean meal, peptone and tryptone. Downstream processing: Cell disruption by physical, chemical and biological methods. Membrane filtration and ultrafiltration, centrifugation, solvent-solvent extraction, precipitation, lyophilization and spray drying.	20%	6
Unit 4: Microbial production of industrial products Microorganisms, fermentation and recovery strategies: citric acid, ethanol, glutamic acid, Vitamin B12. wine, beer, antibiotics (penicillin, streptomycin). Enzymes: amylase, protease, lipase, glucose isomerase and glucose oxidase.	20%	6





Academic Teary 202 / 25		
Unit 5: Enzyme immobilization	20%	6
Methods of immobilization, advantages and applications of immobilization, large		
scale applications of immobilized enzymes (glucose isomerase and penicillin		
acylase).		

List Of Practical	Weightage	Contact hours
1. Screening and Isolation of industrially important microorganism from	10%	4
natural resource (enzyme/ antibiotic/organic acid producer)		
2. Calculation of thermal death point (TDP) of a microbial sample.	10%	4
3. Study parts of bioreactor and its design	10%	4
4. Measure the dissolve oxygen in the fermentation broth	10%	4
5. Preparation of Inoculum (cell count of yeast cell using	10%	4
haemocytometer)		
6. Preparation of fermentation media	10%	4
7. Product recovery and Purification	10%	4
8. Antibiotic Assay (Well and Disk method)	10%	4
9. Enzyme Immobilization	10%	4
10. Production and Analysis of Ethanol	10%	4
11. Production and Analysis of amylase	10%	4

Instructional Method and Pedagogy:

Group discussion, Chalk and board, Audiovisuals, Seminars, Quiz, PPT, Demonstration, Case studies

	Course Objectives:	Blooms' Taxonomy Domain	Blooms' Taxonomy Sub Domain
CO1	Explain the Fundamentals of Industrial Microbiology and Fermentation Processes	Remember	Knowledge and Comprehension
CO2	Analyze and Classify Different Types of Bioreactors	Analyze	Apply, Practice, Interpret, Select, Correlate
CO3	Evaluate Upstream and Downstream Processing Techniques	Analyses and Evaluation	Compare, Classify, Select, Investigate
CO4	Develop Strategies for Microbial Production of Industrial Products	Create	Construct, Develop, Produce
CO5	Apply Techniques for Enzyme Immobilization and Evaluate Their Industrial Applications	Understand	Explain, Describe, outline, Predict, Summarize

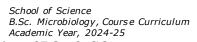
Learning Resources



1.	Reference Books:
	1. Atlas RM. (1997). Principles of Microbiology. 2nd edition. WM.T.Brown Publishers.
	 Pelczar MJ, Chan ECS and Krieg NR. (1993). Microbiology. 5th edition. McGraw Hill Book Company.
	3. Stanbury PF, Whitaker A and Hall SJ. (2006). Principles of Fermentation Technology. 2nd edition, Elsevier Science
2.	Journals & Periodicals
	1. Current Science
	2. Advances in Industrial Microbilogy
3.	Other Electronic resources: NPTEL

Evaluation Scheme	Total Marks			
Theory: Mid semester Marks	20 marks			
Theory: End Semester Marks	40 marks			
Theory: Continuous Evaluation Component Marks	Attendance	05 marks		
	MCQs Open Book Assignment Article Review	10 marks 15 marks 10 marks		
	Total	40 Marks		
Practical Marks		25		
	Attendance Practical Exam	05 marks 30 marks		
	Viva	10 marks		
	Journal	05 marks		
	Total	50 Marks		

	PSO1	PSO2	PSO3	PSO4	PSO5	PSO6
CO1	2	0	0	0	1	1
CO2	2	0	0	0	1	1
CO3	1	1	2	1	2	2
CO4	1	1	2	1	2	2
CO5	2	2	1	2	2	2





	PO1	PO2	PO3	PO4	PO5	PO6
CO1	3	2	0	0	1	2
CO2	3	2	0	0	1	2
CO3	2	2	2	1	2	2
CO4	2	2	2	1	2	2
CO5	2	2	1	2	2	1

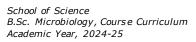


COURSE CODE	COURSE NAME	SEMESTER
BSMO412	RECOMBINANT DNA	IV
	TECHNOLOGY	
•		

Teaching Scheme (Hours)				Teaching	g Credit		
Lecture	Practical	Tutorial	Total Hours	Lecture	Practical	Tutorial	Total Credit
45	30	0	75	3	1	0	4

Course Pre-requisites	Fundamental knowledge of genes, genomes and genetic engineering.
Course Category	Professional Core Course
Course focus	Employability
Rationale	This course equips students with essential knowledge and practical skills in molecular tools, PCR, genetic engineering, protein engineering, and plant genetic engineering, preparing them for applications in biotechnology and research.
Course Revision/ Approval Date:	07/03/2024
Course Objectives (As per Blooms' Taxonomy)	Remember & Understand Molecular Tools and Applications Apply Principles of PCR and Genetic Mapping Analyses Genetic Engineering Techniques and Therapeutic Applications Create Model organisms in recombinant DNA technology
	5. Create Genetic Engineering Strategies in Plants

Course Content (Theory)	Weightage	Contact
		hours
Unit 1: Molecular tools and applications Molecular tools and applications- enzymes, ligases, polymerases, alkaline phosphatase. Episomes, Plasmids and other cloning vectors (Bacteriophage-derived vectors, artificial chromosomes), Microinjection, Electroporation, Ultrasonication.	20%	6
Unit 2: Polymersae chain reaction Principle and applications of Polymerase chain reaction (PCR), primerdesign, and RT- (Reverse transcription) PCR. Restriction and modification system, restriction mapping. Southern and Northern hybridization. Preparation and comparison of Genomic and cDNA library, screening of recombinants, reverse transcription.	20%	6





Unit 3: Genetic Engineering and Therapeutic Applications Genome mapping, DNA fingerprinting, Applications of Genetic Engineering in animals: Production and applications of transgenic mice, role of ES cells in gene targeting in mice. Therapeutic products produced by genetic engineeringblood proteins, human hormones, immune modulators and vaccines (one example each).	20%	6
Unit 4: Advanced Techniques in Protein Engineering Random and site-directed mutagenesis: Primer extension and PCR based methods of site directed mutagenesis, Random mutagenesis, Gene shuffling, production of chimeric proteins, Protein engineering concepts and examples (any two).	20%	6
Unit 5: Genetic engineering in plants Use of Agrobacterium tumefaciens and A.rhizogenes, Ti plasmids, Strategies for gene transfer to plant cells, Direct DNA transfer to plants, Gene targeting in plants, Use of plant viruses as episomal expression vectors.	20%	6

List Of Practical	Weightage	Contact hours
1. Isolation of chromosomal DNA from E. coli	20%	12
2. Plasmid DNA isolation		
3. Qualitative and quantitative analysis of DNA using spectrophotometer		
4. Demonstration of PCR	20%	12
5. Making competent cells		
6. Transformation of competent cells		
7. Restriction digestion of DNA	20%	12
8. Demonstration of Agarose gel electrophoresis.	20%	12
9. Isolation of chromosomal DNA from plant cells.	20%	12

Instructional Method and Pedagogy: Audio-Visual Lectures, Quizzes, Debates, Project works, Case studies, and Assignments. Practical exercises are designed to understand the theory as taught in classroom.



Course Objectives:	Blooms' Taxonomy Domain	Blooms' Taxonomy Sub Domain
After successful completion of the above course, students will be able to:	TT 1 1	Explain,
CO1 Understand the Principles and Applications of Molecular Tools in Genetic Engineering	Understand and Analyse	Describe, Discuss, Recall,
CO2 Apply PCR Techniques and Genetic Analysis Methods	Apply and create	Apply, Practice, Interpret, Select, Correlate
CO3 Evaluate Genetic Engineering Applications and Therapeutic Uses	evaluate	Compare, Classify, Select, Investigate
CO4 Analyze Advanced Techniques in Protein Engineering	Analyze	Construct, Develop, Produce
CO5 Apply Genetic Engineering Principles to Plant Biotechnology	Create	Explain, Describe, outline, Predict, Summarize

	Learning Resources
1.	Reference books:
	 Brown TA. (2006). Gene Cloning and DNA Analysis. 5th edition. Blackwell Publishing, Oxford, U.K. Clark DP and Pazdernik NJ. (2009). Biotechnology-Applying the Genetic Revolution. Elsevier Academic Press, USA. Glick, B.R., Pasternak, J.J. (2003). Molecular Biotechnology-Principles and Applications of recombinant DNA. ASM Press, Washington. Primrose SB and Twyman RM. (2006). Principles of Gene Manipulation and Genomics, 7th edition. Blackwell Publishing, Oxford, U.K. Sambrook J, Fritsch EF and Maniatis T. (2001). Molecular Cloning-A Laboratory Manual. 3rd edition. Cold Spring Harbor Laboratory Press.
2.	Journals & Periodicals
	6. Current Science in RDNA technology7. Advances in R-DNA Technology
3.	Other Electronic resources: NPTEL



Evaluation Scheme	Total Marks		
Theory: Mid semester Marks	20 marks		
Theory: End Semester Marks	40 marks		
Theory: Continuous Evaluation Component	Attendance	05 marks	
Marks	MCQs	10 marks	
	Open Book Assignment	15 marks	
	Article Review	10 marks	
	Total	40 Marks	
Practical Marks	·		
	Attendance	05 marks	
	Practical Exam	30 marks	
	Viva	10 marks	
	Journal	05 marks	
	Total	50 Marks	

	PSO1	PSO2	PSO3	PSO4	PSO5	PSO6
CO1	2	1	1	2	1	1
CO2	1	2	2	1	1	2
CO3	1	1	2	2	1	1
CO4	1	1	2	1	1	1
CO5	2	1	2	1	1	1

Mapping of POs & COs

mg or r or						
	PO1	PO2	PO3	PO4	PO5	PO6
CO1	3	1	2	2	2	0
CO2	2	2	2	2	1	1
CO3	1	1	1	1	1	1
CO4	1	1	1	2	1	0
CO5	2	1	2	1	2	1



COURSE CODE	COURSE NAME	SEMESTER
BSMO413	MICROBIAL PHYSIOLOGY &	IV
	METABOLISM	

Teaching Scheme (Hours) Teaching Credit							
Lecture	Practical	Tutorial	Total Hours	Lecture	Practical	Tutorial	Total Credit
45	30	0	75	3	1	0	4

Course Pre-requisites	Fundamental concepts of microbial growth, nutrition and biochemistry.			
-	Professional Core Course			
Course Category				
Course focus	Employability			
Rationale	To have an overview of microbial growth and physiological processes.			
	The subject also explains the nutrient uptake and transport mechanism in			
	microorganisms.			
Course Revision/Approval	07/03/2024			
Date:				
Course Objectives	1 Domambar To learn shout the major factures of growth and			
(As per Blooms' Taxonomy)	1. Remember To learn about the major features of growth and			
	metabolism of microorganisms, and microbial relationships.			
	2. Apply To study determination of growth curve environmental			
	influence on the microbial growth.			
	3. Analyses To study the biology and evolution of			
	1			
	microorganisms and their interactions with the environment.			
	4. Create To learn about and primary and secondary metabolism.			
	5. Understand To explore about energy source for microorganisms			
	and relationship between metabolism and energy source.			
	and relationship between metabolism and energy source.			

Course Content (Theory)	Weightage	Contact
		hours
Unit 1: Microbial Growth and Effect of Environment on Microbial Growth.	20%	6
Definitions of growth, measurement of microbial growth, Batch culture,		
Continuous culture, generation time and specific growth rate, synchronous		
growth, diauxic growth curve. Microbial growth in response to environmental		
conditions (Temperature, pH, solute and water activity, Oxygen and pressure).		
Microbial growth in response to nutrition and energy sources.		
Unit 2: Nutrient uptake and Transport	20%	6
Passive and facilitated diffusion. Primary and secondary active transport,		
concept of uniport, symport and antiport. Group translocation, Iron uptake.		



Unit 3: Chemoheterotrophic Metabolism- Aerobic Respiration.	25%	8
Concept of aerobic respiration, Concept of anaerobic respiration and		
fermentation. Sugar degradation pathways. Electron transport chain: components		
of respiratory chain, comparison of mitochondrial and bacterial ETC, electron		
transport phosphorylation, uncouplers and inhibitors.		
Anaerobic respiration with special reference to dissimilatory nitrate reduction		
(Denitrification; nitrate/Nitrite and nitrate/ammonia respiration; fermentative		
nitrate reduction). Fermentation - Alcohol fermentation and Pasteur effect;		
Lactate fermentation (homofermentative and heterofermentative pathways),		
concept of linear and branched fermentation pathways		
Unit 4: Chemoheterotrophic Metabolism- Anaerobic respiration	20%	6
Unit 4: Chemoheterotrophic Metabolism- Anaerobic respiration and fermentation	20%	6
<u>-</u>	20%	6
and fermentation	20%	6
and fermentation Introduction to aerobic and anaerobic chemolithotrophy with an example each.	20%	6
and fermentation Introduction to aerobic and anaerobic chemolithotrophy with an example each. Hydrogen oxidation (definition and reaction) and methanogenesis (definition and	20%	6
and fermentation Introduction to aerobic and anaerobic chemolithotrophy with an example each. Hydrogen oxidation (definition and reaction) and methanogenesis (definition and reaction) Introduction to phototrophic metabolism - groups of phototrophic	20%	6
and fermentation Introduction to aerobic and anaerobic chemolithotrophy with an example each. Hydrogen oxidation (definition and reaction) and methanogenesis (definition and reaction) Introduction to phototrophic metabolism - groups of phototrophic microorganisms, anoxygenic vs. Oxygenic photosynthesis with reference to	20%	6
and fermentation Introduction to aerobic and anaerobic chemolithotrophy with an example each. Hydrogen oxidation (definition and reaction) and methanogenesis (definition and reaction) Introduction to phototrophic metabolism - groups of phototrophic microorganisms, anoxygenic vs. Oxygenic photosynthesis with reference to photosynthesis in green bacteria, purple bacteria and cyanobacteria.		,

List Of Practical	Weightage	Contact hours
1. Study and plot the growth curve of E. coli by turbidometry and	10%	4
standard plate count methods.		
2. Calculations of generation time and specific growth rate of bacteria	10%	4
from the graph plotted with the given data.		
3. Diauxic growth curve of E. Coli (Lac Operon)	10%	4
4. Effect of desiccation/drying on the growth of bacteria	10%	4
5. Effect of temperature on growth of E. coli and S. aureus	10%	4
6. Effect of pH on growth of E. coli and S. aureus	10%	4
7. Effect of Temperature, pH, concentration of salt (NaCl) and Sugar	10%	4
(glucose) on growth of E. coli and S. aureus (Turbidimetry)		
8. Effect of carbon and nitrogen sources on growth of E. coli.	10%	4
9. Effect of salt on growth of E. coli.	10%	4
10. Single Enzyme test: (Catalase test, Coagulase test, Oxidase test,	10%	4
Indole test, ONPG (O-nitrophenyl-beta-D-galactopyranoside), Urease		
11. Metabolic Pathway:	10%	4
Carbohydrate fermentation/ Oxidation- fermentation test (Oxidative fermentative -medium -CDC test, -Carbohydrate		
fermentation on TSI -MR/VP test		



12. Decarboxylase/Di hydrolase test	
Deaminase reaction test (Phenylalanine agar)	
Lysine decarboxylase test	
Citrate Utilization (Simmons citrate agar)	
Nitrate reduction test	
Gelatin liquification test	

Instructional Method and Pedagogy:

Audio-Visual Lectures, Quizzes, Debates, Project works, Case studies, and Assignments. Practical exercises are designed to understand the theory as taught in classroom.

	Course Objectives:	Blooms' Taxonomy Domain	Blooms' Taxonomy Sub Domain
CO1	Describing the growth characteristics of the microorganisms capable of growing under unusual environmental condition of temperature, oxygen, and solute and water activity	Remember	Explain, Describe, Discuss, Recall, Locate
CO2	Describing the growth characteristics of the microorganisms capable of growing under unusual environmental condition of temperature, oxygen, and solute and water activity	Apply	Apply, Practice, Interpret, Select, Correlate
CO3	Differentiating concepts of aerobic and anaerobic respiration and how these are manifested in the form of different metabolic pathways in microorganisms	Analyses and Evaluation	Compare, Classify, Select, Investigate
CO4	Differentiating concepts of aerobic and anaerobic respiration and how these are manifested in the form of different metabolic pathways in microorganisms	Create	Construct, Develop, Produce
CO5	Describing the growth characteristics of the microorganisms which require different nutrient for growth and the associated mechanisms of energy generation for their survival like autotrophs, heterotrophs, chemolithoautotrophs etc	Understand	Explain, Describe, outline, Predict, Summarize

Learning Resources				
1.	Reference books:			
	1. Gottschalk G. (1986). Bacterial Metabolism. 2nd edition. Springer Verlag			
	2. Stanier RY, Ingrahm JI, Wheelis ML and Painter PR. (1987). General			
	Microbiology. 5th edition, McMillan Press.			
	3. Willey JM, Sherwood LM, and Woolverton CJ. (2013). Prescott's Microbiology.			
	9th edition. McGraw Hill Higher Education.			
2.	Journals & Periodicals			
	1. Current Science			
	2. Advances in Microbiology			
3.	Other Electronic resources: NPTEL			



School of Science B.Sc. Microbiology, Course Curriculum Academic Year, 2024-25

Evaluation Scheme	Total Marks	
Theory: Mid semester Marks	20 marks	
Theory: End Semester Marks	40 marks	
Theory: Continuous Evaluation Component Marks	Attendance MCQs Open Book Assignment Article Review Total	05 marks 10 marks 15 marks 10 marks 40 Marks
Practical Marks	Attendance Practical Exam Viva Journal Total	05 marks 30 marks 10 marks 05 marks 50 Marks

Mapping of PSOs & COs

	PSO1	PSO2	PSO3	PSO4	PSO5	PSO6
CO1	2	2	2	0	2	2
CO2	2	2	2	0	2	2
CO3	2	2	0	1	2	2
CO4	2	2	0	1	2	2
CO5	2	2	2	1	2	2

Mapping of POs & COs

mg orror a con						
	PO1	PO2	PO3	PO4	PO5	PO6
CO1	3	2	2	0	2	0
CO2	3	2	2	0	2	0
CO3	2	2	0	1	2	2
CO4	2	2	0	1	2	2
CO5	3	3	2	1	2	2

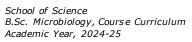


COURSE CODE	COURSE NAME	SEMESTER
BSCM415	CHEMISTRY II	IV

Teaching Scheme (Hours)				Teachin	g Credit			
	Lecture	Practical	Tutorial	Total Hours	Lecture	Practical	Tutorial	Total Credit
	30	30	00	60	2	1	00	3

Course Pre-requisites	Basic knowledge of physical and organic chemistry
Course Category	Generic elective
Course focus	Employability
Rationale	Fusion of various sciences is occurring at a rapid pace. Biochemists and microbiologists need indepth knowledge in Chemistry. To fill this gap the course, namely, Chemistry II, is inducted into the curriculum.
Course Revision/ Approval Date:	14/03/2023
Course Objectives (As per Blooms' Taxonomy)	 To impart the knowledge of solution state and the laws governing thereof. To equip the students with the indepth knowledge of Phase equilibria and their industrial relevance To make the students skilled in operating electroanalytical devices, like, pH meter, potentiometer and conductometery by imparting fundamental knowledge of electrochemistry To impart knowledge pertaining to carboxylic acids, Amines and diazonium Salts. To equip the students with knowledge of amino acids, peptides and proteins and the properties and conversions thereof.

Course Content (Theory)	Weightage %	Contact hours
Unit 1: Solutions Thermodynamics of ideal solutions Ideal solutions and Raoult's law, deviations from Raoult's law – nonideal solutions. Vapour pressure-composition and temperature composition curves of ideal and non-ideal solutions. Distillation of solutions. Lever rule. Azeotropes. Partial miscibility of liquids: Critical solution temperature; effect of impurity on partial miscibility of liquids. Immiscibility of liquids- Principle of steam distillation. Nernst distribution law and its applications, solvent extraction.	20	6





Unit 2: Phase Equilibrium	20	6
Phases, components and degrees of freedom of a system, criteria	-	Ŭ
of phase equilibrium. Gibbs Phase Rule and its thermodynamic		
derivation. Derivation of Clapeyron- Clausius equation and its		
importance in phase equilibria. Phase diagrams of one-component		
systems and two component systems		
Unit 3: Conductance & Conductivity		
Introduction, Equivalent and molar conductivity and their		
variation with dilution for weak and strong electrolytes. Kohlrausch		
law of independent migration of ions. Transference number and its		
experimental determination using Hittorf and Moving boundary		
methods. Ionic mobility. Applications of conductance	20	6
measurements: determination of degree of ionization of weak		Ü
electrolyte, solubility and solubility products of sparingly soluble		
salts, ionic product of water, hydrolysis constant of a salt.		
Conductometric titrations.		
Unit 4: Carboxylic acids and their derivatives (aliphatic and		
aromatic)		
Preparation: Acidic and Alkaline hydrolysis of esters. Reactions		
Hell – Volhard - Zelinsky Reaction. Acid chlorides, Anhydrides,		
Esters and Amides from acids and their interconversion.		
Reactions: Comparative study of nucleophilicity of acy		
derivatives. Reformatsky Reaction, Perkin condensation.		
Amines and Diazonium Salts Amines (Aliphatic and		
Aromatic): (Upto 5 carbons) Preparation: from alkyl halides,		
Gabriel's Phthalimide synthesis, Hofmann Bromamide reaction.		
Reactions: Hofmann vs. Saytzeff elimination, Carbylamine test,	20	10
Hinsberg test, with HNO ₂ , Schotten – Baumann Reaction.	20	12
Electrophilic substitution (case aniline): nitration, bromination,		
sulphonation. Diazonium salts: Preparation: from aromatic		
amines. Reactions: conversion to benzene, phenol, dyes.		
Unit 5: Amino Acids, Peptides and Proteins		
Ome 3. Animo Acius, reputies and riotenis	20	12
	20	14



Preparation of Amino Acids: Strecker synthesis using Gabriel's phthalimide synthesis. Isoelectric point and Electrophoresis.
Reactions of Amino acids: ester of –COOH group, acetylation of – NH2 group, complexation with Cu2+ ions, ninhydrin test.
Overview of Primary, Secondary, Tertiary and Quaternary
Structure of proteins. Determination of Primary structure of
Peptides by N-terminal and C-terminal. Carbohydrates:
Classification, and General Properties, Glucose and Fructose (open chain and cyclic structure), Determination of configuration of monosaccharides, absolute configuration of Glucose and Fructose, ascending and descending in monosaccharides. Structure of disacharrides and polysacharrides.

List of Practicals	Weightage	Contact hours
1. To determine the relative viscosity of ethyl acetate with water using Ostwald's viscometer	15	4
2. To determine the Critical Micellar Concentration (CMC) of a given surfactant Sodium Dodecyl Sulphate (SDS) using conductivity method	15	4
3. To study the effect of temperature on rate of reaction between hypo solution and HCl	15	4
4. To prepare sodium tris-oxalato ferrate (III)	15	4
5. Preparation of ammonium nickel (II) sulfate hexahydrate, (NH ₄) ₂ Ni(SO ₄) ₂ . 6H ₂ O	15	4
6. To prepare hexakis thiourea plumbus (II) nitrate hexahydate	15	4
7. To prepare tetraamine copper sulphate	10	4

Instructional Method and Pedagogy:

Classroom lecture, discussion, question and answer method, Case studies, quizes, presentations, role play, expert lecture(Consultant)



Course outcomes:	Blooms' Taxonomy Domain	Blooms' Taxono my Sub Domain
After successful completion of the above course, students will be able to:		
CO1: Understand the laws governing the solution state and apply the same for practical utility	Understand and apply	
CO2: Understand phase rule, phase equilibria, phase diagrams and their industrial utility	Understand and apply	
CO3: Understand and apply the concepts of electrochemistry	Understand and apply	
CO4: Synthesize and convert amino acids, diazonium sales and carboxylic acids	Create	
CO5: Synthesize amino acids, peptides and proteins and study their properties and conversions thereof.	Create	

	Learning resources
1	Reference books:
	Samuel Glasstone, Elements of Physical Chemistry
	Arun Bhal, BS Bhal, GD Tuli, Essentials of Physical Chemistry
	Puri, Sharma, Pathania, Principles of Physical Chemistry
	G. M. Barrow: Physical chemistry Tata McGraw-Hill (2007).
	Nelson, D. L. & Cox, M. M. Lehninger's Principles of Biochemistry 7th Ed., W.H.
	G. W. Castellan: Physical Chemistry 4th Ed. Narosa (2004).
	J. C. Kotz, P. M. Treichel, J. R. Townsend, General Chemistry, Cengage Learning India Pvt. Ltd.: New Delhi (2009).
	B. H. Mahan: University Chemistry, 3rd Edn. Narosa (1998).
	R. H. Petrucci, General Chemistry, 5th Edn., Macmillan Publishing Co.: New York (1985).



	Morrison, R. T. & Boyd, R. N. Organic Chemistry, Dorling Kindersley (India) Pvt. Ltd. (Pearson Education).
	Finar, I. L. Organic Chemistry (Volume 1), Dorling Kindersley (India) Pvt. Ltd. (Pearson Education).
	Finar, I. L. Organic Chemistry (Volume 2), Dorling Kindersley (India) Pvt. Ltd. (Pearson Education).
2	Journals & Periodicals: Journal of Chemical Sciences Energy and Environmental Science Journal of Chemical Educaiton
3	Other Electronic Resources: NPTEL, SWAYAM, MERLOT (Links available in GSFC University Link

Evaluation Scheme	Total Marks	,
Theory: Mid semester Marks	20 marks	
Theory: End Semester Marks	40 marks	
Theory: Continuous Evaluation Component Marks	Attendance	05 marks
Warks	MCQs	10 marks
	Open Book Assignment	15 marks
	Article Review	10 marks
	Total	40 Marks
Practical Marks		
	Attendance	05 marks
	Practical Exam	30 marks
	Viva	10 marks
	Journal	05 marks
	Total	50 Marks

	PSO1	PSO2	PSO3	PSO4	PSO5	PSO6
CO1	2	3	2	1	1	3
CO2	1	2	1	1	1	1
CO3	3	3	3	1	1	3
CO4	3	3	3	1	1	3
CO5	3	3	3	1	1	3

Mapping of POs & COs





	PO1	PO2	PO3	PO4	PO5	PO6
CO1	2	3	2	1	1	3
CO2	1	2	1	1	1	1
CO3	3	3	3	1	1	3
CO4	3	3	3	1	1	3
CO5	3	3	3	1	1	3

1: Slight (low); 2: Moderate (Medium); 3: Substantial (High); 0 None

School of Science B.Sc. Microbiology, Course Curriculum Academic Year, 2024-25



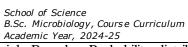
COURSE CODE	COURSE NAME	SEMESTER
BSMA415	BIOSTATISTICS	IV

Teaching Scheme (Hours)				Teachin	g Credit		
Lecture	Practical	Tutorial	Total Hours	Lecture	Practical	Tutorial	Total Credit
30	0	15	45	2	0	1	3

Course Pre-requisites	Students should have basic knowledge of Calculus & Algebra.		
Course Category	Core course		
Course focus	Skill development		
	1		
Rationale	Biostatistics is the application of statistical methods to biological, health, and medical data. It involves collecting, analysing, and interpreting data to draw meaningful conclusions in fields like epidemiology, genetics, and public health. Biostatisticians design experiments, develop models, and utilize advanced statistical techniques to uncover patterns, trends, and relationships within biological systems. They play a crucial role in medical research, clinical trials, and public health interventions by providing insights into disease prevention, treatment effectiveness, and healthcare policies.		
Course Revision/Approval	07/03/2024		
Date:			
Course Objectives	To enable the student to:		
(As per Blooms' Taxonomy)	1 Remember: Use discrete and continuous probability distributions,		
	including requirements, mean and variance, and making decisions.		
	2 Apply: Use Poisson, exponential distributions to solve statistical problems.		
	3 Understand, Apply: Identify the type of statistical situation to which		
	different distributions can be applied.		
	4 Understand: Define and distinguish between population parameters		
	and sample statistics.		
	5 Understand , Apply: Explain what is meant by statistical inference		

Course Content (Theory)	Weightage	Contact
		hours
Unit 1: Concept of sampling, Introduction to Probability, basic results and	20%	6
theorems of Probability.		
Unit 2: Concept of random variable, Probability distribution of random	20%	6
variable, Expectation and variance.		
Unit 3: Study of Binomial, Poisson and Normal Distribution, Application	20%	6
of this distribution in Bio –Sciences.		
Unit 4: Basic principles of statistical inference, Point estimation, Internal	20%	6
estimation, Statistical Hypothesis framing.		
Unit 5: Test of Significance, p- value, t - test, F - test, chi - square test,	20%	6
ANOVA etc.		

List Of Practical	Weightage	Contact
		hours
1: Tutorials Based on Basic Probability.	20%	3





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2: Tutorials Based on Probability distribution and random variables.	20%	3
3: Tutorials based on Binomial, Poisson and Normal Distribution.	20%	3
4: Tutorials based on Hypothesis testing.	20%	3
5: Tutorials based on Tests of Significance	20%	3

Instructional Method and Pedagogy: (Max. 100 words) Chalk-board, PPT, Computer Based learning, Use of M.S. Excel and Geogebra Toolbox.

Course outcome:	Blooms' Taxonomy Domain	Blooms' Taxonomy Sub Domain
After successful completion of the above course, students will be		
able to:		
CO1: Understand: Define and distinguish between populations		
and samples		
CO2: Apply: Compute a sample and polulation mean, sample		
variance, and sample standard deviation		
CO3: Remember, Understand: Know the practical issues arising		
in sampling studies.		
CO4: Apply, Analyse: Appropriately interpret results of analysis		
of variance tests.		
CO5: Analyse: Analyse statistical data using MS-Excel.		

Learning Res	sources
1.	Reference Books:
	1. Danial W (2004) Biostatistics: A foundation for Analysis in Health Sciences,
	John Wiley and Sons Inc.
	2. Fundamentals of Mathematical Statistics by S C Gupta & V K Kapoor, Sultan
	Chand & Sons, New Delhi 2009.
	3. Le CT (2003) Introductory biostatistics. 1st edition, John Wiley, USA
	4. Glaser AN (2001) High Yield TM Biostatistics. Lippincott Williams and Wilkins,
	USA.
2.	Journals & Periodicals:
	Style: name of the journal, volume (issue number), range of pages, and year.
3.	Other Electronic Resources:
	Geometry and Algebra: Geogebra.org/Calculator
	MATLAB : Mathworks.com/
	https://www.tutorialspoint.com/matlab/matlab_syntax.htm





Evaluation Scheme	Total Marks	
Theory: Mid semester Marks	20 marks	
Theory: End Semester Marks	40 marks	
Theory: Continuous Evaluation Component Marks	Attendance MCQs Open Book Assignment Open Book Assignment Total	05 marks 10 marks 15 marks 10 marks 40 Marks
Practical Marks	Attendance Practical Exam Viva Journal Discipline Total	05 marks 20 marks 10 marks 10 marks 50 Marks

1.1.mpp								
	PSO1	PSO2	PSO3	PSO4	PSO5	PSO6		
CO1	1	2	2	1	0	0		
CO2	1	2	0	1	0	0		
CO3	2	2	2	1	0	0		
CO4	1	3	2	0	0	0		
CO5	0	1	2	0	0	0		

Mapping of POs & COs

Trupping	PO1	PO2	PO3	PO4	PO5	PO6
CO1	2	2	2	0	1	0
CO2	1	2	2	0	1	0
CO3	2	1	1	1	1	0
CO4	1	2	1	0	0	0
CO5	0	0	1	0	0	0

School of Science B.Sc. Microbiology, Course Curriculum Academic Year, 2024-25

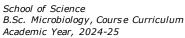


COURSE CODE
BSBO305
COURSE NAME
Biophysics
III

Teaching Scheme (Hours)					Teaching	g Credit	
Lecture	Practical	Tutorial	Total Hours	Lecture	Practical	Tutorial	Total Credit
30	0	0	30	2	0	0	2

Course Pre-requisites	Understanding of basic sciences up to school level (10+2 level).				
Course Category	Mandatory courses				
Course focus	Employability				
Rationale It equips biotechnologists with a strong foundation in the phenderlying biological systems, enabling them to contribute innovation, and technological advancements in biotechnology both and international levels. It is important in understanding biological molecular modeling, biophysical techniques, biomolecular instrumentation, international collaboration, and emerging techniques.					
Course Revision/ Approval Date:	14/03/2020				
Course Objectives	Course Objectives: Students will be able to:				
(As per Blooms' Taxonomy)	1: To provide introductory knowledge of Biophysics to the students				
	2: To familiarize students with relevant applications required for study of biological systems 3: To enable the students to address any elementary thermodynamic problems in biological systems 4: To familiarize the student with chemical thermodynamics of biological systems 5: To familiarize students with modern developments in the area of biomaterials				

Course Content (Theory)	Weightage	Contact
		hours
Unit 1: Building Blocks & Structure of Living State:	20%	6
Atoms and ions, molecules essential for life, what is life. Living state interactions: Forces		
and molecular bonds, electric & thermal interactions, electric dipoles, casimir		
interactions, domains of physics in biology.		
Unit 2: Heat Transfer in biomaterials:	20%	6
Heat TransferMechanism, The Heat equation, Joule heating of tissue. Living State		
Thermodynamics: Thermodynamic equilibrium.		
Unit 3: Thermodynamics:	20%	6
First law of thermodynamics and conservation of energy. Entropy and the second law of		
thermodynamics, Physics of many particle systems, two state systems, continuous energy		
distribution, Composite systems, Casimir contribution of free energy, Protein folding and		
unfolding.		





Academic rear, 2024 25		
Unit 4: Thermodynamics:	20%	6
Open systems and chemical thermodynamics: Enthalpy, Helmholtz Free energy, Gibbs		
Free Energy and chemical potential.		
Unit 5: Thermodynamics:	20%	6
Activation energy and rate constants, enzymatic reactions, ATP hydrolysis & synthesis,		
Entropy of mixing, The grand canonical ensemble, Haemoglobin.s		

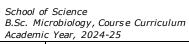
List Of Practical	Weightage	Contact hours
Based on theory		

Instructional Method and Pedagogy: (Max. 100 words)
Audiovisuals, Seminars, Quiz, PPT, Demonstration, Case studies

Course Outcome:	Blooms' Taxonomy Domain	Blooms' Taxonomy Sub Domain
After successful completion of the above course, students will be able to:		
CO1:Students will have sufficient knowledge of Biophysics for undergraduate studies		Describe
	Understand	
CO2: Students will be familiar with concepts that help them prepare	Understand and	Describe
for modern courses like Bioinstrumentation.	Create	
CO3: Students will be able to understand and appreciate the	Evaluate and	Describe and
interdisciplinary nature of the modern researches	Analyse	Explain
CO4: Students will be able to prepare working models on Biophysical	Remember, Apply	Describeand classify
systems	and Create	
CO5: Students will be able to continue learning through various e-	Understand and	Classify and Explain
resources	apply	

Learning Resources 1. Reference Books:

- I. Introductory Biophysics, J. Claycomb, JQP Tran, Jones & Bartlett Publishers.
- II. Aspects of Biophysics, Hughe S W, John Willy and Sons.
- III. Essentials of Biophysics by P Narayanan, New Age International.





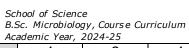
2.	Journals & Periodicals: 1. Journal of Young Investigators (JYI) 2. Biophysics – Frontiers
3.	Other Electronic Resources: 1. For detailed further study: Physics of Biological systems (https://onlinecourses.nptel.ac.in/noc20_ph02/preview)

Evaluation Scheme	Total Marks					
Theory: Mid semester Marks	20 marks					
Theory: End Semester Marks	40 marks					
Theory: Continuous						
Evaluation Component Marks	Attendance	05 marks				
	MCQs	10 marks				
	Open Book Assignment	15 marks				
	Research Paper Review	10 marks				
	Total	40 Marks				
Practical Marks						
	Attendance	05 marks				
	Practical Exam	30 marks				
	Viva	10 marks				
	Journal	05 marks				
	Total	50 Marks				

wrapping	wrapping of 1 50s & COs								
	PSO1	PSO2	PSO3	PSO4	PSO5	PSO6			
CO1	2	1	-	2	-	1			
CO2	1	1	1	1	1	1			
CO3	1	2	1	1	1	1			
CO4	1	2	1	1	-	1			
CO5	1	-	-	1	-	-			

Mapping of POs & COs

Wrapping of Fos & Cos								
	PO1	PO2	PO3	PO4	PO5	PO6		
CO1	2	2	-	2	1	-		
CO2	1	1	1	2	1	-		



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CO3	1	2	1	1	1	1
CO4	1	1	-	2	1	-
CO5	1	-	-	1	-	-



COURSE CODE	COURSE NAME	SEMESTER
AECC401	ENVIRONMENTAL SCIENCE	IV

Teaching Scheme (Hours) Teaching			Teaching	Credit			
Lecture	Practical	Tutorial	Total Hours	Lecture Practical Tutorial Total			Total Credit
30	0	0	30	2	0	0	2

Course Pre-	10 +2 (With Arts/Science/Commerce)
requisites	
Course Category	Ability Enhancement Compulsory Course.
Course focus	Employability
Rationale	The fundamental concepts of environmental studies help in understanding the ecosystem and biogeochemical cycle that connects humans with their biosphere. Moreover, understanding pollution & treatment to treat a variety of pollution will enhance problem-solving skills of the students.
Course	07/03/2024
Revision/	
Approval Date:	
Course Objectives	To enable the student to:
(As per Blooms'	1. Remember: To acquire an awareness of and sensitivity to the total
Taxonomy)	environment and its allied problems.
•	2. Understand: To make educated judgments about environmental issues.
	3. Apply: Develop skills and a commitment to act independently and collectively to environmental sustainability
	4. Analyse: Students can able to debate environmental science with use of appropriate scientific information
	5. Create: Engaging with students of all disciplines to think critically, ethically, and creatively when evaluating environmental issues.

Course Content (Theory)	Weightage	Contact hours
Unit 1: Introduction of Ecology Ecology-Objectives and Classification Concepts of an ecosystem-structure & function of ecosystem components of ecosystem, Hydrological cycle, carbon cycle, oxygen cycle, Nitrogen cycle, Sulphur cycle	20%	6
Unit 2: Unit 2: Ecological pyramids of various ecosystems Forest Ecosystem, Grassland Ecosystem, Desert Ecosystem, Aquatic ecosystem, Estuarine Ecosystem.	20%	6
Unit 3: Air pollution and its control Introduction, Classification of air pollutants, air pollutants and their effects, acid rain, photochemical smog, particulates. Characteristics and biochemical effects of some important air pollutants, Effect of air pollutants on man and environment, Air quality standard, air monitoring and control of air pollution	20%	6



Unit 4: Water pollution and its control Introduction, Classification of water pollutants, physical, chemical and biological characteristics of waste water, waste water treatment: Primary treatment- Sedimentation, coagulation, equalization, neutralization, secondary treatment-aerobic treatment-aerated lagoons, trickling filter, activated sludge process, oxidation ditch process, oxidation pond, anaerobic treatment-anaerobic sludge digestion, sludge treatment and disposal and tertiary treatment-evaporation, ion exchange, adsorption, chemical precipitation, Electrodialysis, reverse osmosis.	20%	6
Unit 5: Solid and hazardous waste: pollution, treatment and disposal Introduction, Classification and origin, characteristics of solid wastes, objectives and considerations in solid waste management, methods of solid waste treatment and disposal - composting, land filling, thermal processes- incineration, pyrolysis, recycling and reuse of solid waste-co-disposal, bioconversion.	20%	6

Instructional Methodand Pedagogy:

Utilizing models, PowerPoint Presentations, group discussions, and seminars are some of the methods adopted to improve the student's ability to grasp the principles of environmental studies.

Course Outcome:	Blooms' Taxonomy Domain	Blooms' Taxonomy Subdomain	
After successful completion of the above course, students will be	able to:		
CO1: Skills for identifying environmental problems: Evaluate information from popular electronic and print media	Understand & remember	Define, Classify & Describe	
CO2: Interdisciplinary - When encountering environmental problems students will assess necessary scientific concepts and data, consider likely social dynamics, and establish integ cultural contexts	C III C I S tell I C,	Define, Classify, Describe, Demonstrate & & Examine	
CO3: Communication - Students will communicate with precision, effective art, and sound rhetoric in writing, in speech, and in digital media	Understand, Remember & Apply	Define, Classify, Describe & Demonstrate	
CO4: Research - When faced with questions that lie	Understand,	Define, Classify,	



beyond their current knowledge base, students will actively research data, concepts, histories, and narratives necessary for adequate consideration of the issue.	Remember, Apply & Analyse	Describe, Demonstrate & Examine
CO5: Intellectual Flexibility - Students will possess the intellectual flexibility necessary to view environmental questions from multiple perspectives, prepared to alter their understanding as they learn new ways of understanding.	Understand, Remember, Analyse & Apply	Define, Describe & Demonstrate

Learning Resources

- 1. Reference/Text Books:
 - 1. Fundamentals of Ecology by EP Odum Cengage
 - 2. Big Questions in Ecology & Evolution by TN Sherratt & DM Wilkinson, Oxford.
 - 3. Ecology: Experimental Analysis of Distribution & Abudance by CJ Krebs, Pearson Education, London
 - 4. Concept of Ecology by EJ Kormondy, Pearson Education, London
 - 5. Conservation Biology: Voices from the Tropics. Bys Sodhi, N.S., Gibson, L. & Raven, P.H. (eds) John Wiley & Sons
 - 6. Plastic and Environment by RE Hester and RM Harrison, Royal Society of Chemistry, Thomas Graham House, Science Park, Milton Road, Cambridge, CB4 0WF, UK
 - 7. Fundamental concepts in Environmental studies by DD Mishra, S. Chand Publishing, India
 - 8. Cell Biology, Genetics, Molecular Biology, Evolution and Ecology by PS Verma and VK Agarwal, S. Chand Publication, India
 - 9. Fundamentals of Ecology by PD Sharma, Rastogi Publications
- 2. Journals & Periodicals:
 - 1. Environmental Pollutants and Bioavailability
 - 2. Clean Air Journal
 - 3. Emerging Contaminants
 - 4. Environment: Science and Policy for Sustainable Development
 - 5. Annual Review of Environment and Resources
 - 6. Renewable Energy
 - 7. Renewable & Sustainable Energy Reviews
- 3. Other Electronic Resources:
 - 1. Green.tv—supported by UNEP—broadband TV channel for films about environmental issues.
 - 2. Climate Change TV—funded by companies, governments and organisations, and produced by the magazine Responding to Climate Change—the world's first web channel specific to climate change videos.
 - 3. Terra: The Nature of Our World video podcast produced in conjunction with the Master of Fine Arts program in Science & Natural History Filmmaking at Montana State University, Filmmakers for Conservation, and PBS—weeklyvideo show about science and natural history.

Evaluation Scheme	Total Marks
Theory: Midsemester Marks	20 marks
Theory: End Semester Marks	40 marks



Theory: Continuous Evaluation Component Marks	Attendance	05 marks
	MCQs	10 marks
	Open Book Assignment	15 marks
	Article Review	10 marks
	Total	40 Marks

	PSO1	PSO2	PSO3	PSO4	PSO5	PSO6
CO1	0	1	0	3	0	0
CO2	0	3	0	3	0	1
CO3	0	2	0	3	0	0
CO4	0	2	0	3	0	0
CO5	0	2	0	3	0	1

Mapping of POs & COs

	PO1	PO2	PO3	PO4	PO5	PO6			
CO1	0	0	0	2	2	2			
CO2	0	1	1	3	3	3			
CO3	0	1	0	1	1	1			
CO4	0	0	0	2	2	2			
CO5	0	0	0	3	3	3			

^{1:} Slight (low); 2: Moderate (Medium); 3: Substantial (High); 0 None



Academic	Year, 2024-25			THE RE			
		Semester – V					
Sr. No.	Course Code	Course Title	L	T	P	С	Mark s
		A. Ability Enhanceme nt Compulsory	Course				
1 AECC501 Disaster Risk Management 2 0				0	0	2	100
		B. Skill Enhancement Courses					
		a) Skill Enhancement compulsory co	ourses				
2	SECC504	Internship	0	0	2	2	50
1		C. Core Course	•				
3	BSMB501	Virology	4	0	2	6	150
4	BSMB502	Medical Microbiology	4	0	2	6	150
		D. Discipline Specific Generic Electives (Any Tv	wo)			
5	BSMB503	Microbial Biotechnology	4	0	2	6	150
6	BSMB504	Research Methodology	4	0	2	6	150
7	BSMB505	Bioanalytical tools	4	0	2	6	150
8	BSMB506	Microbes in SustainableAgriculture and Development	4	0	2	6	150
					Total	28	750



Teaching Scheme Semester-V

Sr. Commo Codo (Hours/week)				Tea	ching C	redit			Evaluation	n Scheme						
No ·	No Course Code Course Name		L	P	Т	Total	L	P	Т	Total	Theory: MS Marks	Theory: CEC Marks	Theory: ES Marks	Theory Marks	Practical Marks	Total Marks
				•	A. Abil	ity Enha	ncemen	t Comp	ulsory (Course						
1	BSEN501	Disaster Risk Management	2	0	0	2	2	0	0	2	20	40	40	100	00	100
						B. Skil	l Enhan	cement (Courses							
					a) Sl	kill Enha	ncemen	t compu	lsory co	urses						
2	SECC504	Internship	0	2	0	2	0	0	0	0	0	0	0	0	50	50
						(C. Core	Course								
3	BSMB501	Virology	4	4	0	8	4	2	0	6	20	40	40	100	50	150
4	BSMB502	Medical Microbiology	4	4	0	8	4	2	0	6	20	40	40	100	50	150
					D. Disci	pline Spe	cific Ger	neric Ele	ctives (A	Any Two))				1	
5	BSMB503	Microbial Biotechnology	4	4	0	8	4	2	0	6	20	40	40	100	50	150
6	BSMB504	Research Methodology	4	4	0	8	4	2	0	6	20	40	40	100	50	150
7	BSMB505	Bioanalytical tools	4	4	0	8	4	2	0	6	20	40	40	100	50	150
8	BSMB506	Microbes in SustainableAgriculture and Development	4	4	0	8	4	2	0	6	20	40	40	100	50	150
		Total								28						750

Note: L = Lecture, P = Practice, T = Tutorial, MS - Mid Semester, CEC - Continuous Evaluation Component, ES - End Semester



COURSE CODE	COURSE NAME	SEMESTER
BSEN501	DISASTER RISK	V
	MANAGEMENT	

Teaching Scheme (Hours)					Teac Cre		
Lecture	Practical	Tutorial	Total Hour	Lecture	Practical	Tutorial	Total Credit
30	0	0	s 30	2	0	0	2
30	U	U	30	2	U	U	2

Course Pre-requisites	Students should have basic knowledge about Immunity.
Course Category	Ability Enhancement Compulsory Course
Course focus	Employability
Rationale	To have an overview of inter-relationship between disaster and development and various disaster management frameworks and strategies.
Course Revision/ Approval Date:	14/03/2020
Course Objectives (As per Blooms' Taxonomy)	 Remember To introduce inter-relationship between disaster and development. Apply To introduce types of disasters with case studies and create awareness. Analyses To study the effective use of science for mitigating disasters Create To study case study of various famous disasters. Understand To introduce various disaster management frameworks and strategies adopted at national and international levels.

Course Content (Theory)	Weightage	Contact
		hours
Unit 1: Understanding the Concepts and Definitions of Disaster, Hazard, Vulnerability Risk, Capacity Disaster and Development, and Disaster Management Fundamental of Disasters-Types, Trends, Causes, Consequences and Control: Geological Disasters, Hydro-Meteorological Disasters, Biological Disasters, Technological Disasters, and Man-made Disasters. Global Disaster Trends – Emerging Risks of Disasters – Climate Change and Urban Disasters.	20%	6



Unit 2: Disaster Management Cycle and Framework-Disaster Management Cycle — Paradigm Shift in Disaster Management, Pre-Disaster — Risk Assessment and Analysis, Risk Mapping, Zonation, Micro zonation, Prevention and Mitigation of Disasters, Early Warning System, Preparedness, Capacity Development; Awareness, During Disaster — Evacuation — Disaster Communication — Search and Rescue, Emergency Operation Centre— Incident Command System — Relief And Rehabilitation. Post -disaster Damage and Needs Assessment, Restoration of Critical Infrastructure — Early Recovery — Reconstruction and Redevelopment; IDNDR, Yokohama Strategy, Hyogo Framework of Action, Sendai framework.	20%	6
Unit 3: Disaster Management in India Disaster Profile of India – Mega Disasters of India and Lessons Learnt, Disaster, Management Act 2005 – Institutional and Financial Mechanism, National Policy on Disaster Management, National Guidelines and Plans on Disaster Management; Role of Government (local, state and national), Non-Government and Inter Governmental Agencies. Disaster Management Act in relation to COVID 19 Pandemic.	20%	6
Unit 4: Role of Science and Technology in Disaster Management Geo-informatics in Disaster Management (RS, GIS, GPS and RS), Disaster Communication System (Early Warning and Its Dissemination), Land, Planning and Development Regulations, Disaster Safe Designs and Constructions, Structural and Non-Structural Mitigation of Disasters, S&T Institutions for Disaster Management in India.	20%	6
Unit 5: Disaster Case Studies Various Case Studies on Disaster and Development, Disaster Prevention and Control, Risk Analysis and Management. Case study relating to COVID -19 to be explored.	20%	6

Instructional Method and Pedagogy:

Audio-Visual Lectures, Quizzes, Debates, Project works, Case studies, and Assignments Practical exercises are designed to understand the theory as taught in classroom. Hands on in practical session.

Course Outcomes:	Blooms' Taxonomy Domain	Blooms' Taxonomy Sub Domain
After successful completion of the above course, students will be able to:	Remember	Explain, Describe,
CO1 Possess awareness to mitigate the effects of disaster.		Discuss, Recall, Locate



CO2 Know local disaster management policies, regulations and authorities.	Apply	Apply, Practice, Interpret,
CO3 Contribute in capacity building measures to mitigate disasters.	Analyses and Evaluation	Select, Correlate Compare, Classify, Select,
CO4 Understanding role of science in mitigating disasters.	Create	Investigate Construct, Develop,
CO5 Contribute to safe society by the study of various disasters.	Understand	Produce Explain, Describe, outline, Predict, Summarize

	Learning Resources						
1.	Reference books:						
	 Alexander, D., Natural Disasters, Kluwer Academic London. Asthana, N. C., Asthana P., Disaster Management, Aavishkar Publishers. Carter, N., Disaster Management: A Disaster Manager's Handbook, Asian Development Bank. Collins, A.E., Disaster and Development, Routledge. Coppola, D.P., Introduction to International Disaster Management, 2nd Edition, Elsevier Science. Goyal, S.L., Encyclopedia of Disaster Management (Vols. 1-3), Deep & Deeep, New Delhi Gupta, A.K., Nair, S.S., Environmental Knowledge for Disaster Risk Management, NIDM, New Delhi. Ibrahimbegovic, A., Zlatar, M., Damage Assessment and Reconstruction after War or Natural Disaster, Springer. Menshikov, V.A., Perminov, A.N., Urlichich, Y.M., Global Aerospace Monitoring and Disaster Modh, S., Introduction to Disaster Management, Macmillian Publishers India Srivastava, H.N., Gupta, G.D., Management of Natural Disasters in Developing Countries, Daya Publishers, NIDM AND NIDMA publications 						
2.	Journals & Periodicals GSDMJ, disaster management act						



5	Other Electronic resources: GIDM,
	NIDM,

Evaluation Scheme	Total Marks			
Theory: Midsemester Marks	20 marks			
Theory: End Semester Marks	40 marks			
Theory: Continuous				
Evaluation Component Marks	Attendance	05 marks		
	MCQs	10 marks		
	Open Book Assignment	15 marks		
	Article Review	10 marks		
	Total	40 Marks		

	PSO1	PSO2	PSO3	PSO4	PSO5	PSO6
CO1	0	3	2	0	1	1
CO2	1	1	1	1	1	1
CO3	3	2	3	0	2	2
CO4	2	1	2	1	1	0
CO5	1	0	3	1	0	0

Mapping of POs & COs

	PO1	PO2	PO3	PO4	PO5	PO6
CO1	0	1	0	2	1	3
CO2	1	1	1	1	1	0
CO3	0	1	1	2	1	0
CO4	0	0	1	3	0	0
CO5	0	0	2	2	1	0



COURSE CODE	COURSE NAME	SEMESTER
BSMB501	VIROLOGY	${f V}$

	Teaching Scheme (Hours)				Teac Cre	ching edit	
Lectur e	Practica l	Tutorial	Total Hour s	Lecture	Practica l	Tutorial	Total Credi t
60	60	-	120	4	0	2	6

Course Pre-requisites	Basic knowledge of microbial world and viruses.			
Course Category	Core Professional			
Course focus	Employability			
Rationale	Students will get an overview of types of viruses, their replication strategies and mechanisms for development of viral infectious diseases.			
Course Revision/ Approval Date:	14/03/2020			
Course Objectives (As per Blooms' Taxonomy)	 Remember The course will give an overve their replication strategies and mechanisms infectious diseases. Apply The distinctive characteristics between eukaryotes and viruses. The biological, chemproperties of viruses. The principles of viral. Analyses The structure of viruses and their. Create The bases of viral genetic variability viral evolution. Understand The interaction virus-host and remainders. 	en prokaryotes, nical and physic taxonomy replicative cyc- ity and the prin	nt of viral cal	
Cour	se Content (Theory)	Weightage	Contact hours	
Unit 1: Nature and Properties of Viruses Introduction: Discovery of viruses, nature and definition of viruses, general properties, concept of viroids, virusoids, satellite viruses and Prions. Theories of viral origin Structure of Viruses: Capsid symmetry, enveloped and non-enveloped viruses Isolation, purification and cultivation of viruses' Viral taxonomy: Classification and nomenclature of different groups of viruses			15	
	sity, classification, one step multiplication curve, mbda phage) concept of early and late proteins, mbda phage.	25%	15	



Unit 3:Viral Transmission, Salient features of viral nucleic acids and Replication Modes of viral transmission: Persistent, non-persistent, vertical and horizontal Salient features of viral Nucleic acid: Unusual bases (TMV,T4 phage), overlapping genes (\$\phi\$X174, Hepatitis B virus), alternate splicing (HIV), terminal redundancy (T4 phage), terminal cohesive ends (lambda phage), partial double stranded genomes (Hepatitis B), long terminal repeats (retrovirus), segmented (Influenza virus), and non-segmented genomes (picornavirus), capping and tailing (TMV) Viral multiplication and replication strategies: Interaction of viruses with cellular receptors and entry of viruses. Replication strategies of viruses as per Baltimore classification (phi X 174, Retroviridae, Vaccinia, Picorna). Assembly, maturation and release of virions.	20%	12
Unit 4: Viruses and Cancer Introduction to oncogenic viruses Types of oncogenic DNA and RNA viruses: Concepts of oncogenes and proto- oncogenes.	20%	12
Unit 5: Prevention & control of viral diseases Antiviral compounds and their mode of action Interferon and their mode of action General principles of viral vaccination Applications of Virology Use of viral vectors in cloning and expression, Gene therapy and Phage display.	10%	06

List Of Practical	Weightage	Contact
		hours
1:Isolation and enumeration of bacteriophages (PFU) from water/sewage	20%	12
sample using double agar layer technique.		

Instructional Method and Pedagogy:

Audio-Visual Lectures, Quizzes, Debates, Project works, Case studies, and Assignments Practical exercises are designed to understand the theory as taught in classroom. Hands on in practical session.

Course Outcomes:	Blooms' Taxonomy Domain	Blooms' Taxonomy Sub Domain
After successful completion of the above course, students will be able to: CO1 Understand the structure and life cycle of viruses as well as viral evolution.	Remember	Explain, Describe, Discuss, Recall, Locate



CO2 Know the mechanisms of host immune responses to viral infections.	Apply	Apply, Practice, Interpret, Select, Correlate
CO3 Understand the pathogenesis of viral infections.	Analyses and Evaluation	Compare, Classify, Select, Investigate
CO4 Understand the methodological approaches in diagnostic investigation of infectious diseases.	Create	Construct, Develop, Produce
CO5 Understand the epidemiology of infections.	Understand	Explain, Describe, outline, Predict, Summarize

	Learning Resources				
1.	Reference books:				
	 Bos L. (1999) Plant viruses-A text book of plant virology by. Backhuys Publishers. Versteeg J. (1985). A Color Atlas of Virology. Wolfe Medical Publication. Dimmock, NJ, Easton, AL, Leppard, KN (2007). Introduction to Modern Virology. 6th edition, Blackwell Publishing Ltd. Carter J and Saunders V (2007). Virology: Principles and Applications. John Wiley and Sons. Flint SJ, Enquist, LW, Krug, RM, Racaniello, VR, Skalka, AM (2004). Principles of Virology, Molecularbiology, Pathogenesis and Control. 2nd edition. ASM press Washington DC. Levy JA, Conrat HF, Owens RA. (2000). Virology. 3rd edition. Prentice Hall publication, New Jersey. Wagner EK, Hewlett MJ. (2004). Basic Virology. 2nd edition. Blackwell Publishing. Mathews. (2004). Plant Virology. Hull R. Academic Press, New York. Nayudu MV. (2008). Plant Viruses. Tata McGraw Hill, India. 				
2.	Journals & Periodicals 1. Journal of Virology.				
	2. Retrovirology.				
	O				
5					
	<i>C.</i>				
5	 Journal of Virology. Retrovirology. 				



Evaluation Scheme	Total Marks	
Theory: Mid semester Marks	20 marks	
Theory: End Semester Marks	40 marks	
Theory: Continuous Evaluation Component Marks	Attendance	05 marks
11242210	MCQs Open Book Assignment	10 marks 15 marks
	Article Review	10 marks
	Total	40 Marks
Practical Marks		
	Attendance	05 marks
	Practical Exam	30 marks
	Viva	10 marks
	Journal	05 marks
	Total	50 Marks

	PSO1	PSO2	PSO3	PSO4	PSO5	PSO6
CO1	2	2	3	1	2	1
CO2	2	2	3	1	2	1
CO3	2	2	2	1	2	3
CO4	2	2	2	1	2	3
CO5	2	2	3	1	2	1

Mapping of POs & COs

8	PO1	PO2	PO3	PO4	PO5	PO6
CO1	3	2	0	1	1	0
CO2	3	2	0	0	1	0
CO3	3	1	3	0	2	2
CO4	3	1	3	0	2	2
CO5	3	2	3	0	2	2



COURSE CODE	COURSE NAME	SEMESTER
BSMB502	MEDICAL MICROBIOLOGY	${f v}$

Teaching Scheme (Hours)					Teachin	g Credit	
Lecture	Practical	Tutorial	Total Hours				Total Credit
60	60	0	120	4	2	0	6

Course Pre-requisites	Students should have basic knowledge about medical microbiology.
Course Category	Core Professional
Course focus	Employability
Rationale	To have an overview of medical microbiology
Course Revision/Approval	14/03/2020
Date:	
Course Objectives	Remember To study normal microflora of human body
(As per Blooms' Taxonomy)	Apply To study host pathogen interactions.
	Analyses To study bacterial, viral, fungal, protozoan diseases.
	Create To study about anti-microbial agents/drugs.
	Understand To study about the mode of action of drugs.

Course Content (Theory)	Weightage	Contact
		hours
Unit 1: Normal microflora of the human body and host pathogen	20%	6
interaction Normal microflora of the human body: Importance of normal		
microflora, normal microflora of skin, throat, gastrointestinal tract,		
urogenital tract Host pathogen interaction: Definitions - Infection,		
Invasion, Pathogen, Pathogenicity, Virulence, Toxigenicity, Carriers and		
their types, Opportunistic infections, Nosocomial infections. Transmission		
of infection, Pathophysiologic effects of LPS. Sample collection, transport		
and diagnosis Collection, transport and culturing of clinical samples,		
principles of different diagnostic tests (ELISA, Immunofluorescence,		
Agglutination based tests, Complement fixation, PCR, DNA probes).		
Unit 2: Bacterial diseases List of diseases of various organ systems	20%	6
and their causative agents. The following diseases in detail with		
Symptoms, mode of transmission, prophylaxis and control respiratory		
diseases: Streptococcus pyogenes, Haemophilus influenzae,		
Mycobacterium tuberculosis Gastrointestinal Diseases: Escherichia		
coli, Salmonella typhi, Vibrio cholerae, Helicobacter pylori Others:		
1		
Staphylococcus aureus, Bacillus anthracis, Clostridium tetani,		
Treponema pallidum, Clostridium deficient.		



Unit 3:Viral diseases List of diseases of various organ systems and their causative agents. The following diseases in detail with Symptoms, mode of transmission, prophylaxis and control Polio, Herpes, Hepatitis, Rabies, Dengue, AIDS, Influenza with brief description of swine flu, Ebola, Chikungunya, Japanese Encephalitis Protozoan Diseases List of diseases of various organ systems and their causative agents. The following diseases in detail with Symptoms, mode of transmission, prophylaxis and control Malaria, Kala-azar.	20%	6
Unit 4:Fungal diseases Brief description of each of the following types of mycoses and one representative disease to be studied with respect to transmission, symptoms and prevention Cutaneous mycoses: Tinea pedis (Athlete's foot) Systemic mycoses: Histoplasmosis Opportunistic mycoses: Candidiasis parameters -pH, temperature, dissolved oxygen, foaming and aeration	20%	6
Unit 5:Antimicrobial agents: General characteristics and mode of action. Antibacterial agents: Five modes of action with one example each: Inhibitor of nucleic acid synthesis; Inhibitor of cell wall synthesis; Inhibitor of cell membrane function; Inhibitor of protein synthesis; Inhibitor of metabolism Antifungal agents: Mechanism of action of Amphotericin B, Griseofulvin Antiviral agents: Mechanism of action of Amantadine, Acyclovir, Azidothymidine Antibiotic resistance, MDR, XDR, MRSA,NDM-1.	20%	6

List Of Practical	Weightage	Contact hours
1. Identify bacteria (any three of E. coli, Salmonella, Pseudomonas,	20%	12
Staphylococcus, Bacillus) using laboratory strains		
2. Study of composition and use of important differential media for identification of bacteria: EMB Agar, McConkey agar, Mannitol salt agar, Deoxycholate citrate agar, TCBS.	20%	12
3. Study of bacterial flora of skin by swab method, Perform antibacterial	20%	12
sensitivity by Kirby-Bauer method, Determination of minimal inhibitory concentration (MIC) of an antibiotic.		
4. Study symptoms of the diseases with the help of photographs: Polio,	20%	12
anthrax, herpes, chicken pox, HPV warts, AIDS (candidiasis),		
dermatomycoses (ring worms).		
5. Study of various stages of malarial parasite in RBCs using permanent mounts.	20%	12



Instructional Method and Pedagogy:

Audio-Visual Lectures, Quizzes, Debates, Project works, Case studies, and Assignments Practical exercises are designed to understand the theory as taught in classroom. Hands on in practical session.

	Course Objectives:	Blooms' Taxonomy Domain	Blooms' Taxonomy Sub Domain
CO1	To study normal microflora of human body.	Remember	Explain, Describe, Discuss, Recall, Locate
CO2	To study host pathogen interactions.	Apply	Apply, Practice, Interpret, Select, Correlate
CO3	To study bacterial, viral, fungal, protozoan diseases.	Analyses and Evaluation	Compare, Classify, Select, Investigate
CO4	To study about anti-microbial agents/drugs.	Create	Construct, Develop, Produce
CO5	To study about the mode of action of drugs	Understand	Explain, Describe, outline, Predict, Summarize

Learning Resources 1. Reference books: Ananthanarayan R. and Paniker C.K.J. (2009) Textbook of Microbiology. 8th edition, University Press Publication Brooks G.F., Carroll K.C., Butel J.S., Morse S.A. and Mietzner, T.A. (2013) Jawetz, Melnick and Adelberg's Medical Microbiology. 26th edition. McGraw Hill Publication Goering R., Dockrell H., Zuckerman M. and Wakelin D. (2007) Mims' Medical Microbiology. 4th edition. Elsevier Willey JM, Sherwood LM, and Woolverton CJ. (2013) Prescott, Harley and Klein's Microbiology. 9th edition. McGraw Hill Higher Education Madigan MT, Martinko JM, Dunlap PV and Clark DP. (2014). Brock Biology of Microorganisms. 14th edition. Pearson International. Atlas RM. (1997). Principles of Microbiology. 2nd edition. WM.T.Brown Publishers. 7. Pelzer MJ, Chan ECS and Krieg NR. (1993). Microbiology. 5th edition. McGraw Hill Book Company. Stanbury PF, Whitaker A and Hall SJ. (2006). Principles of Fermentation Technology. 2nd edition, Elsevier Science Ltd.



2.	Journals & Periodicals 1) Journal of Clinical Microbiology 2) Microbiology
3.	Other Electronic resources: https://www.microbiologyresearch.org/content/journal/jmm

Evaluation Scheme	Total Marks					
Theory: Mid semester Marks	20 marks					
Theory: End Semester Marks	40 marks					
Theory: Continuous Evaluation Component Marks	Attendance MCQs Open Book Assignment Article Review Total	05 marks 10 marks 15 marks 10 marks 40 Marks				
Practical Marks	Attendance Practical Exam Viva Journal Total	05 marks 30 marks 10 marks 05 marks 50 Marks				

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	PSO1	PSO2	PSO3	PSO4	PSO5	PSO6
CO1	3	3	3	0	0	0
CO2	3	2	2	0	2	0
CO3	3	2	2	0	2	0
CO4	3	3	3	0	2	0
CO5	3	2	2	0	2	0

Mapping of POs & COs

┺.							
		PO1	PO2	PO3	PO4	PO5	PO6
	CO1	3	2	3	0	1	0
	CO2	2	2	2	0	0	0
	CO3	2	2	2	0	0	0



CO4	2	2	2	0	1	0
CO5	2	2	2	0	0	0



COURSE CODE COURSE NAME SEMESTER
BSMB503 MICROBIAL V
BIOTECHNOLOGY

Teaching Scheme (Hours)					Teachin	g Credit	
Lecture	Practical	Tutorial	Total Hours	Lecture	Practical	Tutorial	Total Credit
60	60	0	120	4	2	0	6

Course Pre-requisites	Students should have basic knowledge of microbial biotechnology.
Course Category	Core Professional
Course focus	Employability
Rationale	To have an overview of microbial biotechnology.
Course Revision/Approval	14/03/2020
Date:	
Course Objectives	Remember To introduce students to the developments/advances
(As per Blooms' Taxonomy)	made in the field of microbial biotechnology.
	Apply To know the usage of microbes and their products in therapeutics.
	Analyses To understand the applications of microbes in Biotransformation. Create To demonstrate the recovery of microbial products and an overview of Intellectual Property Rights.
	Understand To get insights in using microbes for Bioenergy and bioremediations.

Course Content (Theory)	Weightage	Contact hours
Unit 1: Microbial biotechnology: Scope and its applications in human therapeutics, agriculture (Biofertilizers, PGPR, Mycorrhizae), environmental, and food technology Use of prokaryotic and eukaryotic microorganisms in biotechnological applications Genetically engineered microbes for industrial application: Bacteria and yeast	20%	6
Unit 2:Recombinant microbial production processes in pharmaceutical industries - Streptokinase, recombinant vaccines (Hepatitis B vaccine) Microbial polysaccharides and polyesters, Microbial production of bio-pesticides, bioplastics Microbial biosensors. Therapeutics and Host pathogen interactions	20%	6



Unit 3: RNAi & Applications of Microbes in Biotransformation RNAi and its applications in silencing genes, drug resistance; Microbial based transformation of steroids and sterols Bio- catalytic processes and their industrial applications: Production of high fructose syrup and production of cocoa butter substitute.	20%	6
Unit 4:Microbial Products, their Recovery & Intellectual Property Rights. Microbial product purification: filtration, ion exchange & affinity chromatography techniques Immobilization methods and their application: Whole cell immobilization; Patents, Copyrights, Trademarks.	20%	6
Unit 5:Bio-ethanol and bio-diesel production: commercial production from lignocellulosic waste and algal biomass, Biogas production: Methane and hydrogen production using microbial culture. Microorganisms in bioremediation: Degradation of xenobiotics, mineral recovery, removal of heavy metals from aqueous effluents	20%	6

	List Of Practical	Weightage	Contact hours
1	. Study yeast cell immobilization in calcium alginate gels	20%	12
2	2. Study enzyme immobilization by sodium alginate method	20%	12
3	3. Isolation of xylanase or lipase producing bacteria	20%	12
4	Pigment production from fungi or bacteria.	20%	12

Instructional Method and Pedagogy:Audio-Visual Lectures, Quizzes, Debates, Project works, Case studies, and Assignments Practical exercises are designed to understand the theory as taught in classroom. Hands on in practical session.

	Course Objectives:	Blooms' Taxonomy Domain	Blooms' Taxonomy Sub Domain
CO1	Remember To introduce students to the developments/advances made in the field of microbial biotechnology.	Remember	Explain, Describe, Discuss, Recall, Locate
CO2	ply To know the usage of microbes and their products in therapeutics.	Apply	Apply, Practice, Interpret, Select, Correlate
CO3	Analyses To understand the applications of microbes in Biotransformation.	Analyses and Evaluation	Compare, Classify, Select, Investigate



CO4	Create To demonstrate the recovery of microbial products and an overview of Intellectual Property Rights.	Create	Construct, Develop, Produce
CO5	iderstand To get insights in using microbes for Bioenergy and bioremediations.	Understand	Explain, Describe, outline, Predict, Summarize

Learning Res	sources			
1.	Reference books:			
	1. Gupta PK (2009) Elements of Biotechnology 2nd edition, Rastogi Publications,			
	2. Crueger W, Crueger A (1990) Biotechnology: A text Book of Industrial Microbiology 2ndedition Sinauer associates, Inc.			
	3. Ratledge, C and Kristiansen, B. (2001). Basic Biotechnology, 2nd Edition, Cambridge University Press.			
	4. Demain, A. L and Davies, J. E. (1999). Manual of Industrial Microbiology and Biotechnology, 2nd Edition, ASM Press.			
	5. Swartz, J. R. (2001). Advances in Escherichia coli production of therapeutic proteins. Current Opinion in Biotechnology, 12, 195–201.			
	6. Prescott, Harley and Klein's Microbiology by Willey JM, Sherwood LM, Woolverton CJ (2014), 9th edition, Mc Graw Hill Publishers.			
	7. Glazer AN and Nikaido H (2007) Microbial Biotechnology, 2nd edition, Cambridge University Press			
	8. Glick BR, Pasternak JJ, and Patten CL (2010) Molecular Biotechnology 4th edition, ASM Press			
	9. Stanbury PF, Whitaker A, Hall SJ (1995) Principles of Fermentation			
	Technology 2nd edition., Elsevier Science			
2.	Journals & Periodicals			
	1) Enzyme and Microbial Technology			
	2) Applied Biochemistry and Biotechnology			
	3) Microbial biotechnology			
	4) Applied microbiology and biotechnology			
	5) Current Science			



3.	Other Electronic resources:
	1.Science Daily – Microbiology News
	2.https://www.sciencenews.org/topic/microbes
	3.https://www.labroots.com/trending/microbiolog
	y 4.Google books:Microbial biotechnology: https://www.google.co.in/books/edition/Microbial_Biotechnology_Basic_Research_a/q_LvDwAAQBAJ?hl=en&gbpv= 1&dq=microbial+biotechnology&printsec=frontcover.

Evaluation Scheme	Total Marks	
Theory: Mid semester Marks	20 marks	
Theory: End Semester Marks	40 marks	
Theory: Continuous Evaluation Component Marks	Attendance MCQs Open Book Assignment Article Review Total	05 marks 10 marks 15 marks 10 marks 40 Marks
Practical Marks	Attendance Practical Exam Viva Journal Total	05 marks 30 marks 10 marks 05 marks 50 Marks

	PSO1	PSO2	PSO3	PSO4	PSO5	PSO6
CO1	3	2	2	1	2	2
CO2	3	1	3	1	1	3
CO3	3	3	3	2	1	1
CO4	3	3	3	2	1	1
CO5	3	2	2	1	2	2

Mapping of POs & COs

	PO1	PO2	PO3	PO4	PO5	PO6
CO1	3	2	2	1	1	1
CO2	3	3	3	1	3	1



CO3	3	3	2	1	1	3
CO4	3	3	2	1	1	3
CO5	3	2	2	1	1	1



COURSE CODE BSMB504	COURSE NAME RESEARCH METHODOLOGY	SEMESTER V
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Teaching Scheme (Hours)				Teaching Credit			
Lecture	Practical	Tutorial	Total Hour s	Lecture	Practica l	Tutorial	Total Credit
60	60	-	120	4	0	2	6

Course Pre-requisites	Basic knowledge of Research Methodology.
Course Category	Core Professional
Course focus	Employability
Rationale	Students will get an overview of Methodology of research and handling processes
Course Revision/ Approval Date:	14/03/2020
Course Objectives (As per Blooms' Taxonomy)	 Remember The objective of this course is to familiarize students with the research methodology and importance of project writing. Apply understand some basic concepts of research and its methodologies. Analyses Students should understand a general definition of research design Create Students should know why educational research is undertaken, and the audiences that profit from research studies. Understand Students should be able to identify the overall process of designing a research study from its inception to its report.



Course Content (Theory)	Weightage	Contact hours
Unit 1: Theory: Basic concepts of Research: definition and types of research (Descriptive vs analytical; applied vs fundamental; quantitative vs qualitative; conceptual vs emperical). Research methods vs methodology. Literature-review and its consolidation; Library research; field research; laboratory research. Defining the Research Problem: Selecting and defining a research problem, Reviewing and conducting literature search, Developing a research plan.	20%	12
Unit 2: Theory: Data collection and documentation of observations: Methods of data collection, Maintaining a laboratory record; Tabulation and generation of graphs. Imaging of tissues specimens and application of scale bars.	20%	12
Unit 3: Theory: Model organisms in biology (A Brief overview): Genetics, Physiology, Biochemistry, Molecular Biology, Cell Biology, Genomics, Proteomics- Transcriptional regulatory network. Designing of Experiment: Different experimental designs – single and multifactorial design, Making measurements and sources of error in measurements	20%	12
Unit 4: Theory: General laboratory practices: Common calculations in laboratories. Understanding the details on the label of reagent bottles. Molarity and normality of common acids and bases. Preparation of solutions. Dilutions. Percentage solutions. Molar, molal and normal solutions. Technique of handling micropipettes; Knowledge about common toxic chemicals and safety measures in their handling.	20%	12
Unit 5: Theory: The art of scientific writing and its presentation: Numbers, units, abbreviations and nomenclature used in scientific writing. Writing references. Powerpoint presentation. Poster presentation. Scientific writing and ethics, Introduction to copyright-academic misconduct/plagiarism	20%	12

List Of Practical	Weightage	Contact hours
1.Each student will be asked to make presentation about the project including literature available, objective sought and work plan including methodologies as described above.	20%	12
2. The student will write standard operating protocols (SOPs) and identify requirement for equipment and reagents.	20%	12
3. Experiments based on chemical calculations	20%	12
4. Poster presentation on defined topics.	20%	12



5.Technical writing on topics assigned.	20%	12

Course Outcomes:	Blooms' Taxonomy Domain	Blooms' Taxonomy Sub Domain
After successful completion of the above course, students will be able to: CO1 Understand some basic concepts of research and its methodologies.	Remember	Explain, Describe, Discuss, Recall, Locate
CO2 Identify appropriate researchtopics.	Apply	Apply, Practice, Interpret, Select,
CO3 Select and define appropriate research problem and parameters. CO4 Prepare a project proposal (to undertake a project).	Analyses and Evaluation	Correlate Compare, Classify, Select, Investigate Construct,
	Create	Develop,
CO5 Organize and conduct research (advanced project) in a more appropriate manner.	Understand	Produce Explain, Describe, outline, Predict, Summarize



	Learning Resources
1.	 Reference books: Dawson, C. (2002). Practicalresearch methods. UBS Publishers, New Delhi. Research in Education (1992) 6th ed., Best, J.W. and Kahn, J.V., Prentice Hall of India Pvt. Ltd. At the Bench: A Laboratory Navigator (2005) Barker, K., Cold Spring Harbor Laboratory Press (New York), ISBN: 978-087969708-2. 51 Research Methodology - Methods and Techniques (2004) 2nd ed., Kothari C.R., New Age International Publishers. Research Methodology: A Step by Step Guide for Beginners (2005) 2nd ed., Kumar R., Pearson Education
2.	Journals & Periodicals 1. Journal of Mixed Method Research 2. Journal of Research practice. 3. Survey Research Methods
5	Other Electronic resources NPTL

Evaluation Scheme	Total Marks	
Theory: Mid semester Marks	20 marks	
Theory: End Semester Marks	40 marks	
Theory: Continuous Evaluation Component Marks	Attendance MCQs Open Book Assignment Article Review Total	05 marks 10 marks 15 marks 10 marks 40 Marks
Practical Marks	Attendance Practical Exam Viva Journal Total	05 marks 30 marks 10 marks 05 marks 50 Marks



	PSO1	PSO2	PSO3	PSO4	PSO5	PSO6
CO1	0	1	2	3	1	0
CO2	1	1	1	1	1	0
CO3	0	1	1	2	1	0
CO4	0	0	1	3	0	0
CO5	0	0	2	2	1	0

Mapping of POs & COs

	PO1	PO2	PO3	PO4	PO5	PO6
CO1	0	1	2	3	1	0
CO2	1	1	1	1	1	0
CO3	0	1	1	2	1	0
CO4	0	0	1	3	0	0
CO5	0	0	2	2	1	0



COURSE CODE	COURSE NAME	SEMESTER
BSMB505	BIOANALYTICAL TOOLS	V

Teaching Scheme (Hours)					ching edit		
Lecture	Practical	Tutorial	Total Hour s	Lecture	Practica l	Tutorial	Total Credit
60	60	-	120	4	0	2	6
Course Pro	e-requisites	Stude analy	ents should tical tools for	1	basic knowledge and the or	•	biological l.
Course Ca	tegory	Core	Professional				
Course foc	eus	Empl	Employability				
Rationale		Toun	derstand the p	rinciple of var	ious biophysic	al techniques.	
Course Re Approval I		14/03	/2020				
Course Objectives (As per Blooms' Taxonomy)		2. 4. 6 5. 1	Remember To bimolecular se Apply To und techniques. Analyses To uvarious technic Create To for characteriz Understand Tanalytical mind	eparation and perstand the perstand the ques. learn variation of the bifosolve diverse	purification. rinciple of va instrumentatio rious biophy omolecules.	rious biophysi n and applicat ysical tech	cal ion of niques

Course Content (Theory)	Weightage	Contact hours
Unit 1: Simple microscopy, phase contrast microscopy, fluorescence and electron microscopy (TEM and SEM)	20%	12
Unit 2: pH meter, absorption and emission spectroscopy, Principle and law of absorption fluorimetry, colorimetry, spectrophotometry (visible, UV, infra-red).	20%	12
Unit 3: Centrifugation, cell fractionation techniques, isolation of sub- cellular organelles and particles.	10%	6
Unit 4: Introduction to the principle of chromatography. Paper chromatography, thin layer chromatography, column chromatography: silica and gel filtration, affinity and ion exchange chromatography, gas chromatography, HPLC.	25%	15



roduction to electrophoresis. Starch-gel, polyacrylamide gel (native PAGE). Introduction to agarose-gel electrophoresis, pulse field gel esis, immuno- electrophoresis, isoelectric focusing, Western troduction to Biosensors and blogy and their applications.	3 = 0 /	15	
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List Of Practical	Weightage	Contact hours
1: Demonstration of Fluorescence microscope	10%	4
2: Preparation of protoplasts/spheroplast from leaves/ Bacteria	10%	4
3: Verify Beer's and Lambert's law using KMnO4	10%	4
4: Separation of cellular components using sucrose gradient	10%	4
5: Separation of Serum and blood cells	10%	4
6: Paper, TLC and column chromatography of amino acids, lipids and plant pigments	20%	8
7: Working of HPLC,GC (Demo)	20%	8
8: SDS PAGE of Protein	10%	4
9: Native Gel electrophoresis of Protein	10%	4

Course Outcomes:	Blooms' Taxonomy Domain	Blooms' Taxonomy Sub Domain
After successful completion of the above course, students will be able to: CO1 Assimilate the principles and applications of centrifuge, electrophoresis.	Remember	Explain, Describe, Discuss, Recall, Locate



CO2 Employ the knowledge for the separation of proteins/peptides by selecting appropriate separation techniques.	Apply	Apply, Practice, Interpret, Select, Correlate
CO3 Characterize certain functionalities of biomolecules by using spectroscopic techniques.	Analyses and Evaluation	Compare, Classify, Select, Investigate
CO4 Assimilate the principles and applications of chromatography in research and related experiments.	Create	Construct, Develop, Produce
CO5 Plan experiments for separations and purifications of biomolecules.	Understand	Explain, Describe, outline, Predict, Summarize

I			
	Learning Resources		
1.	Reference books:		
	1. Douglas A. Skoog, James J. Leary; Principles of instrumental analysis. ISBN: 0-03-075398-8		
	2. Willard, H.H., Merritt L.L. Dean J.A. and Settle F.A., & quot; Instrumental Methods of Analysis & quot;, 7th Ed., Wadsworth Publishing Co., 1986.		
	3. Instrumentation measurements and analysis – 2nd edition (2003). Nakra and Choudhari, Tata Mc Graw Hill, India.		
	4. K. Wilson, J. M. Walker, Eds., Principles and techniques of biochemistry and molecular biology (Cambridge University Press, Cambridge, UK: New York, 7th ed., 2009).		
	5. I. D. Campbell, Biological spectroscopy (Benjamin/Cummings Pub. Co, Menlo Park, Calif, 1984), Biophysical techniques series		
	6. D. L. Spector, R. D. Goldman, Eds., Basic methods in microscopy: protocols and concepts from cells: a laboratory manual (Cold Spring Harbor Laboratory Press, Cold Spring Harbor, N.Y, 2006).		
	7. R. F. Boyer, Modern experimental biochemistry (Benjamin Cummings, San Francisco, 3rd ed., 2000)		
2.	Journals & Periodicals Analytical		
	methods		
	Chromatography Today, The Scientist		
5	Other Electronic resources		
	https://edu.rsc.org/resources/analysis		



Evaluation Scheme	Total Marks	
Theory: Mid semester Marks	20 marks	
Theory: End Semester Marks	40 marks	
Theory: Continuous Evaluation Component Marks	Attendance	05 marks
	MCQs Open Book Assignment	10 marks 15 marks
	Article Review Total	10 marks 40 Marks
Practical Marks		
	Attendance Practical Exam	05 marks 30 marks
	Viva Journal	10 marks 05 marks
	Total	50 Marks

	PSO1	PSO2	PSO3	PSO4	PSO5	PSO6
CO1	3	2	1	0	1	1
CO2	2	2	2	1	2	2
CO3	1	2	2	2	2	2
CO4	2	2	1	1	2	2
CO5	2	2	2	1	2	2

Mapping of POs & COs

	PO1	PO2	PO3	PO4	PO5	PO6
CO1	3	2	3	1	2	2
CO2	2	2	2	3	1	1
CO3	3	2	3	2	2	1
CO4	1	1	2	2	2	2
CO5	2	2	2	2	2	1



COURSE CODE	COURSE NAME	SEMESTER
BSMB506	Microbes in sustainable	${f V}$
	agriculture and development	

Teaching Scheme (Hours)				Teachin	g Credit		
Lecture	Practical	Tutorial	Total Hours	Lecture	Practical	Tutorial	Total Credit
60	60	0	120	4	2	0	6

Course Pre-requisites	Students should have basic knowledge about microbes in sustainable		
_	agriculture and development.		
Course Category	Core Professional		
Course focus	Employability		
Rationale	To have an overview of microbes in sustainable agriculture and		
	development		
Course Revision/ Approval	14/03/2020		
Date:			
Course Objectives	Remember To learn balancing of our economic, environmental		
(As per Blooms' Taxonomy)	and social needs.		
	ply To study role of microbes in sustainable agriculture.		
	Analyses To study the biology and evolution of microorganisms and their interactions with the environment. Create To ensure healthy lives and promote well-being for all at all ages.		
	derstand To explore about good practices in agriculture establishment using microbes.		

Course Content (Theory)	Weightage	Contact
		hours
Unit 1: Soil Microbiology - Soil as Microbial Habitat, Soil profileand	20%	12
properties, Soil formation, Diversity and distribution of microorganisms		
in soil		
Unit 2: Mineralization of Organic & Inorganic Matter in Soil.		
Mineralization of cellulose, hemicelluloses, lignocelluloses, lignin and		
humus, phosphate, nitrate, silica, potassium.		
Unit 3:Microbial Activity in Soil and Green House Gases. Carbon	20%	12
dioxide, methane, nitrous oxide, nitric oxide – production and control		



Unit 4:Microbial Control of Soil Borne Plant Pathogens Biocontrol mechanisms and ways, Microorganisms used as biocontrol agents against Microbial plant pathogens, Insects, Weeds Biofertilization, Phyto stimulation, Bioinsecticides. Plant growth promoting bacteria, biofertilizers – symbiotic (Brady rhizobium, Rhizobium, Frankia), Non-Symbiotic (Az spirillum, Azotobacter, Mycorrhizae, MHBs, Phosphate solubilizers, algae), Novel combination of microbes as biofertilizers, PGPRs.	20%	12
Unit 5:Secondary Agriculture Biotechnology Biotech feed, Silage, Bio manure, biogas, biofuels – advantages and processing parameters. GM crops - Advantages, social and environmental aspects, Bt crops, golden rice, transgenic animals.		

List Of Practical	Weightage	Contact hours
1. Study soil profile.	20%	12
2. Study microflora of different types of soils.	20%	12
3. Rhizobium as soil inoculants characteristics and field application.	20%	12
4. Azotobacter as soil inoculants characteristics and field application.	20%	12
5. Design and functioning of a biogas plant. Isolation of cellulose degrading organisms.	20%	12

${\bf Instructional\ Method\ and\ Pedagogy:}$

	Course Objectives:	Blooms' Taxonomy Domain	Blooms' Taxonomy Sub Domain
CO1	To learn balancing of our economic, environmental and social needs.	Remember	Explain, Describe, Discuss, Recall, Locate
CO2	To study role of microbes in sustainable agriculture.	Apply	Apply, Practice, Interpret, Select, Correlate
CO3	To study the biology and evolution of microorganisms and their interactions with the environment.	Analyses and Evaluation	Compare, Classify, Select, Investigate
CO4	To ensure healthy lives and promote well-being for all at all ages.	Create	Construct, Develop, Produce



To explore about good practices in agriculture establishment using microbes.

Understand

Explain, Describe, outline, Predict, Summarize

Learning Re	sources
1.	Reference books:
	1. Agrios GN. (2006). Plant Pathology. 5th edition. Academic press, San Diego
	2. Singh RS. (1998). Plant Diseases Management. 7th edition. Oxford & IBH, New
	Delhi.
	3. Glick BR, Pasternak JJ, and Patten CL (2010) Molecular Biotechnology 4th edition, ASM Press
	4. Atlas RM and Bartha R. (2000). Microbial Ecology: Fundamentals &
	Applications. 4th edition.
	5. Benjamin/Cummings Science Publishing, USA
	6. Maier RM, Pepper IL and Gerba CP. (2009). Environmental Microbiology.
	2nd edition, Academic Press.
	7. Barton LL & Northup DE (2011). Microbial Ecology. 1st edition, Wiley
	Blackwell, USA. 2.Campbell RE. (1983). Microbial Ecology.
	8. Blackwell Scientific Publication, Oxford, England. 3. Coyne MS. (2001). Soil
	Microbiology: An Exploratory Approach. Delmar Thomson Learning. 4.Altman
	A (1998). Agriculture Biotechnology, Ist edition, Marcel decker Inc.
	9. Mahendra K. Rai (2005). Hand Book of Microbial Biofertilizers, The Haworth
	Press, Inc. New York
	10. Reddy, S.M. et. al. (2002). Bioinoculants for Sustainable Agriculture and Forestry, Scientific Publishers.
	11. Saleem F and Shakoori AR (2012) Development of Bioinsecticide, Lap
	Lambert Academic Publishing GmbH KG
2.	Journals & Periodicals
	1. Current Sciences in Microbiology.
	2. Beneficial Microbes for Sustainable Agriculture.
5	Other Electronic management
5	Other Electronic resources:
	NPTEL

Evaluation Scheme	Total Marks
Theory: Mid semester Marks	20 marks
Theory: End Semester Marks	40 marks



Theory: Continuous		
Evaluation Component	Attendance	05 marks
Marks	MCQs	10 marks
	Open Book Assignment	15 marks
	Article Review	10 marks
	Total	40 Marks
Practical Marks	Attendance	05 marks
Practical Marks		
	Practical Exam	30 marks
	Viva	10 marks
	Journal	05 marks
	Total	50 Marks

	PSO1	PSO2	PSO3	PSO4	PSO5	PSO6
CO1	2	1	2	0	1	3
CO2	1	0	3	3	2	3
CO3	2	2	2	0	2	3
CO4	1	0	3	3	2	3
CO5	2	1	2	0	1	3

Mapping of POs & COs

99 0 0 0	• • • • • •					
	PO1	PO2	PO3	PO4	PO5	PO6
CO1	3	1	0	1	0	0
CO2	1	2	3	0	1	0
CO3	1	2	3	1	0	3
CO4	1	2	3	0	1	0
CO5	3	1	0	1	0	0



		Semester – VI								
Sr. No.	Course Code									
A. Ability Enhanceme nt Compulsory Course										
1	AECC601	Indian Constitution	2	0	0	2	100			
		B. Core Course	J							
2	BSMB601	Immunology	4	0	2	6	150			
3	BSMB602	Bioinformatics & Drug Discovery, Design and Development	4	0	2	6	150			
		C. Discipline Specific Generic Electives (A	Any Tv	vo)						
4	BSMB603	Food & Dairy Microbiology	4	0	2	6	150			
5	BSMB604	Biosafety and Intellectual Property Rights	4	0	2	6	150			
6	BSMB605	Environmental Microbiology	4	0	2	6	150			
7	BSMB606	Advances in Microbiology	4	0	2	6	150			
		<u></u>		<u> </u>	Total	26	700			

School of Science B.Sc. Microbiology, Course Curriculum Academic Year, 2024-25



Teaching Scheme Semester-VI

Sr	Course Code	Course Name		Sch	ching neme s/week)			Teac	ching C	Credit	Evaluation Scheme					
N o	course code	Course Name	L	P	Т	Tota l	L	P	Т	Tota 1	Theory :MS Marks	Theory :CEC Marks	Theory :ES Marks	Theor y Mark s	Practic al Marks	Total Mark s
					A. Abil	ity Enha	ncemen	t Comp	ulsory	Course						
1	AECC601	Indian Constitution	2	0	0	2	2	0	0	2	20	40	40	100	00	100
	B. Core Course															
2	BSMB601	Immunology	4	4	0	8	4	2	0	6	20	40	40	100	50	150
3	BSMB602	Bioinformatics & Drug Discovery, Design and Development	4	4	0	8	4	2	0	6	20	40	40	100	50	150
				(C. Discip	oline Spe	ecific Ge	neric Ele	ectives	(Any Tw	vo)		•	•		
4	BSMB603	Food & Dairy Microbiology	4	4	0	8	4	2	0	6	20	40	40	100	50	150
5	BSMB604	Biosafety and Intellectual Property Rights	4	4	0	8	4	2	0	6	20	40	40	100	50	150
6	BSMB605	Environmental Microbiology	4	4	0	8	4	2	0	6	20	40	40	100	50	150
7	BSMB606	Advances in Microbiology	4	4	0	8	4	2	0	6	20	40	40	100	50	150
		Total								26						750

Note: L = Lecture, P = Practice, T = Tutorial, MS - Mid Semester, CEC - Continuous Evaluation Component, ES - End Semester



COURSE CODE	COURSE NAME	SEMESTER
AECC601	INDIAN CONSTITUTION	VI

Teaching Scheme (Hours)				Teaching Credit				
Lecture	Practical	Tutorial	Total Hours	Lecture Practical Tutor		Tutorial	Total Credit	
30	0	0	30	2	0	0	2	

Course Pre-requisites	10 +2 (With Arts/Science/Commerce)			
Course Category	Ability Enhancement Compulsory Course.			
Course focus	Skill development			
Rationale	The fundamental concepts of Indian Constitution help in understanding the role of government and framework of Indian Constitution. This subject also enables the students to understand the administrative organizational structure of India.			
Course Revision/ Approval Date:	14/03/2020			
Course Objectives	To enable the student:			
(As per Blooms' Taxonomy)	 To understand Indian Constitution. To know the framework of Indian Constitution. To aware role of government of the union. To aware role of the state government. To understand administration organization. 			

Course Content (Theory)	Weightage	Contact hours
Unit 1: Constitution – Strategies and Principles	20%	6
1. Meaning and important of constitution	20%	0
2. Making of Indian constitution – sources		
3. Salient Features of Indian constitution.		
Unit 2: Fundamental Rights and Directive Principles		_
1. Fundamental Rights	20%	6
2. Fundamental Duties		
3. Directive Principles		



		0.0000
Unit 3: Government of the Union	200/	
1. President of India – Election and powers	20%	6
2. Prime Minister and council of ministers		
3. Lok Sabha – composition and Powers		
4. Rajya Sabha – Composition and Powers.		
Unit 4: Government of the States & The Judiciary		
1. Governor – Powers		
2. Chief Minister and Council of ministers	20%	6
3. Legislative Assembly – Composition and Powers		
4. Legislative Council – Composition and Powers		
5. Features of judiciary system in India		
6. Supreme Court – Structure and Jurisdiction.		
Unit 5: Administrative Organization and Constitution		
1. Federalism in India – features		
2. Local Government – Panchyats and Powers and functions 73rd and	20%	6
74th Amendments		
3. Election Commission - Organization and functions		
4. Citizen Oriented Measure - RTI and PIL - Provisions and		
Significance.		

Utilizing models, PowerPoint Presentations, group discussions, and seminars are some of the methods adopted to improve the student's ability to grasp the principles of environmental studies.

Course Outcome:	Blooms' Taxonomy Domain	Blooms' Taxonomy Subdomain
After successfulcompletion of the above course	e, students will be able	to:
CO1: Analyse importance of Indian constitution	Understand	Define, Classify & Describe
CO2: Analyse importance of Indian constitution	Apply	Define, Classify, Describe, Demonstrate & Examine
CO3: Know powers of state and union government.	Analyses and Evaluation	Define, Classify, Describe & Demonstrate



I-								
CO4: U	nderstand administration of Indian Constitution		Define, Classify,					
			Describe,					
			Demonstrate &					
		Apply & analyse	Examine					
CO5: U	nderstand administration of Indian Constitution.	Remember &	Define, Describe &					
		apply	Demonstrate					
	Learning Resources							
1.	Reference/Tex	t Books:						
	1. Indian's Constitution by M.V. Pylee, New Delhi S. Chand Publication							
	2. The Constitutional Law of India by J.N. Pand	ay Allahabad Central	Law Agency					
	3. Constitution of India by National Portal of India	•	,					
	4. https://www.india.giv.in/sites/upload-files/coi_	part_full.pdf.						
2.	Journals & Per	riodicals:						
	1. Constitution of India							
	1. Constitution of findia							
	2. National Portal of India.							
3.	3. Other Electronic Resources:							
	1. https://legislative.gov.in/constitution-of-india							
	1 5							

Evaluation Scheme	Total Marks				
Theory: Midsemester Marks	20 marks				
Theory: End Semester Marks	40 marks				
Theory: Continuous					
Evaluation Component Marks	Attendance	05 marks			
	MCQs	10 marks			
	Open Book Assignment	15 marks			
	Article Review	10 marks			
	Total	40 Marks			

	PSO1	PSO2	PSO3	PSO4	PSO5	PSO6
CO1	0	1	0	3	0	0
CO2	0	3	0	3	0	1
CO3	0	2	0	3	0	0



CO4	0	2	0	3	0	0
CO5	0	2	0	3	0	1

	PO1	PO2	PO3	PO4	PO5	PO6
CO1	0	0	0	2	3	1
CO2	0	1	1	2	3	1
CO3	0	1	0	2	3	3
CO4	0	0	0	2	3	3
CO5	0	0	0	2	3	1



COURSE CODE	COURSE NAME	SEMESTER
BSMB601	IMMUNOLOGY	VI

Teaching Scheme (Hours)					ching edit		
Lecture	Practical	Tutorial	Total Hour s	Lecture	Practical	Tutorial	Total Credit
6	60	0	120	4	2	0	6

Course Pre-requisites	Students should have basic knowledge about Immunity.
Course Category	Core Professional.
Course focus	Employability
Rationale	To have an overview of Immune response and it's components. The subject also explains the regulation of immunoglobulin gene, major histocompatibility complexes, vaccines and vaccine development and immunodiagnostics.
Course Revision/ Approval Date:	14/03/2020
Course Objectives (As per Blooms' Taxonomy)	 Remember Outline, compare and contrast the key mechanisms and cellular players of innate and adaptive immunity. Apply Elucidate the genetic basis for immunological diversity and the generation of adaptive immune responses. Analyses Outline key events and cellular players in antigen presentation, and how the nature of the antigen will shape resulting effectors responses. Create Understand and explain the basis of allergic diseases and immunodeficiencies related diseases. Understand The principles governing vaccination and the mechanisms of protection against disease.

Course Content (Theory)	Weightage	Contact hours
Unit 1: Immune Response Theory: An overview, components of mammalian immune system, molecular structure of Immunoglobulins or Antibodies. Humoral & Cellular immune responses, T- lymphocytes & immune response (cytotoxic T-cell, helper T-cell, suppressor T-cells), Tcell receptors, genome rearrangements during Blymphocyte differentiation, Antibody affinity maturation class switching, assembly of T-cell receptor genes by somatic recombination.	20%	12



Unit 2: Regulation of immunoglobulin gene expression Theory: Clonal selection theory, allotypes & idiotypes, allelic exclusion, immunologic memory, heavy chain gene transcription, genetic basis of antibody diversity, hypotheses (germ line & somatic mutation), antibody diversity.	20%	12
Unit 3: Regulation of immunoglobulin gene expression Theory: Clonal selection theory, allotypes & idiotypes, allelic exclusion, immunologic memory, heavy chain gene transcription, genetic basis of antibody diversity, hypotheses (germ line & somatic mutation), antibody diversity.	20%	12
Unit 4: Vaccines & Vaccination Theory: Adjuvants, cytokines, DNA vaccines, recombinant vaccines, bacterial vaccines, viral vaccines, vaccines to other infectious agents, passive & active immunization.	20%	12
Unit 5:Immunodiagnostics Theory: Introduction to immunodiagnostics – RIA, ELISA. etc	20%	12

List Of Practical	Weightage	Contact hours
1: Differential leucocytes count	10%	4
2: Total leucocytes count	10%	4
3: Total RBC count	10%	4
4: Separation of serum from blood	10%	4
5: Hemagglutination assay	10%	4
6: Hemagglutination inhibition assay	10%	4
7: Latex Agglutination Assay	10%	4
8: Single radial ImmunoAssay	10%	4
9: Double immunodiffusion test using specific antibody and antigen.	10%	4
10: Dot ELISA	10%	4
11: Sandwich ELISA	10%	4



Course Outcomes:	Blooms' Taxonomy Domain	Blooms' Taxonomy Sub Domain
After successful completion of the above course, students will be able to: CO1 Outline, compare and contrast the key mechanisms and cellular players of innate and adaptive immunity.	Remember	Explain, Describe, Discuss, Recall, Locate
CO2 Elucidate the genetic basis for immunological diversity and the generation of adaptive immune responses.	Apply	Apply, Practice, Interpret, Select, Correlate
CO3 Outline key events and cellular players in antigen presentation, and how the nature of the antigen will shape resulting effector responses	Analyses and Evaluation	Compare, Classify, Select, Investigate
CO4 Understand and explain the basis of allergic diseases and immunodeficiencies related diseases.	Create	Construct, Develop, Produce
CO5 Understand the principles governing vaccination and the mechanisms of protection against disease.	Understand	Explain, Describe, outline, Predict, Summarize

	Learning Resources
1.	Reference books:
	 Delves P, Martin S, Burton D, Roitt IM. (2006). Roitt's Essential Immunology. 11th edition Wiley- Blackwell Scientific Publication, Oxford. Abbas AK, Lichtman AH, Pillai S. (2007). Cellular and Molecular Immunology. 6 th edition Saunders Publication, Philadelphia. Murphy K, Travers P, Walport M. (2008). Janeway's Immunobiology. 7th edition Garland Science Publishers, New York. Peakman M, and Vergani D. (2009). Basic and Clinical Immunology. 2nd edition Churchill Livingstone Publishers, Edinberg. Richard C and Geiffrey S. (2009). Immunology. 6th edition. Wiley Blackwell Publication. Textbook: 1. Goldsby RA, Kindt TJ, Osborne BA. (2007). Kuby's Immunology. 6th edition W.H. Freemanand Company, New York
2.	Journals & Periodicals
2.	1. Journal of Immunology
	2. Molecular Immunology
	Nature Review immunology The Scientist
5	Other Electronic resources: https://www.immunology.org/



Evaluation Scheme	Total Marks	
Theory: Mid semester Marks	20 marks	
Theory: End Semester Marks	40 marks	
Theory: Continuous Evaluation Component Marks	Attendance MCQs Open Book Assignment	05 marks 10 marks 15 marks
Practical Marks	Article Review Total	10 marks 40 Marks
Fractical Warks	Attendance Practical Exam Viva	05 marks 30 marks 10 marks
	Journal Total	05 marks 50 Marks

	PSO1	PSO2	PSO3	PSO4	PSO5	PSO6
CO1	2	1	0	1	1	0
CO2	1	1	1	2	0	0
CO3	1	0	0	1	2	1
CO4	2	0	2	2	2	2
CO5	2	1	0	1	0	2

Mapping of POs & COs

ping of t Os						
	PO1	PO2	PO3	PO4	PO5	PO6
CO1	3	2	0	2	2	1
CO2	2	1	1	2	0	0
CO3	2	0	0	1	2	1
CO4	2	1	2	3	2	2
CO5	2	1	0	1	0	3



COURSE CODE	COURSE NAME BIOINFORMATICS	SEMESTER
BSMB602	AND INTRODUCTION TO DRUG	VI
	DISCOVERY, DESIGN AND	
	DEVELOPMENT	

Teaching Scheme (Hours)					ching edit		
Lecture	Practical	Tutorial	Total Hour s	Lecture Practical Lutorial			Total Credit
60	60	0	120	4	2	0	6

Course Category	Students should have basic knowledge about fundamental theories and practices of bioinformatics and it will also provide an overview of the drug development process, focusing on drug development science, regulation. Core Professional.		
Course focus	Employability		
Rationale	To have an overview of the fundamental theories and practices of bioinformatics and an overview of the drug development process, focusing on drug development science, regulation.		
Course Revision/ Approval Date:	14/03/2020		
Course Objectives (As per Blooms' Taxonomy)	 Remember provide an overview of the drug development process, focusing on drug development science, regulation. Apply to introduce the fundamental theories and practices of bioinformatics, which is an interdisciplinary thrust area merging Biology and Information Technology. Analyses Safety evaluation, bioavailability, clinical trials, and the essentials of patent law. Create drug development as a process involving target selection, lead discovery using computer-based methods and combinatorial chemistry/high-throughput screening. Understand Understand biological databases, analysis of macromolecular Sequences and phylogenetic analysis. 		



Course Content (Theory)	Weightage	Contact hours
Unit 1: Basic Computer- MS-WORD, EXCEL, Microsoft PowerPoint, Adobe Photoshop Introduction to Bioinformatics: Introduction to Bioinformatics, Introduction to biological macromolecules, History of Bioinformatics, Scope of Bioinformatics Introduction to Biological databases, Nucleic acid databases: Introduction to biological databases in general, Classification- Primary, Secondary, Composite databases, Flat files in databases, various file formats, FASTA, GENBANK, Nucleic acid sequence databases- GENBANK, EMBL, DDBJ etc.	20%	12
Unit 2: Protein databases and specialized databases: Protein sequence and structure databases – UNIPROT, PIR, SWISS-PROT, PDB etc., Other databases of patterns/ motifs/ metabolic pathways/ Immunology/genetic disorders etc, Composite databases like NCBI, Data base searching, Bibliographic databases like PUBMED	20%	12
Unit 3: Genome databases and search protocol: Genome databases at NCBI, SANGER, TIGR etc.Secondary database search protocol, Use of Boolean operators	20%	12
Unit 4: Sequence alignment: Pairwise alignment techniques Global alignment, Local alignment, pairwise alignment techniques- continued, Significance of alignment- Z-score, P-score, E-value, Multiple sequence alignment-Dynamic programing Sequence alignment, BLAST analysis: Multiple sequence alignment techniques- continued- profile, Hidden Markov Model, BLAST analysis, E-value, Different types of BLAST, PSI-BLAST, PHI-BLAST.	20%	12
Unit 5: Introduction to Drug Discovery, Design and Development: Introduction to Drug Design and Development, Drug targets, Lead Identification and Modification, Computer-Aided Drug Design, Drug Delivery, Pre-clinical and Clinical Testing Overview and career opportunities	20%	12

List Of Practical	Weightage	Contact hours
1: Use of various tools in NCBI and Familiarizing with various databases like PROSITE, PUBMED, OMIM, KEGG etc.	20%	12
2: Retrieval of nucleotide and protein sequences from databases like GENBANK, UNIPROT etc and Homology modeling using SWISS-PROT	20%	12
3: Familiarizing with genome specific databases like TAIR, GENE db. Etc.	20%	12
4: Sequence alignment- Pairwise and Multiple Sequence alignment and Phylogeny analysis BLAST analysis and Variations in BLAST analysis.	20%	12



5: Use of visualizing software like PYMOL, CHIMERA and Using various web-	20%	12
based software for primer designing, sequence and structure analysis like		
PROTPARAM, PROCHECK etc.		

Course Outcomes:	Blooms' Taxonomy Domain	Blooms' Taxonomy Sub Domain
After successful completion of the above course, students will be able to: CO1 Basic Computer skills.	Apply	Explain, Describe, Discuss, Recall, Locate
CO2 Acquire knowledge about various biological databases and how to retrieve and use data from these databases.	Understand	Apply, Practice, Interpret, Select, Correlate
CO3 Understand the concepts involved in sequence alignment and phylogeny Analysis.	Remember	Compare, Classify, Select, Investigate
CO4 Be able to describe the process of drug discovery and development.	Create	Construct, Develop, Produce
CO5 Be able to discuss the challenges faced in each step of the drug discovery process.	Analyses and Evaluation	Explain, Describe, outline, Predict, Summarize

	Learning Resources			
1.	Reference books:			
	 Mount DW Cold. 2001. Bioinformatics: Sequence and Genome Analysis. Spring. Attwood TK & Parry-Smith DJ. 2003. Introduction to Bioinformatics. Pearson Education. 			
	 Rastogi SC, Mendiratta N & Rastogi P. 2004. Bioinformatics: Concepts, Skillsand Applications. CBS Introduction to Bioinformatics. 2014- Arthur M Lesk Oxford University Press Drug Discovery and Development; Technology in Transition. HP Rang. Elsevier Ltd 1 st edition2006. 			



2.	Journals & Periodicals https://www.longdom.org/scholarly/drug-delivery-using-nanotechnology-journals-articles-ppts-list-747.html
3.	Other Electronic resources: 1. https://www.ebi.ac.uk/training/online/course/bioinformatics-terrified-0 2. https://www.ebi.ac.uk/training/online/course/biomedical-data-ethical-legal- and-social-implication 3. https://www.ebi.ac.uk/training/online/course/bringing-data-life-data-management-biomolecular-sciences. 4. https://www.ebi.ac.uk/training/online/course/chebi-quick-tour 5. https://www.ebi.ac.uk/training/online/course/chebi-online-chemical-dictionary-small-molecules 6. http://www.angelfire.com/ga2/nestsite2/bioinform.html 7. https://pubs.acs.org/doi/full/10.1021/acs.jchemed.6b00596 8. https://pharmafactz.com/tutorials-and-guides/

Evaluation Scheme	Total Marks	
Theory: Mid semester Marks	20 marks	
Theory: End Semester Marks	40 marks	
Theory: Continuous Evaluation Component Marks	Attendance	05 marks
	MCQs Open Book Assignment	10 marks 15 marks
	Article Review Total	10 marks 40 Marks
Practical Marks		
	Attendance Practical Exam	05 marks 30 marks
	Viva	10 marks
	Journal Total	05 marks 50 Marks
	2000	ov Manns

	PSO1	PSO2	PSO3	PSO4	PSO5	PSO6
CO1	1	0	1	2	1	1
CO2	2	1	1	1	1	0
CO3	0	1	1	2	2	1
CO4	1	0	0	2	1	2
CO5	0	2	2	0	1	1



	PO1	PO2	PO3	PO4	PO5	PO6
CO1	2	2	1	2	1	1
CO2	1	1	0	3	0	0
CO3	2	1	0	2	1	1
CO4	1	2	1	2	0	0
CO5	2	1	2	3	0	2



COURSE CODE COURSE NAME SEMESTER
BSMB603 FOOD AND DAIRY VI
MICROBIOLOGY

Teaching Scheme (Hours)				Teachin	g Credit		
Lecture	Practical	Tutorial	Total Hours	Lastura Prostusal Tutorial		Total Credit	
4	4	0	8	4	2	0	6

Course Pre-requisites	Students should have basic knowledge of food and dairy microbiology.		
Course Category	Core Professional.		
Course focus	Employability		
Rationale	To have an overview of the basic food and dairy microbiology.		
Course Revision/Approval Date:	14/03/2020		
Course Objectives	Remember Developed a clear understanding of the multifarious		
(As per Blooms' Taxonomy)	roles of microorganisms in soil, in association with plants and thus		
	in the field of agriculture.		
	Apply Are able to describe the role of microorganisms in the production of food, its spoilage, including their role in homemade fermented foods. Analyses Are able to identify the role of microorganisms in the causation of the diseases and how to protect against food-borne pathogens.		
	eate Developed experimental skills for testing the milk and different foods for the presence of microorganisms.		
	iderstand To develop Understanding in dairy sciences.		

Course Content (Theory)	Weightage	Contact
		hours
Unit 1: Foods as a substrate for microorganisms. Intrinsic and extrinsic factors that affect growth and survival of microbes in foods, natural flora and source of contamination of foods in general. Microbial spoilage of various foods. Principles, Spoilage of vegetables, fruits, meat, eggs, milk and butter, bread, canned Foods.	20%	12
Unit 2: Principles, physical methods of food preservation: temperature (low, high, canning, drying), irradiation, hydrostatic pressure, high voltage pulse, microwave processing and aseptic packaging, chemical methods of food preservation: salt, sugar, organic acids, SO2, nitrite and nitrates, ethylene oxide, antibiotics and bacteriocins.	20%	12



Unit 3: Dairy starter cultures, fermented dairy products: yogurt, acidophilus milk, kumiss, kefir, dahi and cheese, other fermented foods:	20%	12
dosa, sauerkraut, soy sauce and tampeh, Probiotics: Health benefits, types		
of microorganisms used; probiotic foods available in market		
Unit 4: Food intoxications: Staphylococcus aureus, Clostridium	20%	12
botulinum and mycotoxins; Food infections: Bacillus cereus, Vibrio		
parahaemolyticusEscherichia coli, Salmonellosis, Shigellosis, Yersinia		
enterocolitica, Listeria monocytogenes and Campylobacter jejuni.		
Unit 5: HACCP, Indices of food sanitary quality and sanitizers Cultural	20%	12
and rapid detection methods of food borne pathogens in foods and		
introduction to predictive microbiology		

List Of Practical	Weightage	Contact hours
1. MBRT of milk samples and their standard plate count.	20%	12
2. Alkaline phosphatase test to check the efficiency of pasteurization of milk.		
3. Isolation of any food borne bacteria from food products.	20%	12
4. Isolation of spoilage microorganisms from spoiled vegetables/fruits.	20%	12
5. Isolation of spoilage microorganisms from bread.	20%	12
6. Preparation of Yogurt/Dahi.	20%	12

	Course Objectives:	Blooms' Taxonomy Domain	Blooms' Taxonomy Sub Domain
CO1	Developed a clear understanding of the multifarious roles of microorganisms in soil, in association with plants and thus in the field of agriculture.	Remember	Explain, Describe, Discuss, Recall, Locate
CO2	Are able to describe the role of microorganisms in the production of food, its spoilage, including their role in homemade fermented foods.	Apply	Apply, Practice, Interpret, Select, Correlate
CO3	Are able to identify the role of microorganisms in the causation of the diseases and how to protect against foodborne pathogens.	Analyses	Compare, Classify, Select, Investigate
CO4	Developed experimental skills for testing the milk and different foods for the presence of microorganisms.	Create	Construct, Develop, Produce
CO5	To develop Understanding in dairy sciences.	Understand	Explain, Describe, outline, Predict, Summarize



sources
Reference books:
1. Adams MR and Moss MO. (1995). Food Microbiology. 4th edition, New Age
International (P) Limited Publishers, New Delhi, India.
2. Banwart JM. (1987). Basic Food Microbiology. 1st edition. CBS Publishers
and Distributors, Delhi, India.
3. Davidson PM and Brannen AL. (1993). Antimicrobials in Foods. Marcel
Dekker, New York.
4. Dillion VM and Board RG. (1996). Natural Antimicrobial Systems and
Food Preservation. CAB International, Wallingford, Oxon.
5. Frazier WC and Westhoff DC. (1992). Food Microbiology. 3rd edition. Tata
McGraw-Hill Publishing Company Ltd, New Delhi, India.
6. Gould GW. (1995). New Methods of Food Preservation. Blackie Academic
and Professional, London.
7. Jay JM, Loessner MJ and Golden DA. (2005). Modern Food Microbiology.
7th edition, CBS Publishers and Distributors, Delhi, India.
8. Lund BM, Baird Parker AC, and Gould GW. (2000). The Microbiological
Safety and Quality of Foods. Vol. 1-2, ASPEN Publication, Gaithersberg, MD.
9. Tortora GJ, Funke BR, and Case CL. (2008). Microbiology: An Introduction.
9th edition. Pearson Education.
Journals & Periodicals
1. Journal of Food Science.
2. Current opinion on Food Sciences.
Other Electronic resources:
1. NPTEL

Evaluation Scheme	Total Marks	
Theory: Mid semester Marks	20 marks	
Theory: End Semester Marks	40 marks	
Theory: Continuous Evaluation Component Marks	Attendance MCQs Open Book Assignment Article Review Total	05 marks 10 marks 15 marks 40 Marks



Practical Marks		
	Attendance	05 marks
	Practical Exam	30 marks
	Viva	10 marks
	Journal	05 marks
	Total	50 Marks

	PSO1	PSO2	PSO3	PSO4	PSO5	PSO6
CO1	2	1	2	1	1	1
CO2	2	1	2	1	1	0
CO3	2	2	3	1	1	2
CO4	2	2	3	1	1	1
CO5	2	1	2	1	1	1

Mapping of POs & COs

8	PO1	PO2	PO3	PO4	PO5	PO6
CO1	3	0	2	1	2	1
CO2	1	1	1	1	2	0
CO3	3	1	3	1	1	0
CO4	3	1	3	1	1	0
CO5	1	1	2	1	1	0



COURSE CODE Course Name SEMESTER
BSMB604 BIOSAFETY, BIOETHICS VI
AND INTELLECTUAL
PROPERTY RIGHTS

Teaching Scheme (Hours)					Teachin	g Credit	
Lecture	Practical	Tutorial	Total Hours	Lecture Practical Tutorial			
4	4	0	8	4	2	0	6

Course Pre-requisites	Students should have basic knowledge of biosafety regulations,
	bioethics and Intellectual Property Rights (IPRs).
Course Category	Discipline Specific Course.
Course focus	Employability
Rationale	To have an overview of the basic knowledge of biosafety
	regulations, bioethics and Intellectual Property Rights (IPRs)
Course Revision/Approval	14/03/2020
Date:	
Course Objectives	Remember To understand various aspects of biosafety regulations,
(As per Blooms' Taxonomy)	bioethics used in biological materials in the laboratory in context of
	environment and human health and Intellectual Property Rights
	concerns arising from the commercialization of biotech products.
	Apply Students will gain awareness about Intellectual Property
	Rights (IPRs) to take measure for the protection of their ideas.
	alyses They will be able to devise business strategies by taking
	account of IPRs.
	eate They will be able to assist in technology upgradation and
	enhancing competitiveness.
	iderstand They will acquire adequate knowledge in the use of
	genetically modified organisms and its effect on human health.

Course Content (Theory)	Weightage	Contact
Unit 1: Introduction of Biosafety in the laboratory, General principles for the laboratory and environmental biosafety, Biosafety guidelines, Biosafety levels, laboratory biosecurity concepts, Regulatory framework; National and international biosafety policies and law.	20%	hours 12
Unit 2: Health aspects of Biosafety; toxicology, allergenicity, antibiotic resistance, Risk management issues and containment. Radiation safety and non-radio isotopic procedure.	20%	12
Unit 3: Good microbiological techniques and good laboratory practice (GLP), Basic principles of quality control (QA) and quality assurance (QC), Guidelines for QA and QC: raw materials, sterilization, media, stock cultures and products, Role of culture collection centre, public health laboratories and regulatory agencies.	20%	12
Unit 4: Concept of Intellectual properties, copyrights, trademarks, trade secrets, patents, geographical indications.	20%	12



Unit	5:	Implications	of	intellectual	property	rights	s on	the	20%	12
commer	cializat	ion of biotecl	hnolo	gy products.	National	and	intern	ational		
patent a	ct and	amendments,	pater	t filing.						

List Of Practical	Weightage	Contact hours
1: Understanding of the importance of good laboratory management and practices	20%	12
2: Lab SOPs/Protocols management.	20%	12
3: Lab QA/QC.	20%	12

Instructional Method and Pedagogy:
Audio-Visual Lectures, Quizzes, Debates, Project works, Case studies, and Assignments Practical exercises are designed to understand the theory as taught in classroom. Hands on in practical session.

	Course Objectives:	Blooms' Taxonomy Domain	Blooms' Taxonomy Sub Domain
CO1	To understand various aspects of biosafety regulations, bioethics used in biological materials in the laboratory in context of environment and human health and Intellectual Property Rights concerns arising from the commercialization of biotech products.	Remember	Explain, Describe, Discuss, Recall, Locate
CO2	Students will gain awareness about Intellectual Property Rights (IPRs) to take measure for the protection of their ideas.	Apply	Apply, Practice, Interpret, Select, Correlate
CO3	They will be able to devise business strategies by taking account of IPRs.	Analyses	Compare, Classify, Select, Investigate
CO4	They will be able to assist in technology upgradation and enhancing competitiveness	Create	Construct, Develop, Produce
CO5	They will acquire adequate knowledge in the use of genetically modified organisms and its effect on human health.	Understand	Explain, Describe, outline, Predict, Summarize

Learning Resources



	and the second s
1.	Reference books: 1. Mike Martin and Roland Schinzinger, "Ethics in Engineering", McGraw-Hill, New York 1996.
	2. Govindarajan M, Natarajan S, Senthil Kumar V. S, "Engineering Ethics", Prentice Hall of India, New Delhi, 2004.
	3. Biosafety in Microbiological and Biomedical Laboratories (BMBL) 5th Edition
	4. Richardson JH, Barkley WE, editors. Biosafety in microbiological and biomedical laboratories. 1st ed.Washington, DC. 1984.
	5. Harding AL, Byers KB. Epidemiology of laboratory-associated infections.
	6. Fleming DO, Hunt DL, editors. Biological safety: principles and practices. 3rd ed. Washington, DC: ASM Press; 2000:35-54.
	7. Bodenheimer, Jurisprudence – The Philosophy and Method of Law (1996) Universal, Delhi.
	8. Fitzgerald, (ed.) Salmond on Jurisprudence (1999) Tripathi, Bombay.
	9. Singh BD. 2007. Biotechnology: Expanding Horizon. Kalyani.
2.	Journals & Periodicals
	1. https://www.researchgate.net/publication/314667237_Biosafety_Bioethics_and_IP R_Issues_in_Plant_Biotechnology.
	2. http://www.jnkvv.org/PDF/09042020221517Syllabus%20(11%20files%20merged). pdf
5	Other Electronic resources:
	1. https://online-learning.harvard.edu/course/fda-and-prescription-drugs-current-controversiescontext delta=0
	2. https://www.oreilly.com/library/view/ipr-biosafet
	and/9788131774700/xhtml/chapter017.xhtml
	3. http://www.biotech.iastate.edu/bioethics-case-studies/

Evaluation Scheme	Total Marks	
Theory: Mid semester	20 marks	
Marks		
Theory: End Semester	40 marks	
Marks		
Theory: Continuous		
Evaluation Component Marks	Attendance	05 marks
Marks	MCQs	10 marks
	Open Book Assignment	15 marks
	Article Review	10 marks
	Total	40 Marks



Practical Marks		
	Attendance	05 marks
	Practical Exam	30 marks
	Viva	10 marks
	Journal	05 marks
	Total	50 Marks

	PSO1	PSO2	PSO3	PSO4	PSO5	PSO6
CO1	1	1	1	1	0	1
CO2	2	2	1	2	1	2
CO3	2	1	2	1	2	1
CO4	1	1	2	2	2	2
CO5	1	2	1	1	1	2

Mapping of POs & COs

	PO1	PO2	PO3	PO4	PO5	PO6
CO1	2	1	1	2	2	2
CO2	1	2	2	2	0	3
CO3	1	0	0	1	2	1
CO4	2	1	2	3	2	2
CO5	2	2	1	1	1	1



COURSE CODE	COURSE NAME	SEMESTER
BSMB605	ENVIRONMENTAL	VI
	MICROBIOLOGY	

Teaching Scheme (Hours)				Teachin	g Credit		
Lecture	Practical	Tutorial	Total Hours	Lecture	Practical	Tutorial	Total Credit
60	60	0	120	4	2	0	6

Course Pre-requisites	Students should have basic knowledge about environmental science and students of all disciplines can think critically, ethically, and creatively when evaluating environmental issues		
Course Category	Discipline Specific Course.		
Course focus	Employability		
Rationale	To have an overview of the environmental science and students of all disciplines can think critically, ethically, and creatively when evaluating environmental issues.		
Course Revision/Approval Date:	14/03/2020		
Course Objectives (As per Blooms' Taxonomy)	Remember To introduce environmental microbiology and its scope Apply To understand the role of microbes in biogeochemistry		
	Analyses To understand various modes of biotic interaction of microbes		
	Create Toget insights of role of microbes in pollution control		
	Understand To get acquainted with applied aspects of environmental microbiology		

Course Content (Theory)	Weightage	Contact hours
Unit 1: Microorganisms and their Habitats: Structure and function of ecosystems	20%	12
Terrestrial Environment: Soil profile and soil microflora Aquatic Environment:		
Microflora of fresh water and marine habitats Atmosphere: Aero microflora and		
dispersal of microbes Animal Environment: Microbes in/on human body		
(Microbiomes) & animal (ruminants) body. Extreme Habitats: Extremophiles:		
Microbes thriving at high & low temperatures, pH, high hydrostatic & osmotic		
pressures, salinity, & low nutrient levels	200/	10
Unit 2: Microbial Interactions Microbe interactions: Mutualism, synergism,	20%	12
commensalism, competition, amensalism, parasitism, predation Microbe-Plant interaction: Symbiotic and non-symbiotic interactions Microbe-animal		
interaction: Symbiotic and horisymbiotic interactions wherooc-animal interaction: Microbes in ruminants, nematophagus fungi and symbiotic		
luminescent bacteria		
Unit 3: Biogeochemical Cycling Carbon cycle: Microbial degradation	20%	12
of cellulose, hemicelluloses, lignin and chitin Nitrogen cycle: Nitrogen		
fixation, ammonification, nitrification, denitrification and nitrate		
reduction Phosphorus cycle: Phosphate immobilization and		
solubilisation Sulphur cycle: Microbes involved in sulphur cycle Other		
elemental cycles: Iron and manganese.		



Unit 4: Waste Management Solid Waste management: Sources and types of solid waste, Methods of solid waste disposal (composting and sanitary landfill) Liquid waste management: Composition and strength of sewage (BOD and COD), Primary, secondary (oxidation ponds, trickling filter, activated sludge process and septic tank) and tertiary sewage treatment	20%	12
Unit 5: Microbial Bioremediation and Water Potability: Principles and degradation of common pesticides, hydrocarbons (oil spills). Treatment and safety of drinking (potable) water, methods to detect potability of water samples: (a) standard qualitative procedure: presumptive test/MPN test, confirmed and completed tests for fecal coliforms (b) Membrane filter technique and (c) Presence/absence tests.	20%	12

List Of Practical	Weightage	Contact hours
1. Analysis of soil - pH, moisture content, water holding capacity	10%	4
2. Determination of Chlorine in Water	10%	4
3. Finding MPN (Most Probable Number) of the water sample	10%	4
4. Measure mineral and phenolphthelin Acidity	10%	4
5. Study the presence of microbial activity by detecting	10%	4
(qualitatively) enzymes (dehydrogenase, amylase, urease) in soil.		
6. Isolation of Rhizobium from root nodules	10%	4
7. Isolation of microbes (bacteria & fungi) from rhizosphere and	10%	4
8. rhizoplane.		
9. Find Dissolved Oxygen (DO) of the water sample	10%	4
10. Find Chemical Oxygen Demand (COD) of the water sample	10%	4
11. Find Biological Oxygen Demand (BOD) of the water sample	10%	4
12. Perform Presumptive test and Confirm test for Coliform	10%	4

	Course Objectives:	Blooms' Taxonomy Domain	Blooms' Taxonomy Sub Domain
CO1	Student will be able to demonstrate microbial diversity	Apply	Explain, Describe, Discuss, Recall, Locate
CO2	The students will be updated interaction of microbes with other organisms	Remember	Apply, Practice, Interpret, Select, Correlate
CO3	The students will be updated with general concepts role of microbes in biogeochemical cycling	Analyses	Compare, Classify, Select, Investigate
CO4	The students gain knowledge on general characteristics of fungi, bacteria, virus and mycoplasma like organisms causing plant diseases.	Create	Construct, Develop, Produce



Students will be acquainted with reproduction in fungi and fungal like organisms causing plant diseases.

Understand

Explain, Describe, outline, Predict,

Summarize

	Learning Resources
1.	Reference books:
1.	I.L. Pepper and C.P. Gerba (2004) Environmental Microbiology A Laboratory Manual, Elsevier/Academic Press
	 Christon J. Hurst (eds.) (2016) The Mechanistic Benefits of Microbial Symbionts, Springer International Publishing
	3. Hurst, Christon J.; Crawford, Ronald L.; Garland, Jay L.; Lipson, David A.; Mills, Aaron L.; Stetzenbach, Linda D. (Eds.) (2007) Manual of Environmental Microbiology, American Society for Microbiology
	4. Myung-Bo Kim eds. (2008) Progress in Environmental Microbiology, Nova Biomedical Books New York
	5. Moo-Young, M., Anderson, W. A., & Chakrabarty, A. M. (Eds.). (2013). Environmental biotechnology: principles and applications. Springer Science & Business Media.
	6. Lynch JM & Hobbie JE. (1988). Microorganisms in Action: Concepts & Application in Microbial Ecology. Blackwell Scientific Publication, U.K.
	7. Martin A. (1977). An Introduction to Soil Microbiology. 2nd edition. John Wiley & Sons Inc. New York & London.
	8. Stolp H. (1988). Microbial Ecology: Organisms Habitats Activities. Cambridge University Press, Cambridge, England.
	9. Subba Rao NS. (1999). Soil Microbiology. 4th edition. Oxford & IBH Publishing Co. New Delhi.
	10. Willey JM, Sherwood LM, and Woolverton CJ. (2013). Prescott's Microbiology. 9th edition. McGraw Hill Higher Education
2	Journals & Periodicals
	Applied and Environmental Microbiology Gritical Paviance in Microbiology
	Critical Reviews in Microbiology Nature Reviews Microbiology
	4. Nature Microbiology
	5. Microbiology
	6. BMC Microbiology
	7. Trends in Microbiology
	8. Gavrilescu, Maria. "Environmental biotechnology: achievements, opportunities and
	challenges." Dynamic biochemistry, process biotechnology and molecular biology 4.1
	(2010): 1-36.
	9. Verstraete, Willy, and Eva Top. "Holistic environmental biotechnology." Microbial control of pollution. (1992): 1-17.
	10. Grommen, Roeland, and Willy Verstraete. "Environmental biotechnology: the ongoing quest." Journal of Biotechnology 98.1 (2002): 113-123.
	11. Michalak, Izabela. "The application of seaweeds in environmental biotechnology."
	Advances in Botanical Research. Vol. 95. Academic Press, 2020. 85-111.
	12. Kalogerakis, Nicolas, et al. "The role of environmental biotechnology in exploring,
	exploiting, monitoring, preserving, protecting and decontaminating the marine environment." New biotechnology 32.1 (2015): 157-167.



3	Other Electronic resources					
	1. https://sfam.org.uk/					
	2. https://www.isme-microbes.org/					
	3. https://www.asmscience.org/VisualLibrary					
	4. https://microbe.net/resources/microbiology-web-resources/					
5. https://www.epa.gov/						
	6. https://microbiologyonline.org/teachers/resources					

Evaluation Scheme	Total Marks	
Theory: Mid semester Marks	20 marks	
Theory: End Semester Marks	40 marks	
Theory: Continuous Evaluation Component Marks	Attendance MCQs Open Book Assignment Article Review Total	05 marks 10 marks 15 marks 10 marks 40 Marks
Practical Marks	Attendance Practical Exam Viva Journal Total	05 marks 30 marks 10 marks 05 marks 50 Marks

	PSO1	PSO2	PSO3	PSO4	PSO5	PSO6
CO1	1	1	1	2	1	2
CO2	2	1	1	1	1	1
CO3	2	1	1	2	2	1
CO4	1	1	2	2	1	2
CO5	0	2	2	0	1	2

Mapping of POs & COs

	PO1	PO2	PO3	PO4	PO5	PO6
CO1	0	0	1	1	0	1
CO2	1	1	0	1	0	0
CO3	1	1	0	2	1	1
CO4	1	0	1	2	1	1
CO5	0	1	0	1	1	2



COURSE CODE	COURSE NAME	SEMESTER
BSMB606	ADVANCES IN	VI
	MICROBIOLOGY	

Teaching Scheme (Hours)			Teaching Credit				
Lecture	Practical	Tutorial	Total Hours	Lecture	Practical	Tutorial	Total Credit
4	4	0	8	4	2	0	6

Course Pre-requisites	Students should have basic knowledge of advances in microbiology		
Course Category	Professional Elective/Core Professional/Generic Elective/Courses Offered by other departments/Open Elective		
Course focus	Employability		
Rationale	To have an overview of the basic knowledge of advances in microbiology		
Course Revision/Approval Date:	14/03/2020		
Course Objectives	Remember Understand the basic microbial structure and functions		
(As per Blooms' Taxonomy)	of various physiological groups of prokaryotes and eukaryotes and also learn the theory and practical skills in microscopy handling and staining techniques Apply Know various Culture media and their applications and understand various physical and chemical means of sterilization and also learn various techniques for isolation of pure cultures Analyses Comprehend the various methods for identification of unknown microorganisms and study microbial metabolism – Autotrophy and heterotrophy modes of nutrition.		
	eate Understand the microbial physiology and know the various Physical and Chemical growth requirements of bacteria and get equipped with various methods of bacterial growth measurement		
	derstand .		
	derstand the regulation of biochemical pathway and possible process modifications for improved control over microorganisms for microbial product synthesis.		

Course Content (Theory)		Contact
		hours
Unit 1: Evolution of Microbial Genomes. Salient features of sequenced	20%	12
microbial genomes, core genome pool, flexible genome pool and concept		
of pangenome, Horizontal gene transfer (HGT), Evolution of bacterial		
virulence - Genomic islands, Pathogenicity islands (PAI) and their		
characteristics.		



Unit 2:Metagenomics. Brief history and development of metagenomics, understanding bacterial diversity using metagenomics approach, Prospecting genes of biotechnological importance using metagenomics. Basic knowledge of viral metagenome, meta transcriptomics, metaproteomic and metabolomics.	20%	12
Unit 3: Metagenomics. Brief history and development of metagenomics, understanding bacterial diversity using metagenomics approach, Prospecting genes of biotechnological importance using metagenomics. Basic knowledge of viral metagenome, meta transcriptomics, metaproteomic and metabolomics.	20%	12
Unit 4: Molecular Basis of Host-Microbe Interactions. Biofilms: types of microorganisms, molecular aspects and significance in environment, health care, virulence and antimicrobial resistance.	20%	12
Unit 5: Systems and Synthetic Biology. Networking in biological systems, Quorum sensing in bacteria, Co-ordinated regulation of bacterial virulence factors, Basics of synthesis of poliovirus in laboratory, Future implications of synthetic biology with respect to bacteria and viruses. Gene editing concept with suitable example	20%	12

List Of Practical	Weightage	Contact hours
1) Extraction of metagenomic DNA from soil.	20%	12
Understand the impediments in extracting metagenomic DNA from soil.	20%	12
PCR amplification of metagenomic DNA using universal 16s ribosomal gene primers.	20%	12
4) Case study to understand how the poliovirus genome was synthesized in the laboratory.	20%	12
5) Case study to understand how networking of metabolic pathways in bacteria takes place	20%	12

	Course Objectives:	Blooms' Taxonomy Domain	Blooms' Taxonomy Sub Domain
CO1	Understand the basic microbial structure and functions of various physiological groups of prokaryotes and eukaryotes and also learn the theory and practical skills in microscopy		Explain, Describe, Discuss, Recall, Locate



	handling and staining techniques		
CO2	Know various Culture media and their applications and understand various physical and chemical means of sterilization and also learn various techniques for isolation of pure cultures	Apply	Apply, Practice, Interpret, Select, Correlate
CO3	Comprehend the various methods for identification of unknown microorganisms and study microbial metabolism – Autotrophy and heterotrophy modes of nutrition.	Analyses	Compare, Classify, Select, Investigate
CO4	Understand the microbial physiology and know the various Physical and Chemical growth requirements of bacteria and get equipped with various methods of bacterial growth measurement	Create	Construct, Develop, Produce
CO5	Understand the regulation of biochemical pathway and possible process modifications for improved control over microorganisms for microbial product synthesis.	Understand	Explain, Describe, outline, Predict, Summarize

Learning Res	sources			
1.	Reference books:			
	1. Fraser CM, Read TD and Nelson KE. Microbial Genomes, 2004, Humana			
	Press			
	2. Miller RV and Day MJ. Microbial Evolution- Gene establishment, survival and exchange, 2004, ASM Press.			
	3. Bull AT. Microbial Diversity and Bioprospecting, 2004, ASM Press			
	4. Sangdun C. Introduction to Systems Biology, 2007, Humana Press			
	5. Klipp E, Liebermeister W. Systems Biology – A Textbook, 2009, Wiley – VCH Verlag			
	6. Caetano-Anolles G. Evolutionary Genomics and Systems Biology, 2010, John Wiley and Sons			
	7. Madigan MT, Martink JM, Dunlap PV and Clark DP (2014) Brook's			
	Biology of Microorganisms, 14th edition, Pearson-Bejamin Cummings			
	8. Wilson BA, Salyers AA Whitt DD and Winkler ME (2011) Bacterial Pathogenesis- A molecular Approach, 3rd edition, ASM Press,			
	9. Bouarab K, Brisson and Daayf F (2009) Molecular Plant-Microbe interaction CAB International			
	10. Voit EO (2012) A First Course in Systems Biology, 1st edition, Garland Science			
2.	Journals & Periodicals			
	1) International Journal of Microbiology			
	2) Journal of Advances in Microbiology			
5	Other Electronic resources:			
	. NPTEL			

Evaluation Scheme	Total Marks



Theory: Mid semester	20 marks	
Marks		
Theory: End Semester	40 marks	
Marks		
Theory: Continuous		
Evaluation Component	Attendance	05 marks
Marks	MCQs	10 marks
	Open Book Assignment	15 marks
	Article Review	10 marks
	Total	40 Marks
Practical Marks		
	Attendance	05 marks
	Practical Exam	30 marks
	Viva	10 marks
	Journal	05 marks
	Total	50 Marks

	PSO1	PSO2	PSO3	PSO4	PSO5	PSO6
CO1	3	2	3	0	1	1
CO2	2	3	3	0	1	1
CO3	3	2	2	0	1	1
CO4	2	2	2	0	1	1
CO5	3	2	3	0	1	1

Mapping of POs & COs

	PO1	PO2	PO3	PO4	PO5	PO6				
CO1	2	0	2	1	0	0				
CO2	2	0	2	0	0	0				
CO3	2	0	2	0	0	0				
CO4	2	0	2	0	0	0				
CO5	2	0	2	1	0	0				